Pentopyranosyl Oligonucleotide Systems

Communication No. 13¹)

The α -L-Arabinopyranosyl-(4' \rightarrow 2')-oligonucleotide System: Synthesis and Pairing $\mathbf{Properties}^{1})$

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Dedicated to *Jack Dunitz*, teacher, colleague, and friend, on the occasion of his 80th birthday

Among the members of a family of diastereoisomeric pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide systems derived from D -ribose, D -xylose, L -lyxose, and L -arabinose, the α -arabinopyranosyl system shows by far the strongest Watson-Crick base pairing. The system is, in fact, one of the strongest oligonucleotide-type basepairing systems known. It undergoes efficient cross-pairing with all the other members of the pentopyranosyl family, but not with RNA and DNA. The paper describes the synthesis and pairing of the properties of α -Larabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides.

1. Introduction. - Among the six pentopyranosyl oligonucleotide systems studied in our laboratories [1] [3 - 6], the α -L-arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide system turned out to be the most surprising of all: here, we encountered the largest gap between expectation and observation with respect to a system's base-pairing properties. The qualitative conformational pairing criteria that had been conspicuously successful in predicting the base-pairing capabilities of homo-DNA [7] and p-RNA [4a] had led us to expect that an arabino analog of p-RNA would be a weaker pairing system than pyranosyl-RNA [4a]. Fact is that α -L-arabinopyranosyl-(4' \rightarrow 2')-oligonucleotides constitute the strongest base-pairing system among all potentially natural nucleic acid alternatives studied so far, surpassing the pairing strengths of both natural RNA and pyranosyl-RNA by a large margin [4]. Recently, Jaun and co-workers [8] determined the structure of an octamer duplex of the α -L-arabinopyranosyl-oligonucleotide series by NMR spectroscopy and provided therewith the basis for a rationalization of remarkable properties of the arabino system. Here, we report experimental aspects of

¹⁾ Communication No. 12: [1]. The paper is also communication No. 38 in the series Ω -Chemistry of α -Aminonitriles'. For communication No. 37, see [2].

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the synthesis and the pairing behavior of α -D- and α -L-arabinopyranosyl- $(4' \rightarrow 2')$ oligonucleotides, complementing the two preliminary publications on the topic [3a] [3b] and present a discussion of the system in comparison with the other members of the pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide family.

Scheme 1 illustrates the selection problem that arose when choosing the nucleoside configuration for each of the pentopyranosyl nucleoside members for a comparative study of four of the family's eight members. For the ribopyranosyl member [4], the first studied experimentally, the choice had been straightforward, since, of the two possible nucleoside D-diastereoisomers⁴), only the β -D-epimer could be expected to exist in a well-defined single-type of conformation. In this isomer, the nucleobase would unambiguously occupy an equatorial position at the pyranosyl chair. Furthermore, the β -D-nucleoside epimer promised to be the more accessible one, since, in this epimer, the nucleobase and the substituents at $C(2')$ are *trans* to each other. In choosing the other members of the family, the equatorial position of the nucleobase in the more-stable pyranose-chair conformation became the predominant selection criteria by reasons that were both chemical and etiological in nature. According to this criterion, the xylopyranosyl member had obviously to be the β -D-diastereoisomer [1]. Less certain was the choice in the cases of the remaining lyxo- and arabinopyranosyl members, since, for both, it could be expected that, in each of the two epimers, the nucleobase would occupy an equatorial position. In the case of lyxopyranose, the α -L-lyxopyranosyl series was chosen because its nucleosides were expected to be readily accessible (nucleobase *trans* to substituent at $C(2')$), and, furthermore, the sense of chirality at the nucleobase-bearing anomeric center is the one corresponding to that in the β -D-riboand β -D-xylopyranosyl series. In the arabinose case, we originally chose to synthesize β - $\texttt{D-arabinopyranosyl-nucleosides}$ to be able to study a quartet of pentopyranosyl-(4′ \rightarrow 2)-oligonucleotides in which the configurational differences between its members would follow a consistent and transparent pattern, allowing a meaningful comparison and interpretation of pairing properties. This pattern would have involved having the phosphodiester group 2',4'-diequatorial in the β -D-ribo and β -D-xylo systems, 2'equatorial/4'-axial in the α -L-lyxo system, 2'-axial/4'-equatorial in the β -D-*arabino* system, and the nucleobases equatorial in all four of them, with the anomeric center in all cases possessing the same sense of local chirality $(1/R)$.

Serious experimental difficulties in attempts to prepare the thymine nucleoside of the β -D-arabinopyranosyl series with its *cis*-arrangement of nucleobase and the substituent at $C(2')$ made us reconsider the choice of target. Since the epimeric α -Darabinopyranosyl nucleoside persistently dominated the product mixtures from such nucleosidation experiments, we eventually made this epimeric nucleoside (and its adenine analog) our new and much more conveniently accessible target. This switch in the configuration at the anomeric center demanded switching at the same time from the D- to the L-arabinopyranosyl series (Scheme 2). For practical reasons, we developed the methodology for the synthesis of α -arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides in the (chirally wrong) D-series and made the switch to the (chirally correct) α -Larabinopyranosyl series in a later phase of our work. After this change, the pattern

⁴⁾ α/β -Nomenclature: α -diastereoisomer in the pentopyranose series: nucleobase at C(1) is cis to the OH group at $C(4)$, whereas these two substituents are *trans* in the β -diastereoisomer [9].

of configurational differences within the family of the four pentopyranosyl-oligonucleotide systems becomes one in which both the nucleobase and the substituent at $C(2')$ are equatorial in all four systems and in which the configurational differences refer only to the substituents at $C(3')$ and $C(4')$. In all four systems, the local sense of chirality at the anomeric center is the same; this is essential for intersystem cross-pairing studies. The formulae of the four oligonucleotide systems studied experimentally are framed in Scheme 1.

2. Synthesis. $-$ 2.1. Preparation of α -D-Arabinopyranosyl Nucleosides Containing Adenine (A) and Thymine (T), and of α -L-Arabinopyranosyl Nucleoside Building Blocks Containing Adenine (A) , Thymine (T) , Cytosine (C) , and Guanine (G) . The Scheme 2. Constitution, Configuration, and Idealized Pairing Conformation of a -1-Arabinopyranosyl-(4' \rightarrow $2'$)-oligonucleotides. **B** = nucleobase.

synthesis of α -arabinopyranosyl-(4′ \rightarrow 2′)-oligonucleotides to be described in Sect. 2.2 was executed in two parts: the initial approach made use of a Fmoc-protecting-group strategy, resulting in the first preparations of adenine and thymine containing the α arabinopyranosyl oligonucleotides in the D-series. The approach was then extended to the preparation of adenine, thymine, cytosine, and guanine oligonucleotides in the series. Eventually, a follow-up synthesis of α -arabinopyranosyl oligonucleotides containing all four canonical nucleobases in the L-arabinose series by a DMTprotecting-group strategy paralleling the syntheses of oligonucleotides in the ribo-, xylo-, and lyxopyranosyl series [1] [4] [5] was developed. This latter development was induced by the demands of material for the recently published NMR-structure analysis of the pa $[{\rm (CGAATTCG)}]$, duplex [8].

Although a part of the work on α -arabinopyranosyl oligonucleotides described here had been initially carried out in the D-arabinose series (see above), all reactions reported in this paper in *Schemes* $2-5$ are uniformly formulated in terms of the L series. Special mention will be made whenever data are presented that were collected in the D-series only. There are reports in the literature where arabinopyranosyl nucleosides have been prepared in the α -L-, α -D-, and β -D-series, but in a different context [10].

 $2.1.1.$ Synthesis of α -D- and α -L-Arabinopyranosyl Nucleoside Building Blocks Containing the Fmoc Protecting Group. Nucleosidation of the readily accessible 1,2,3,4 tetra-O-benzoyl- β -L-arabinopyranose (1) [11] with the appropriately protected nucleobases N⁶-benzoyladenine (A^{Bz}), thymine (T), N⁴-benzoylcytosine (C^{Bz}), and N²isobutyryl guanine (G^{i-bu}) under the *Vorbrüggen-Hilbert-Johnson* conditions [12] furnished the nucleoside derivatives $2a - 2d$ in modest (44%) to high (84%) yields (Scheme 3). Reaction with N^2 -isobutyrylguanine proved to be low-yielding, with interference from the $N⁷$ -isomer 2e necessitating time-consuming purification by

column chromatography (CC)⁵). The constitutional assignment of N^9 - vs. N^7 -guanine regioisomers was accomplished by NMR-spectral-data comparisons between 2d and 2e as documented in Table 1, and UV-spectral data comparisons of the completely deprotected guanine derivatives (*Table 1* [13]), obtained by hydrazinolysis (NH₂NH₂· H₂O, EtOH, 12 h, r.t.) of 2d and 2e. Selective deprotection of the BzO groups without touching acyl protection of nucleobases was brought about by basic treatment (with either NaOH or Na/MeOH) to afford the triols $3a - 3d$ with yields in the 80% range⁶). At this stage, configurational assignments of compounds $3a - 3d$ were accomplished by ¹H-NMR analysis: typically large coupling constants $(J = 9 - 12 \text{ Hz})$ between the vicinal $H - C(1')$ and $H - C(2')$ as documented in *Table 2* confirmed that these nucleosides $3a - 3d$ have the α -configuration, and that the preferred conformation in solution $((D_6)$ DMSO) is the one depicted in their respective formulas of *Scheme 3*. For the thymine derivative $3b$ in the D-arabinose series, an X-ray analysis provided the final confirmation (*Fig. 1*)⁷).

Table 1. Constitutional Assignments of N^7 - and N^9 -(α -L-Arabinopyranosyl)guanine-Regioisomers Based on Comparison of Selected NMR Data of 2d and 2e, and UV Data of the Corresponding Free Triols

$Compounda$)	UV^b $\lambda_{\max}(\varepsilon)$ [nm]	¹ H-NMR ^c) $((D6)$ DMSO) δ [ppm]	13 C-NMR ^c) $((D_6)$ DMSO) δ [ppm]
N^7 -Isomer 2e		6.75 (br. s, $H - C(1')$) 8.60 (s, H – C(8))	84.63 $(C(1'))$
$7-(\alpha$ -L-Arabinopyranosyl) guanine N^9 -Isomer 2d	285 (5300)	6.02 (d, $J=9.5$, H-C(1')) 8.49 (s, H – C(8))	81.31 $(C(1'))$
$9-(\alpha$ -L-Arabinopyranosyl) guanine	252 (11400)		

^a) Free triols were obtained by hydrazinolysis ($NH_2NH_2 \cdot H_2O$, EtOH, 12 h, r.t.) of 2e and 2d, respectively. ^b) Measured in H₂O, pH 7, $c \approx 1$ mg/100 ml. Assignment based on the consistent behavior that N⁹-guanine nucleosides exhibit a maxima around 250 nm, while, in the corresponding N^7 -guanine isomers, it is shifted bathochromically to ca. 285 nm [13a]; also cf. Table 1 on page 1428 of [13c]. ^c) Assignment based on the fact that the $H-C(1')$, $H-C(8)$, and $C(1')$ chemical shifts are consistently downfield for the N^7 -guanine nucleosides as compared to the N^9 -derivatives [13].

Exploratory experiments in the D-series with free nucleosides (mostly 3b) and corresponding 2'-acylated derivatives of $3a - 3d$ with reagents such as BzCl and DMTchloride led to the observation that the relative reactivities of the three OH groups follows the order $HO-C(2') \approx HO-C(3') \geq HO-C(4')^8$). It was decided, therefore, that the least-reactive 4-OH function should carry the phosphoramidite group, while

⁵⁾ Further improvements using 2-amino-6-chloropurine as a surrogate for guanine in the nucleosidation reaction were promising, but difficulties with purification led to their abandonment.

In the case of cytosine nucleoside 2c, it was necessary to monitor the reaction carefully by TLC and stop the reaction before the loss of the $N⁴$ -benzoyl protecting group.

⁷⁾ Carried out by Raj K. Chadha, TSRI. Crystallographic data (excluding structural factors) for the structure reported in this paper has been deposited with the Cambridge Crystallographic Data Centre as deposition No. CCDC 201895. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB12 1EZ, UK (fax: 44 (1233) 336 0333; e-mail: deposit@ccdc.cam.ac.uk).

⁸⁾ For example, tritylation of the 2-acetate derivative of arabinopyranosyl thymine 3b led to the tritylation at $C(3')$ in 38% yield with no tritylation occurring at the 4'-OH.

Table 2. $^1H\text{-}NMR\ J(H-C(T),H-C(Z'))$ Values [Hz] for Selected Compounds (in $(D_6)DMSO$) Depicted in Scheme 3

Compound	a			
	(A)	'T'	(C)	(G)
2	9.1			9.5
3	9.2	12.4	9.0	9.3
4	9.6	9.7	9.3	9.5
5	9.7	9.8	8.5:8.8 (exolendo)	9.7
6	9.3	9.3	9.0	9.3
	9.3	8.0		9.3

Fig. 1. *X-Ray structures of* 2b, 6b, and 11a. Nucleosidic torsion angle $(\text{max} \pm \text{are as follows})$: $O-C(1')-N(1)-C(2)$: 2b: 86.2°; 6b: 125.4°. $O-C(1')-N(9)-C(4)$: 11a: -102.1° (see *Footnotes 7, 10*, and 14).

the more-reactive 2-OH would act as the nucleophile on the crucial chain elongation on the CPG-solid-support synthesis; such an approach is the opposite of that undertaken in the synthesis of oligonucleotides in the ribo- [4], xylo- [1], and the lyxopyranosyl series [5]. A regioselective introduction of a protecting group at the 2- OH position is to be achieved via a transient regioselective blockade of the cis-vicinal $3'$ -OH and $4'$ -OH groups by, e.g., cyclic isopropylidine acetal group, easily introduced or removed under mildly acidic conditions. The choice of an acid-labile isopropylidine acetal protecting group for the $3'$, $4'$ -positions precluded the use of DMT- as a $2'$ -Oprotecting group and thus dictated the choice of the Fmoc group as an acid resistant, base labile protecting group for the 2-OH position.

Reaction of triols 3a, 3b, and 3d with 2-methoxyprop-1-ene in DMF with HCl or TsOH as catalyst afforded the corresponding $3'$, $4'$ -isopropylidene derivatives $4a$, $4b$, and 4d, respectively, in over 80% yields, as expected with no signs of a competing formation of a *(trans!)* 2^{\prime} , 3^{\prime} -isopropylidene derivative. For triol **3c**, the use of 4anisaldehyde (4-methoxybenzaldehyde) dimethyl acetal in place of 2-methoxyprop-1 ene under similar reaction conditions (DMF, TsOH) afforded a mixture of diastereoisomeric $3'$,4'-benzylidene acetals **4c**, in 89% yield, as the sole product⁹). Introduction of Fmoc group at $C(2')$ was achieved by treatment of $4a - 4d$ with Fmoc-Cl in pyridine at 4° to give $5a - 5d$ in good yields $(73 - 88\%)$. Subsequent acid-catalyzed cleavage of the $3'$, 4'-acetal protecting groups by treatment with 80% aq. AcOH for 5a and $5d$, with $4M$ HCl in dioxane for $5b$, and with 0.7 M TFA in MeOH/THF 2:1 for $5d$, generated the 2'-Fmoc derivatives $6a - 6d$ in yields ranging from $70 - 99\%$ ¹⁰). As pointed to before, the lack of reactivity of the 4-OH group towards electrophiles came in handy for the selective introduction of the Bz group at 3-O-position by simply treating $6a - 6c$ with ca. 1 equiv. of BzCl at $-20-4^{\circ}$ to afford the 3'-O-Bz derivatives **7a** $-7d$ in 70–83% yields¹¹). No additional benzovlation at the nucleobases under these conditions were observed. The appropriately protected 3-O-Bz-2-Fmoc derivatives $7a - 7d$ were the starting materials for the preparation of building blocks suitable for automated solid-support synthesis.

The phosphitylations of $7a - 7d$ were conducted under standard conditions [14] with $(2$ -cyanoethoxy)bis(diisopropylamino)phosphine and with $1H$ -tetrazole as a catalyst (Scheme 4). A full equivalent of $1H$ -tetrazole was used to prevent the loss of the Fmoc

²⁻Methoxyprop-1-ene, in fact, could be used in the case of $3c$ with no difficulty for the formation of $3'$, $4'$ isopropylidene acetal derivative; but, after formation of 2-Fmoc derivative, problems (side-product formation) were encountered in the subsequent deprotection of 3,4-isopropylidene acetal group to generate 6c.

¹⁰) In the case of the thymine derivative **6b** in the D-arabinose series, X-ray analysis (carried out by *Raj K*. Chadha, TSRI) confirmed the constitution and configuration (Fig. 1). Crystallographic data (excluding structural factors) for the structure reported in this paper has been deposited with the Cambridge Crystallographic Data Centre as deposition No. CCDC 201894. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB12 1EZ, UK (fax: $+44$ (1233) 336 0333; e-mail: deposit@ccdc.cam.ac.uk).

¹¹) Among the benzoylating reagents tried BzCl, BzCN, and Bz₂O, BzCl was found to be the reagent of choice. While Bz_2O (in the presence of DMAP or 1-methyl-1H-imidazole) was unreactive and resulted in the cleavage of the base-sensitive Fmoc group, use of BzCN in the case of 6b led to mixtures of 3-,4-, and 3,4-benzoylated products under the reaction conditions employed.

group by the $(i-Pr)_{2}NH$ generated during the reaction¹²). The purification and isolation of the phosphoroamidites were carried out by very quick $(10-15 \text{ min.})$ CC over silica gel at room temperature (silica gel pretreated with toluene containing 1% Hünig's base, followed by careful washing with toluene to remove all of the base) to afford 8a -8d in yields of up to 95%. Such precautions were necessitated by the observations that these phosphoramidites readily decompose under normal chromatographic isolation procedures.

The sensitivity of the Fmoc group to base interfered, once again, with the loading of the solid support CPG with nucleobases $7a - 7d$; compromises were nevertheless found, which led to satisfactory production of suitably loaded solid supports $10a - 10d$. As the first step, the 4'-OH group was derivatized as the $4 - 0$ -succinate by reacting **7a** – **7d** with 2.0 equiv. of succinic anhydride in the presence of $0.1 - 0.5$ equiv. of DMAP in CH₂Cl₂ as solvent over a $5 - 7$ day period at room temperature [15]. Minimizing the amount of DMAP (and monitoring the reaction by TLC) was found to eliminate the unwanted loss of the Fmoc group. The crude $4'-O$ -succinates $9a-9b$ (confirmed by ¹H-NMR spectroscopy and mass spectrometry) were directly used in the next step of attachment to the specially derivatized LCAMA-CPG-solid support, which has been shown to be stable to the conditions of base induced eliminative deprotection of the Fmoc group [16]. Commercially available glyceryl-modified CPG was reacted with N,N'-dimethylhexane-1,6-diamine in the presence of $1,1'-$ carbonyldi $[1H$ -imidazole] to generate the LCAMA-CPG (which has the advantages of a long spacer arm and the large pore size of 500 ä, characteristics considered to be advantageous for larger loading values and synthesizing long oligonucleotides). Reaction of the LCAMA-CPG with 4'-Osuccinylated nucleosides $9a - 9d$ in the presence of TOTU [17] as a condensing agent, followed by capping with excess Ac_2O in the presence of 1-methyl-1H-imidazole, led to suitably derivatized LCAMA-CPGs $10a - 10d$. The loading capacities were determined to range from $12 - 19 \mu \text{mol/g}$ by treatment of $5 - 10 \text{ mg}$ of the CPG with 5 ml of 0.1m soln. of DBU in MeCN, and by measuring the absorbance of the solution at 305 nm with a molar absorptivity value (ε) for methylidene-fluorene of 9100 (see *Exper. Part, Sect. 2* for details).

With derivatives $8a - 8d$ and $10a - 10d$, the requisite building blocks, suitable for proceeding to the synthesis of α -D- and α -L-arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides by automated solid-phase synthesis and the Fmoc strategy, were in hand.

2.1.2. Synthesis of α -L-Arabinopyranosyl Nucleoside Building Blocks Containing the DMT Protecting Group. While the above Fmoc-protecting-group approach was useful for furnishing enough phosphoramidites for the synthesis of α -D-arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides in a small-scale $(0.5 - 1 \text{ \mu mol})$ range (see Sect. 2.2), the low loading capacity of the CPG solid support and the problems presented by the Fmoc group due to its inherent instability under basic conditions made this approach

¹²) These conditions were chosen over the usual Δ +basic^{*} reaction conditions, *i.e.*, phosphitylation with (chloro) (diisopropylamino)phosphine with either Hünig's base or with collidine and N-methyl-1Himidazole, because the sensitivity of the Fmoc group to the base used or produced in the phosphitylation reaction. These 'basic' reaction conditions were found to yield only ca. 60% of the phosphitylated products, compounded by complications due to loss of Fmoc group and the incomplete removal of the unreacted phosphitylating reagent by CC.

unsuitable for large-scale synthesis. Yet, the need for such a 'large-scale' production $(ca. 10-15 mg)$ of pa $(CGAATTCG)$ came up in the project to determine the structure of α -arabinopyranosyl-(4′ \rightarrow 2′)-oligonucleotide duplex by NMR spectroscopy [8]. Therefore, we resumed our efforts to develop an approach based on the use of DMT as a protecting group with the sole purpose of having access to large quantities of α -Larabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides.

The first requirement of this approach was to find a way to produce the 2-O-trityl derivatives with high selectivity and in high yields. Toward this goal, exploratory studies on the tritylation of triols 3a and 3b were conducted. Treatment of triol 3a with $Me₃SiNEt₂$ (DMF/ – 40 to –55°) followed by tritylation (DMTCl/AgOTf/collidine/ MeCN/room temperature/3 d) furnished the 4-O-DMT derivative, in low yields (31%) . Attempts to achieve a regioselective benzovlation with BzCN *via* a cyclic tributyl stannane *cis*-diol derivative [18], furnished the 3'-O-benzoylated derivatives of 3a and 3b, subsequent tritylation (DMTCl/AgOTf/collidine) provided the 3'-Obenzoyl-4-O-trityl derivative in 53% yield in the case of 3a, while a mixture of the 3- O-benzoyl-4-O-trityl and the 3-O-benzoyl-2-O-trityl derivatives was obtained starting from 3b. Once again, the formation of product mixtures and low yields in the tritylation step prompted us to abandon these approaches.

While searching for a selective protection group for the *cis*-vicinal $3'$, $4'$ -diol function, which could be removable again in the presence of a O-DMT group, we came across a recent paper by $R.$ A. Jones et al. [19] describing the use of the reagent N , N dimethylformamide dimethyl acetal for the reversible protection of vicinal diols. Cyclic (dimethylamino)methyleneacetals of cis-diols were found to be cleaved with simple protic solvents such as MeOH. The use of this reagent¹³) solved our problem for a selective and efficient tritylation of the 2'-OH group.

Another improvement we wanted to realize while shifting from the Fmoc- to the DMT-based strategy concerned the problems encountered in the nucleobase-protection step of the oligonucleotide synthesis (see Sect. 2.2.1). Those problems were to be overcome by replacing the $NH₂$ protecting groups for adenine (benzoyl), guanine (isobutyryl), and cytosine (benzoyl) by phenoxyacetyl (−PAC×, for adenine as well as guanine [20]) and isobutyryl (for cytosine [20]), the former for the sake of being more easily removable, the latter for being more stable (see *Footnote 6*).

Scheme 5 summarizes the preparation of the four nucleoside building blocks $15a -$ **15d** that were required for the DMT-based approach to α -arabinopyranosyl- $(4' \rightarrow 2')$ oligonucleotides, work that was done now exclusively in the L -arabinose series. Central intermediates were the corresponding trihydroxy nucleosides 12a, 3b, 12c, and 12d, of which the thymine member was available from the earlier work (see Scheme 3), whereas the cytosine member was made from scratch by nucleosidation of Larabinopyranosyl-tetrabenzoate 1 with N-isobutyrylcytosine to 11c, followed by hydrolytic cleavage of all protecting groups and selective reprotection of the cytosine nucleus's primary amino group with isobutyric anhydride in DMF. The trihydroxypurine nucleosides $12a$ and $12d$, on the other hand, were made by exchange of the Nprotecting groups in the nucleosides $2a$ and $2d$ used earlier. This was accomplished by

¹³⁾ Initiated by A. L. The publication [19] appeared while our work was in progress.

complete deprotection by treatment with $2M NH₃/MeOH$ to give $11a¹⁴$ and $11b$, followed by introduction of the new PAC protection group into the purine nuclei by a three-operation procedure consisting of O-silylation [21], N-acylation [20], and Odesilylation, producing the nucleosides 12a and 12d in high yields.

With the four trihydroxy nucleosides 12a, 3b, 12c, and 12d in hand, we were ready for the critical introduction of the trityl group the $2'-O$ -position *via* transient protection of the vicinal 3',4'-diol groups utilizing the chemistry described by R . A. Jones et al. [19]. We were relieved to find that a one-pot procedure of 3',4'-amide-acetal formation with nucleosides 12a, 3b, 12c, and 12d (1.5 equiv. N,N-dimethylformamide dimethyl acetal, pyridine, collidine, DMF, room temperature, $1-3$ h) followed by tritylation $(1.2-1.5)$ equiv. DMT-Cl, collidine, CH_2Cl_2 , room temperature, $6-12$ h) and subsequent careful methanolysis¹⁵) of the $3'$,4'-amide-acetal group worked reliably and effectively furnished the $3'$.4'-dihydroxy-2'-O-tritylated nucleosides $13a-13d$ in yields of $71-$ 61% after chromatographic purification¹⁶).

In our search for a solution to the last protection problem, that involving the 3-OH function, we eventually had to accept a compromise. In deciding to stick to benzoate as the protection group, the strategy we had successfully used in the ribopyranosyl series [4], we first had to learn that, in this DMT-protected arabino series, it is not the equatorial 3-OH group that can regioselectively be benzoylated but rather the axial 4- OH group. The behavior is the opposite of what we had observed previously while exploring the 2'-Fmoc-protected 3',4'-dihydroxy nucleosides $6a - 6d$ (see Scheme 3); it is clearly due to steric hindrance to 3-benzoylation exerted by the trityl group. In welcome analogy to our experience in the synthesis of ribopyranosyl- $(4' \rightarrow 2')$ oligonucleotide [4], where the observation of a $(2' \rightarrow 3')$ -migration of the O-Bz group had been crucial, we showed that the (under mild conditions¹⁷)) regioselectively accessible 4'-monobenzoates $14a - 14d$ undergo a clean equilibration in CH₂Cl₂ solution and the presence of Et_3N at room temperature producing a ca. 1:1 mixture of 4'- and 3'monobenzoates from which the pure $3'$ -monobenzoates $15a - 15d$ could be isolated by chromatography in yields of $35-38%$ (with respect to $14a-14d$). This we accepted as a compromise for the preparation of the desired nucleosides $15a - 15d$, taking into account that the recovered 4-benzoates could be (and were) recycled.

Phosphitylation of $15a - 15d$ with the commercially available chloro(2-cyanoethoxy)-(diisopropylamino)phosphane in the presence of $(i-Pr)$ ₂NH gave phosphoroamidites **16a** $-$ **16d** in 79 $-$ 85% yields (*Scheme 6*). For the preparation of the CPG solid support,

¹⁴) The structure was also confirmed by X-ray analysis (*Fig. 1*; carried out by *Raj K. Chadha*, TSRI). Crystallographic data (excluding structural factors) for the structure reported in this paper has been deposited with the Cambridge Crystallographic Data Centre as deposition No. CCDC 201893. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB12 1EZ, UK (fax: 44 (1233) 336 0333; e-mail: deposit@ccdc.cam.ac.uk).

¹⁵⁾ Continuous TLC monitoring of the hydrolytic workup is important due to the high lability of PAC protecting group. After addition of MeOH, a spot with lower R_f (corresponding to presumed hemi-acetal) appears within minutes, which, then, over a period of 30 min converts to a more-polar spot corresponding to the desired triols.

¹⁶⁾ Chromatographic isolation in the case of 13a and 13d was performed with i-PrOH instead of MeOH to minimize the loss of PAC protecting groups.

¹⁷) Benzoylation with BzCl/pyridine in CH₂Cl₂ at -30° to -40° ; the guanine derivative **13d** required special conditions $(Bz_2O/pyridine, DMPA, r.t.).$

Scheme 5. Preparation of the A, T, C, and G Nucleoside Building Blocks for the Synthesis of α -L-Arabinopyranosyl-(4' \rightarrow 2')-oligonucleotides via the DMT-Protecting-Group Strategy

the standard one-pot protocol used previously for the pyranosyl *ribo*, $xylo$, and $lyxo$ series $[1][4][5]$, (formation of the 4'-O-succinic acid derivative by treatment with succinic anhydride in the presence of DMAP, followed by CPG derivatization with TOTU [17] as an activator and subsequent capping with Ac_2O), was found to work with 15c to give 17c without any problem. The loading capacity [13c] was determined to be 53 mol/g, what we considered to be high. The cytosine-containing CPG solid support was the only one needed, since $pa(4'-CGAATTCG-2')$ was the only arabinopyranosyloligonucleotide strand we synthesized with the DMT-based strategy.

2.2. Synthesis of α -Arabinopyranosyl-(4' \rightarrow 2')-oligonucleotides. Table 3 lists all α arabinopyranosyl sequences synthesized in the course of our study. By reasons mentioned in the Introduction, sequences made in the first phase of the project contained the D-arabinopyranose unit, while those prepared at a later stage belong to the L-arabinopyranosyl series. With the exception of the self-complementary sequence $pa(CGAATTCG)$, which was prepared in 'large' amounts (ca. 10 mg) for the NMR structure analysis [8], the amount made in the case of all other sequences was in the range of 1 µmol. The methodology followed was in essence the one previously used in the ribo- [4], xylo- [1], and lyxopyranosyl [5] series on an automated DNA synthesizer (Perseptive's Expedite), taking into account, as detailed in Sect. 2.2.1 and 2.2.2 below, the specific aspects and demands of the two different protection strategies. In these two

strategies, the solid-phase synthesis of α -arabinopyranosyl oligonucleotide strands proceeds in opposite directions, namely, the 3'(phosphityl) \rightarrow 2'(hydroxy) direction in the Fmoc-protection strategy and the $2'(p \text{hosphityl}) \rightarrow 3'(hydroxy)$ direction in the DMT-protection strategy. If, in a future preparation of α -arabinopyranosyl- $(4' \rightarrow 2')$ oligonucleotides, one had to choose between the two strategies with respect to the criterion of expediency and yield, the DMT-protection strategy would clearly be preferred.

	<i>Entry</i> Base sequences ^a) (all $4' \rightarrow 2'$)	Deprotection ^b) O.D. 260 nm Anal. HPLC Method	(yield)	ion exchange ^c)	$MALDI-TOF-MSd)$ $[M+H]^+$	
				gradient, t_{R} [min]	obs.	calc.
1	p -pa (A_8)	A	4.4 (5%)	10-50% in 30 min, t_R 14.8	2573	2572
2	$L-pa(A_8)$	А	$4.8(9\%)$	10-50% in 30 min, t_{R} 14.8	2571	2572
3	$L-pa(A_{12})$	A		4.9 (5%) 10-50% in 30 min, t_R 20.2	3890	3889
$\overline{4}$	p -pa (T_4)	B		20 – 80% in 30 min, t_{R} 14.1		1223 (Na ⁺) 1224 (Na ⁺)
5	p -pa (T_8)	А		4.2 (16%) 0-100% in 30 min, $t_{\rm p}$ 21.0	2501	2500
6	$L-pa(t_8)$	А		8.8 (18%) 0-100% in 30 min, t_{R} 21.0	2501	2500
7	$L-pa(T_{12})$	А		4.8 (7%) $0-100\%$ in 30 min, t_p 22.9	3780	3780
8	$D-pa(T_4A)$	B		10-50% in 30 min, t_{R} 11.5** 1547		1546
9	$D-pa(T_4A_2)$	B		10-50% in 30 min, t_{R} 15.2** 1876		1876
10	$D-pa(T_4A_3)$	B		10-50% in 30 min, t_{R} 25.9** 2206		2205
11	p -pa (TA_8)	B			2891	2892
12	p -pa $((ATATA)$	B		8.5 (31%) 10-50% in 30 min, t_R 17.5	1564	1566
13	D-pa(TATAT)	\boldsymbol{A}		20 – 80% in 30 min, $t_{\rm p}$ 10.3	1557	1557
14	D-pa(ATATATA)	А		20 – 80% in 30 min, $t_{\rm p}$ 12.7	2214	2214
15	$D-pa(A_4T_4)$	C		7.6 (13%) 10-50% in 30 min, t_p 28.2	2534	2534
16	$D-pa(T_4A_4)$	B		7.6 (13%) 10-50% in 30 min, t_R 28.4** 2535		2534
17	$D-pa((AT)4)$	A		$20-80\%$ in 30 min, $t_{\rm p}$ 14.4	2536	2536
18	p -pa $(TA)_4$	\boldsymbol{A}		20 – 80% in 30 min, t_{R} 15.1	2536	2536
19	L-pa(TATTTTAA)	A		7.4 (16%) 0-100% in 30 min, t_R 19.0** 2527		2527
20	L-pa(TTAAAATA)	\boldsymbol{A}		13.1 (25%) 0-100% in 30 min, t_p 16.2	2544	2545
21	$L-pa(G_8)$	A		2.7 (9%) 0-100% in 30 min, t_{R} 26.6	2010	2009
22	$L-pa(C_8)$	\overline{A}		4.3 (16%) 10 – 50% in 30 min, t_p 9.9	1771	1770
23	L-pa(ATTCAGCG)	D		4.3 (7%) $0-100\%$ in 30 min, t_R 19.1	2538	2538
24	L-pa(CGCTGAAT)	D		2.2 (6%) 0-100% in 30 min, t_R 18.9	2536	2538
25	L -pa(CGAATTCG)* E			446 (11%) 0-100% in 30 min, t_R 21.5	2537	2538

Table 3. HPLC and MS Data of α -Arabinopyranosyl-(4' \rightarrow 2')-oligonucleotides

^a) All sequences were synthesized by the Fmoc-protection strategy with the exception of the sequence 25, labeled *, which was synthesized by the DMT-protection strategy. \overline{b}) Method A: 0.15M MeONH₂ in 25% aq. NH₃/EtOH 3 : 1 at r.t. for 10 h (24 h for mixed AT and homo-G sequences); Method B: 25% aq. NH₃ at 4° , 9 – 24 h; Method C: 25% aq. NH₃ at 4° , 21 h followed by 25% aq. NH₂NH₂, r.t., 10 min; Method D: ethylenediamine/EtOH 1:1, r.t., 2 h [22]; Method E: pyridine/Et₃N 4:1, r.t., 5 h; 40% aq. MeNH₂, r.t., 1 h. ^c) All oligonucleotides were purified by ion-exchange chromatography on a Mono Q HR 5/5 column (58 \times 6.0 mm, Pharmacia); elution with 10 mm Na₂HPO₄ in H₂O and a linear gradient of 1m NaCl with a flow of 1 ml/ min. pH $10.5 - 11.5$; followed by desalting on *Sep-Pak* cartridges, peak purity (260 nm) $95 - 99\%$. ** Elution with 25 mm Tris and 1.5m urea in H₂O, and a linear gradient of 1m NH₄Cl, pH 8.0. ^d) Matrix-assisted laser-desorption ionization time-of-flight mass spectroscopy; matrix: 3-hydropicolinic acid or a-cyanohydroxycinnamic acid or 2,4,6-trihydroxyacetophenone and ammonium citrate buffer.

2.2.1. Oligonucleotide Synthesis with Fmoc-Protected Building Blocks. Since MeCN solutions of Fmoc-phosphoramidites $8a - 8d$ (stored with 4- \AA molecular sieves) were found to decompose slowly (over periods of days), for their use in the solid-support synthesis, all solutions were freshly prepared and used within $24 h^{18}$). The syntheses were performed with ca. 30 mg of solid-support derivatives $10a - 10d$ with the following modifications to the protocols used earlier: a) coupling of phosphoroamidites over a period of 15 min with either 0.5 M pyridine hydrochloride or 0.5 M 1H-tetrazole in MeCN and b) cleavage of Fmoc with 0.1 DBU in MeCN over a 1.5-min period. Coupling efficiencies were monitored by measurement of UV absorbance of fulvene resulting from the cleavage of Fmoc group after each coupling (in analogy to Trityl assay). They ranged from 95 to 98% for T-, C-, and G-containing residues, while around 90% for A-containing residues¹⁹). Coupling yields were found, sporadically, to drop to ca. 80% in the synthesis of longer oligonucleotide strands.

The Fmoc groups of the last residue were removed before detaching the oligonucleotide strand from the solid support. Detachment and concomitant removal of all remaining groups posed a major problem. Under the base treatments normally used (such as aq. NH₃), some of the α -arabinopyranosyl-(4' \rightarrow 2')-oligonucleotides were found to be extremely sensitive towards strand scission. Therefore, various conditions had to be explored and fine-tuned for the specific sequence under consideration, these conditions were $-$ depending on the specific oligonucleotide sequence $-$ either treatment of the dried CPG material with $0.15M$ MeONH₂ \cdot HCl in 25% aq. NH₃/EtOH 3:1 at room temperature for $10-24$ h (*Method A, Table 3*), or with 25% aq. NH₃ at 4° for 9–24 h (*Method B, Table 3*), or with 25% aq. NH₃ at 4° for 21 h, followed by 25% aq. $NH_2NH_2 \cdot H_2O$ at room temperature for 10 min (*Method C*, see Table 3), or with ethylenediamine/EtOH 1:1 at room temperature for 2 h (Method D, Table 3). All these treatments led to the deprotection of the Bz groups on the sugar, the 2-cyanoethyl group on the phosphate, as well as the acylamino groups of adenine, cytosine, and guanine with the concomitant detachment of the oligonucleotides from the CPG solid support. The homo-adenine sequences (*Entries* $1-3$ *, Table 3*) did not fare well under the standard aqueous NH₃ deprotection conditions, yielding lots of strand-scission products; milder treatment with MeONH₂ \cdot HCl in combination with aq. $NH₃$ (*Method A*) resulted in the least strand scission. For the homo-guanine-containing sequence L-pa(G₈) (*Entry 21*), it was observed that the N^2 -isobutyryl groups were not fully deprotected within 10 h; therefore, the deprotection time was prolonged, and a balance had to found between complete deprotection and the amount of strand-scission products. Thus, the deprotection was stopped after 24 h, and the peak corresponding to L -pa(G_8) was isolated²⁰). In the case of L -pa(ATTCAGCG) and its complementary sequence where decomposition was most serious, treatment with ethylenediamine/ EtOH 1:1 at room temperature for $2 h$ [22] led to a marked improvement in minimizing strand scission. In all cases, deprotections were monitored by ion-exchange HPLC for optimal deprotection time. While almost all of the oligonucleotide sequences

¹⁸⁾ Decomposition of Fmoc-phosphoramidites as seen by TLC. The problem was most severe in the case of guanine derivative 7d.

¹⁹) In a typical 0.5-µmol-scale synthesis, each Fmoc-derived fulvene eluate was diluted to 5 ml with MeCN, and the absorbance at 305 nm was measured against a 0.1M soln. of DBU in MeCN as a reference. The value obtained for each fulvene eluate was compared as a percentage to that of the eluate of the previous step.

²⁰) Also isolated were L-pa(G_6) and L-pa(G_7) from this deprotection procedure as strand-scission products.

could be purified by ion-exchange HPLC (target purity 95%), on Mono Q HR $5/5$ column (*Pharmacia*) with 10 mm NaH₂PO₄ buffer at pH 10.5 – 11.5 with a linear gradient of 1_M NaCl (see *Exper. Part, Sect. 6*), some sequences containing thymine and adenine (*Entries 8-10, 16*, and 19) remained still unresolved; these sequences were purified with 25 mm Tris HCl buffer at pH 8.0 containing 1.5M urea with a linear gradient of $1M$ NH₄Cl. Product peaks (monitored at 260 nm) were collected in vials containing 20 μ of a 1*M* aq. AcOH (to prevent further strand scission) and desalted (see *Exper. Part, Sect.* 7), and the desalted oligonucleotides dissolved in 1 ml of H_2O . The concentrations of such purified oligonucleotide stock solutions were determined by UV (at ca. 80 $^{\circ}$) with the following molar extinction coefficients at 260 nm at pH 7: ε (pa(A)) = 15000, ε (pa(T)) = 10000, ε (pa(C)) = 8400, ε (pa(G)) = 11900. The purified oligonucleotides were checked for the correct mass by matrix-assisted laser-desorption ionization time-of-flight (MALDI-TOF) mass spectrometry [23]. For the oligonucleotides synthesized by this procedure and for their HPLC and MS data, see Table 3.

2.2.2. Synthesis of α -L-Arabinopyranosyl-(4'-CGAATTCG-2') with DMT-Protected Building Blocks. This synthesis followed the previously established protocols [1] [4] [5] on the automated DNA synthesizers (*Perseptive*'s *Expedite*) on a 5-umol scale, with the following modifications to the standard program: a) increase of the coupling time to 27 min with a 0.07 solution of phosphoramidite in MeCN (this modification allowed the parsimonious use of only ca. 5-fold excess of phosphoramidite), and b) use of $0.35M$ solution of 5-(ethylthio)-1H-tetrazole in MeCN as activator. Average coupling efficiencies were ca. 93%, as judged by trityl assay (ε (500 nm) = 66000). A total of nine such synthesis operations, carried out in the 'Trityl-Off² mode, were performed to reach an overall 45-umol scale.

The detachment from the solid support and the deprotection of the oligonucleotide was effected in two stages: first, treatment of the dried CPG solid support with pyridine/Et₃N 4:1 for 5 h at room temperature [24] (removal of the 2-cyanoethyl protecting group converting the phosphotriester group into the more-stable phosphodiester group) and, second, exposure to 40% aqueous MeNH₂, effecting the simultaneous deprotection of the sugar and nucleobase protecting groups along with detachment of the oligonucleotide from the solid support. The crude oligonucleotide was purified (target purity $> 95\%$) by ion-exchange HPLC on Mono Q HR 5/5 column (*Pharmacia*) with 10 mm NaH_2PO_4 buffer at pH 11.3 with a linear gradient of 1M NaCl (see *Exper. Part, Sect.* 6)²¹). The product peak (monitored at 260 nm) was collected in a vial containing 0.1 phosphate buffer at pH 7.0 to prevent any strand scission. The oligonucleotide was desalted (see *Exper. Part, Sect.* 7), and its integrity was checked by MALDI-TOF mass spectroscopy (*Entry 25, Table 3*). The counter ion was exchanged to Na⁺ by treatment with Na⁺ form of *DOWEX* (50 W \times 8) to make it suitable for NMR measurements. The concentration of the oligonucleotide measured with ε (260) nm) = 90600, gave a total of 4.92 µmol of the oligomer (15.9 mg isolated; 11% overall yield).

²¹⁾ This high pH was found to be more satisfactory in preventing the self-association of this selfcomplementary oligonucleotide than the usual pH of 10.5.

3. Base-Pairing Properties. – The pairing properties of α -D- and α -L-arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides were characterized by methods used in earlier studies in the ribo- [4], xylo- [1], and lyxopyranosyl oligonucleotide series [5]. These methods are: temperature-dependent UV spectroscopy (for T_m measurements [25]), concentration-dependent T_m measurement by UV spectroscopy (for determination of thermodynamic data [26]), molar-ratio-dependent UV spectroscopy (for checking stoichiometry of complex formation [27]), and temperature-dependent CD spectroscopy. All measurements were made in ϵ +phosphate buffer (10 mm aq. NaH₂PO₄ buffer containing 0.1 mm Na₂(EDTA), 150 mm (or 1m) NaCl at pH 7.0) with a total oligonucleotide concentration of ca . 10 μ m, unless otherwise stated. Tables 4 and 5 list the T_m and thermodynamic data collected for the intra- and intersystem pairing duplexes formed by the α -L-arabinopyranosyl- $(4' \rightarrow 2')$ -oligon ucleotides, while Figs. 3–6 document pairing behavior, complementing information published in preliminary communications [3]. The data of *Table 4* indicate whether a given observation refers to the D- or L -series²²); all data in *Table 5* refer to the L -arabinopyranosyl series.

The T_m values listed in *Tables 4* and 5 document the dominant characteristic of α arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides, namely, that they form more-stable duplexes among all members of the pentopyranosyl-oligonucleotide family studied thus far. Duplexes of pa-octamers have T_m values on the average of *ca*. 20[°] higher than corresponding pl-octamers, their nearest corresponding competitors (Table 5). This property tends to extend to the phenomenon of self-association of non-self-complementary strands, both in homo-oligomer (*Table 4, Entries 2-4*) and hetero-oligomeric strands (*Entries 8, 11*, and 12). While self-association is weak for (A_8) and (A_{12}) oligomers $(T_m < 5^\circ$ and 7.8°, resp.), it is remarkably pronounced in (T_8) and (T_{12}) series $(T_m = 26.9^\circ$ and 52.2 \degree , resp., see Fig. 2). This behavior is also reflected in the behavior of the antiparallel-complementary, but not self-complementary, sequences (4-TATTTTAA-2) and (4-TTAAAATA-2); mixed together, they form a duplex that is stronger than the self-association of the individual strands (Fig. 3, d, and Table 4, Entries $8-10$). The sequences (G_6) and (C_6) , remarkably, do not exhibit discernable self-pairing; yet, mixing of these two sequences produces a very strong duplex $(T_m = 72.3^\circ$ in 0.15m NaCl, *Entry 7*; Fig. 3). The tendency for self-pairing in a sequence with mixed A and T nucleobases is diminished by the introduction of G and C nucleobases. This is demonstrated by the low T_m values of self-association exhibited by the non-self-complementary sequence (4'-ATTCAGCG-2) and its complementary sequence (4-CGCTGAAT-2) (Entries 11 and 12). Once again, the duplex formed after mixing these two sequences has high stability $(T_m = 80.5^\circ$ in 0.15m NaCl, *Entry 13*; Fig. 4). Testifying to the superiority of the α -arabinooligonucleotide sequences in terms of the strength of base-pairing, the self-complementary sequence (4'-CGAATTCG-2') melts at a $T_m = 78.1$ (1 µm, 0.15m NaCl, *Entry 13*; Fig. 4, c), while the T_m value of the corresponding p-RNA sequence is 51°23)!

The CD spectra of the α -L-arabino-oligonucleotide sequences are strikingly similar to those observed in the a -L-lyxo series, but markedly different from the β -D-ribo and β -

²²) In one case, $pa(A_8) + pa(T_8)$, UV- T_m measurements in both the $p-a$ - and $L-a$ -series were made, observed $T_{\rm m}$ values were: 71.1° in the D- α -case (0.15M NaCl, 0.01M Tris buffer) and 68.3° in the L- α -case (0.15M NaCl, 0.01_M phosphate buffer).

²³⁾ Calculated from the thermodynamic data reported in [4b].

	Entry Oligonucleotide System (L-series)	$T_{\rm m}$ $\lceil \degree \rceil^{\rm a}$	Duplexes, $c \approx 10 \mu$ M	ΔG (25°)	ΔH [kcal/mol] (25°)	$T\Delta S$	Ref.
			1M NaCl $0.15M$ NaCl $0.15M$ NaCl ^a) ^b)				
\mathcal{I}	$pa(A_8)$	\leq 5					
$\overline{2}$	$pa(A_{12})$	7.8°)					
3	$pa(T_8)$	26.9°)					
$\overline{4}$	$pa(T_{12})$	52.2°)					
5	$p - pa(A_8) + p - pa(T_8)$		78.7 ^d) 71.1 ^d)		-15.7^{d} -60.6^{d}	$44.9d$)	$\lceil 3a \rceil$
6	$pa(A_{12}) + pa(T_{12})$	> 95	74.3°)				[3b]
7	$pa(G_6) + pa(C_6)$	$-$	72.3	-12.1	-30.7	-18.6	
8	pa(TTAAAATA)	$48.1d$)		-8.2	-31.1	-22.9	
9	pa(TATTTTAA)	(48.9^{d})		-9.4	-53.2	-43.8	
10	$pa(TATTTTAA) + pa(TTAAAATA)$	74.6	65.6	-15.3	-66.1	-50.8	[3b]
11	pa(ATTCAGCG)	\lt 5					
12	pa(CGCTGAAT)	$18.6^{\rm d}$)					
13	$pa(ATTCAGCG) + pa(CGCTGAAT) > 88$		80.5	-17.6	-63.0	-45.4	[3b]
	Self-complementary sequences						
14	$D-pa(A_4T_4)$		$61.2^{\rm d})$		-13.5^{d} -59.9^{d}	$-46.4^{\text{ d}}$	[3a]
15	$D-pa(T_4A_4)$		$69.4^{\rm d}$)		-14.5^{d} -57.6^{d} -43.0^{d}		$\lceil 3a \rceil$
16	$D-pa(T_4A_3)$		47.5^{d})				
17	p -pa $(ATATA)$		29.2^{d})				
18	D-pa(TATAT)		$21.0d$)				
19	D-pa(ATATATA)		$57.4^{\rm d}$)				
20	D-pa(ATATATAT)		$60.0d$)				[3a]
21	D-pa(TATATATA)	$\overline{}$	$60.8d$)				[3a]
22	pa(CGAATTCG)	$\qquad \qquad -$	78.1 $(1 \mu M)$				

Table 4. T_m Values and Thermodynamic Data of α -Arabinopyranosyl-(4' \rightarrow 2')-duplexes

^a) Measurements were made in 0.01m NaH₂PO₄, 0.1 mm Na₂(EDTA) buffer, pH 7.0, unless otherwise indicated. Estimated error of T_m determination $\pm 0.5^{\circ}$. $-$: not measured. ^b) Thermodynamic data from plots of T_m^{-1} vs. ln c; experimental error estimated in ΔH values $\pm 5\%$. ^c) Self-pairing. ^d) T_m measured in 0.01*M Tris* · HCl buffer, pH 7.0. e) T_m measured with no NaCl.

D-xylo series (*Fig.* 5; also Fig. 2 in [3b]). This parallels the conformational relationships that are presumed to prevail within the pentopyranosyl family (see Scheme I). The parallelism reappears in intersystem cross-pairing studies (*Table 5* and *Fig. 6*). The T_m values of the duplexes formed between the pa and pl sequences are consistently higher than any duplex formed from any sequence combinations of pa- with either pr- or pxoligomers (Table 5). This intersystem cross-pairing behavior pervades the entire family [3] (Table 5). Not surprisingly, there is no intersystem cross-pairing between α -Larabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides and the corresponding complementary RNA and DNA sequences; this is similar to the behavior exhibited by the other three pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide members [1] [3 - 5].

All observations made both in the inter- and intrasystem pairing studies are consistent with the *Watson–Crick* base-pairing mode in antiparallel strand orientation. This is confirmed for the α -L-arabinopyranosyl duplex formed by the self-complementary strand pa(4'-CGAATTCG-2') by NMR structure analysis [8].

4. Discussion. - Among the four experimentally studied members of the family of pentopyranosyl-(4' \rightarrow 2')-oligonucleotide systems, the α -arabinopyranosyl member is,

Entry	Oligonucleotide systems	$T_{\rm m}$ [°] Intersystem duplexes 1M NaCl $(c \approx 5 + 5 \mu M)^a$	For comparison: Tm of Intrasystem duplexes 1M NaCl $(c \approx 5 + 5 \mu M)^{a})^{b}$	
	$(A_8) + (T_8)$			
1	$pa + pr$	44.7	$pa + pa$	78.7
$\overline{2}$	$pr + pa$	59.4	$pl + pl$	51.0
\mathfrak{Z}	$pa + px$	42.5	$px + px$	$47.3*$
$\boldsymbol{4}$	$px + pa$	54.7	$pr + pr$	45.5
5	$pa + pl$	66.9	$r + r$	23.0
6	$pl + pa$	67.8		
	$(A_{12}) + (T_{12})$			
$\overline{7}$	$pa + pr$	67.8	$pa + pa$	> 95
$\,$ 8 $\,$	$pr + pa$	83.0	$pl + pl$	74.3
9	$pa + px$	69.5	$px + px$	73.2
10	$px + pa$	83.0	$pr + pr$	67.6
11	$pa + pl$	89.3	$r + r$	53.5
12	$pl + pa$	> 90		
	$(TATTTAAA) + (TTAAATA)$			
13	$pa + pr$	45.3	$pa + pa$	74.6
14	$pr + pa$	53.6	$pl + pl$	46.4*
15	$pa + px$	53.1	$px + px$	44.4
16	$px + pa$	45.8	$pr + pr$	45.9
17	$pa + pl$	61.2	$r + r$	17.2
18	$pl + pa$	62.3		
	$(ATTCAGCG) + (CGCTGAAT)$			
19	$pa + pr$	70.2	$pa + pa$	> 88
20	$pr + pa$	66.8	$pl + pl$	61.9
21	$pa + pl$	73.3	$px + px$	
22	$pl + pa$	73.6	$pr + pr$	67.6
			$r + r$	52.0

Table 5. T_m Values of Duplexes Formed by Inter- and Intrasystem Base Pairing of α -L-Arabinopyranosyl-(4' \rightarrow 2)-oligonucleotides

^a) Measurements were performed in 0.01M NaH₂PO₄, 0.1 mm Na₂(EDTA) buffer, pH 7.0, unless otherwise indicated. ^b) For T_m data and thermodynamic data of pa/pa, pl/pl, px/px, pr/pr, and r/r duplexes determined in 0.15м NaCl, see Table 2 in ref. [3b]. Estimated error of T_m determination \pm 0.5°. All T_m values taken are from [3b]. $pr = \beta$ -D-Ribopyranosyl, $px = \beta$ -D-xylopyranosyl, $pl = \alpha$ -L-lyxopyranosyl, $pa = \alpha$ -L-arabinopyranosyl, $r =$ RNA. Values with $(*)$ were measured in 0.01 M Tris \cdot HCl buffer, pH 7.0.

in a way, the most informative one when it comes to comparatively analyzing the family's base-pairing properties. Some of the structural factors deemed to be responsible for these remarkable properties express themselves most clearly in the structure of the α -arabinopyranosyl system. To these factors belong the conformational unambiguity and rigidity of the pyranosyl chairs, the fact that, in each member, the (relatively) small set of least-strained (idealized) conformations accessible to the repeating unit of a single strand contains at least one repetitive [3] [7] pairing conformation, and, finally, that there is in each of these pairing conformations a large inclination between the (averaged) strand axis and the (averaged) nucleobase pairing axis. The first and second factors are expected to contribute to the conformational preorganization of pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide single strands towards their duplex conformation. The third factor is responsible for the two most-characteristic features by which pentopyranosyl- $(4'\rightarrow 2')$ -oligonucleotide duplexes structurally distinguish themselves from the

Fig. 2. UV and CD data documenting the pairing behavior of A,T-containing $(4' \rightarrow 2')$ -a-L-arabinopyranosyl sequences. a) Self-pairing of the homobasic strands $pa(A_8)$, $pa(A_{12})$, $pa(T_8)$, and $pa(T_{12})$ as demonstrated by UV-spectroscopic T_m curves (heating). b) UV-Spectroscopic T_m curves (heating) of selected duplex formation. c) Temperature-dependent CD curves of $pa(T_8)$ demonstrating T/T self-pairing (temperature range: $0^\circ \rightarrow 55^\circ$). d) UV-Spectroscopic T_m curves of the duplex formed by $pa(4'-TTAAAATA-2')$ with its antiparallel complement pa(4'-TATTTTAA-2') and self-pairing T_m curves of the individual strands. All measurements were made in 10 mm aq. NaH₂PO₄ buffer containing 0.1 mm Na₂(EDTA), 1m NaCl at pH 7.0, unless otherwise indicated. Total oligonucleotide concentrations in all measurements were *ca*. 10 μ m. All T_m curves were fully reversible (no hysteresis). T_m Values are calculated from the maxima of the first derivative curve using kaleidagraph software program.

natural nucleic acids, namely, their quasi-linear (as against helical) duplex conformation and their exclusive adoption of an interstrand (as against intrastrand) base-stacking mode. A large inclination between backbone and nucleobase axes allows the nucleobases to reach optimal stacking distances without the need for strand helicalization²⁴), implying that base stacking within the duplex has to be interstrand²⁵).

²⁴⁾ Strand constriction by either helicalization or inclination are the two major pathways available to a 1,3 phosphodiester-based (6-bond) oligonucleotide strand to lower the distance between vicinal nucleobases and reach optimal stacking distances (see, e.g., [28]).

²⁵⁾ To the extent that nucleobase stacking can be a source of stabilization due to attractive forces between the stacking partners π -systems, interstrand stacking in a duplex is expected to act as a glue between the strands, whereas intrastrand stacking is not, at least to the extent that it can persist in a single strand after de-duplexation.

Fig. 3. UV-T_m, Thermodynamic, and 'mixing-curve' data documenting the pairing behavior of G,C-containing $(4' \rightarrow 2')$ -a-L-arabinopyranosyl sequences. a) UV-Spectroscopic T_m melting curves for pairing between pa(G₆) and $pa(C_6)$, and corresponding curves of individual strands. For conditions of measurements, see caption of Fig. 2. T_m Curves fully reversible (no hysteresis). b) Thermodynamic data of the duplex pa(G_6)/pa(C_6) deduced from concentration dependence of UV-spectroscopically determined T_m values. T_m Values measured in 10 mm aq. NaH₂PO₄ containing 0.1 mm Na₂(EDTA), 150 mm NaCl at pH 7.0; for method, see [26]. c) Molarratiodependence of UV absorption (275 nm, 20 $^{\circ}$) ('mixing curve') considered to indicate a 1:1 duplex stoichiometry for the pairing between $pa(G_6)$ and $pa(C_6)$. Data measured in 10 mm aq. NaH₂PO₄ containing 0.1 mm $Na₂(EDTA)$, 150 mm NaCl at pH 7.0 with a total oligonucleotide concentration of 10 μ m; for method, see [27].

As delineated in the Introduction, there are two conformational subgroups to be distinguished among the four $(4' \rightarrow 2')$ -pentopyranosyl systems investigated (*Scheme 1*); they differ in the conformation of their $4^{\prime},2^{\prime}$ -phosphodiester links; in one

Fig. 4. UV and CD data documenting the pairing behavior of indicated A,T,C,G-containing octamer sequences. a) UV-Spectroscopic T_m curves and b) Temperature-dependent CD curves (temperature range: $55^\circ \rightarrow 95^\circ$). For conditions of measurements, see caption of Fig. 2. T_m Curves fully reversible (no hysteresis). c) UV-Spectroscopic T_m curves of duplex formed by the self-complementary pa(4'-ATTCAGCG-2') sequence ($c \approx$ 1 µм, 0.15м NaCl, 10 mм aq. NaH₂PO₄ containing 0.1 mm Na₂(EDTA), 150 mm NaCl at pH 7.0) [8].

group (β -D-ribo and β -D-xylo), this link is 4',2'-diequatorial, and 4'-axial/2'-equatorial in the other (α -L-lyxo and α -L-arabino). For each of these two groups, a representative NMR structure analysis has been carried out by Jaun and co-workers, with the duplex

Fig. 5. Temperature-dependent CD curves documenting intra- and intersystem pairing between: a) $pa(A_8)$ and $pa(T_s)$, b) $pa(A_s)$ and $pl(T_s)$, c) $pa(A_s)$ and $pr(T_s)$, and d) $pa(A_s)$ and $px(T_s)$. Measurements in 10 mm aq. NaH₂PO₄ containing 0.1 mm Na₂(EDTA), 1m NaCl at pH 7.0. Total oligonucleotide concentrations were *ca*. 10 µm.

Fig. 6. UV-Spectroscopic T_m curves documenting intra- and intersystem pairing between pa(A_8) and pa(T_8), $pa(A₈)$ and $pl(T₈)$, $pa(A₈)$ and $pr(T₈)$, and $pa(A₈)$ and $px(T₈)$. For conditions of measurement, see caption of Fig. 5.

of the self-complementary octamer sequence 4'-CGAATTCG-2' in the β -D-ribopyranosyl [4b] and the α -L-arabinopyranosyl series [8]. The latter study was described and compared with the former in a recent paper [8], so that we can restrict ourselves here to recapitulate those aspects of the α -L-arabinopyranosyl structure that are considered to have a direct bearing on the system's pairing properties.

The axial attachment of the phosphodiester bridge at $C(4')$ in the α -arabino structure compared to its equatorial attachment in the β -ribo series removes a specific source of steric encumberment in duplexes of the latter, namely, the intrastrand steric contact between the equatorial H-atom of the $CH₂(5')$ group of the pyranose chairs and the π -face of the neighboring nucleobase upstream²⁶). The steric confrontation between these two groups is a direct consequence of the di-equatorial positioning of the phosphodiester bridge at $C(4')$ and $C(2')$. It disappears, as Fig. 7 indicates, with the change of the phosphodiester attachment at $C(4')$ from equatorial to axial. The change clearly reflects itself in chemical shifts of the critical $CH₂$ H-atom in the 1 H-NMR spectra of the α -arabinopyranosyl vs. the β -ribopyranosyl octamer duplex: in the spectrum of the latter the critical H-atoms appear shielded by the ring current of a neighboring purine base, whereas the corresponding H-atoms in the α -arabinopyranosyl duplex are not (see Fig. 7).

Base pairing in α -L-arabinopyranosyl (4'-OP axial) duplexes is found to be always much stronger than in corresponding β -D-xylopyranosyl (4'-OP equatorial) duplexes (Table 5 and [1]), and, furthermore, α -L-lyxopyranosyl duplexes tend to be more stable than β -D-ribopyranosyl and β -D-xylopyranosyl duplexes (*Table 5* and [5]). This pattern of relative strength of base pairing is consistent with the assumption that the intrastrand steric contact between pyranose rings and neighboring nucleobases in β -ribo- and β xylopyranosyl duplexes does indeed act as steric repulsion that lowers duplex stability, and that it is the absence of this destabilization in the α -arabino- and α -lyxopyranosyl series that, at least in part, makes their duplexes more stable. The question, however, why it is that α -arabinopyranosyl duplexes are so much more stable than their α lyxopyranosyl analogs, remains.

Fig. 8 points to a conformational divergence between α -arabino- and α -lyxopyranosyl duplexes that can be expected to contribute to this difference in pairing strengths. The figure indicates, within the simplifying constraints of idealized conformations, that the diversity of accessible conformers around the axial 4 attachment of the phosphodiester bridge in a α -lyxopyranosyl single strand is larger than in a α -arabinopyranosyl single strand. The equatorial position of the 3'-OH group in the latter precludes a population of the family of conformers in which the 4-O-P bond of the phosphodiester group is *anti-periplanar* to the arabinopyranose ring's endocyclic $C(4') - C(5')$ bond, in contrast to the α -lyxopyranosyl series, where the population of this family of conformations is unfettered (Fig. 8, b vs. a, see also e). In both systems, on the other hand, the family of conformers that includes the actual pairing conformations27) is sterically unencumbered in both single and duplex strands of the arabino- and lyxopyranosyl series (*Fig. 8,d* and *c*). This means that, at least with respect to this C(4')-domain of the conformational space of pentopyranosyl- $(4' \rightarrow 2')$ -

²⁶⁾ $\text{`Upstream'} = \text{in the } (2' \rightarrow 4')\text{-direction as against the (downstream) } (4' \rightarrow 2')\text{-direction.}$

²⁷) $4'O-P$ Bond *anti*-periplanar to endocyclic $C(4') - C(3')$ bond.

Fig. 7. $\,^1H\text{-}NMR$ Chemical shifts of equatorial and axial H-atoms at C(5') in the (4'-CGAATTC-2') $_2$ duplex in the β - b -ribopyranosyl-(4' \rightarrow 2')- and the a -t-arabinopyranosyl-(4' \rightarrow 2')-series. The data are taken from [4b] and [8] and refer to the nucleoside-units T5 and T6, respectively.

oligonucleotide strands, the single strand of α -arabinopyranosyl nucleotides is more pre-organized toward the pairing conformation than the single strand of α -lyxopyranosyl nucleotides. This constraint in the population of single-strand conformers in the arabinopyranosyl series would be expected to contribute to higher duplex stability.

Fig. 8. Qualitative conformational reasoning referring to the phosphodiester conformations in the C(4')- and $C(2')$ -domains for both the α -D-lyxopyranosyl and α -L-arabinopyranosyl series. The sign $+$ denotes allowed and - disallowed conformational populations according to steric and stereoelectronic criteria. Bold arrow points to
region of strong steric repulsion in (idealized) pairing conformation. For the rules followed in this disallowed conformational populations according to steric and stereoelectronic criteria. Bold arrow points to conformational analysis, see [4a][7].

Appealing as the above argument may appear, it should not detract from recognizing the complexity of the problem of rationalizing relative base-pairing strengths, even in the case of structurally rather closely related oligonucleotide systems, as long as experimental structural information about the single strands involved is absent. One is quickly reminded of this complexity by inspecting b in Fig. 8, by which it becomes clear that there is no way to reliably predict by qualitative conformational reasoning the type of conformation the phosphodiester group will assume in the $C(2')$ domain in either a single- or a double-stranded α -arabinopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide²⁸). In fact, it is in this domain that the NMR structure analysis [8] in the arabino series uncovered a large deviation of the phosphodiester torsion angles from the values belonging to the corresponding (formally highly strained) idealized pairing conformation [8]. It was rather surprising to find, in this analysis, that the type of phosphodiester conformation adopted in the C(2)-domain of the arabinopyranosyl duplex is the same as in the corresponding ribopyranosyl duplex. This can only mean that a pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide strand has, in this domain, no significantly more-stable alternative conformation available; if it had, and that alternative would be a pairing conformation, the NMR structure analysis would have revealed it; if, however, that alternative were not a pairing conformation, the actually observed base-pairing strengths in α -arabinopyranosyl-(4' \rightarrow 2')-oligonucleotide duplexes could not possibly be as high as observed. We are led to conclude, therefore, that, not only with respect to the domain of the C(4)-phosphodiester attachment, but also with respect to the C(2')-domain, α -arabinopyranosyl-(4' \rightarrow 2')-oligonucleotide single strands are pre-organized toward duplexation in spite of the pairing conformation in the latter domain being highly strained. Such reasoning would lead to the expectation that arabinopyranosyl duplexes – irrespective of being more stable relative to their respective single strands $-\text{would have to be thermodynamically less stable against, } e.g.,$ strand scission compared to duplexes of the β -ribopyranosyl series. Indeed, a much higher propensity of α -arabinopyranosyl strands toward strand scission under the basic conditions of protection-group removal after solid-support synthesis has been one of the specific observations made in the α -arabinopyranosyl-(4' \rightarrow 2')-oligonucleotide series²⁹).

The CD spectra reproduced in Fig. 5 refer to intersystem cross-pairing within the pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide family. These spectra convincingly support the assignments of pairing conformations for those two members of the family for which no NMR structure analysis is available $(Scheme 1)$. In full harmony with the relationships expressed by these spectra, it is found that the combination of α arabinopyranosyl strands and complementary α -lyxopyranosyl strands in intersystem cross-pairing leads to duplexes that are consistently more stable than the corresponding combination between α -arabinopyranosyl strands and β -ribo- or β -xylopyranosyl strands (Table 5, Fig. 6, and [3b]). In such intersystem cross-pairings between homobasic purine and homobasic pyrimidine strands duplex stabilities can strongly depend on which backbone bears the pyrimidine, and which bears the purine sequence. This remarkable phenomenon has been discussed at some length in a previous communication [3b].

Exceptionally strong base pairing such as that seen in α -arabinopyranosyl oligonucleotides can be accompanied by the occurrence of strong self-pairing of

²⁸) Quite in contrast to the α -lyxopyranosyl series where the $(C(2')-O)$ torsion angles of both single and double strands are (qualitatively) clearly predictable (Fig. 8,a).

²⁹) Not surprisingly, comparable lability toward strand scission has been observed in the β -xylopyranosyl series [1].

nucleobases. Such self-pairing may be quite selective for a given purine or pyrimidine base and be specific for a given type of backbone (see Table 1 in [3b]). For example, whereas homo-adenine sequences undergo strong (*Hoogsteen*) self-pairing in the homo-DNA series [28], no such pairing is observed in pyranosyl-RNA [4a], and almost none in the α -arabinopyranosyl series (*Table 4*). The situation is more or less inverse with regard to the self-pairing of thymine. The homo-pyrimidine sequence $pa(T_{12})$ shows the remarkable T_m value of 52 $^{\circ}$ (10 μ m, 1m NaCl), yet (fortunately) still surpassed by the value ($>$ 95 $^{\circ}$) for the corresponding canonical A – T pairing in p $\mathbf{a}(\mathrm{A}_{12})$ / $pa(T_{12})$ (Table 4). Understandably, no such self-pairing seems to occur with cytosine³⁰). The possible self-pairing mode for thymine has been discussed in a previous communication (see Footnote 12 in [3b]).

An ominous case of noncanonical self-pairing of non-self-complementary A,Tcontaining strands was encountered in the behavior of the two antiparallel-complementary octamer sequence pa(TTAAAATA) and pa(TATTTTAA). Both undergo strong self-association to duplexes, which have $T_{\rm m}$ values of *ca*. 48° under conditions where the canonical cross-pairing duplex melts around 75° 31). This case may be taken as a reminder of at least two questions that are of specific interest in the etiological context, namely, the one concerning the relationship between base-pairing strength and performance as an informational base-pairing system, and the other referring to the limitations of an informational system that would have to operate with two instead of four nucleobases. We have discussed the first question previously in the context of our work on template-directed sequence copying in the ribopyranosyl- $(4' \rightarrow 2')$ -oligonucleotide series [4c] [4e]. With regard to the two-instead-of-four nucleobase question (see $e.g., [29]$), the case of self-association of the two-base arabinopyranosyl strands mentioned above is just echoing a general experience we made in the course of our extensive studies on pentopyranosyl- $(4' \rightarrow 2')$ -oligonucleotides, namely: the higher the base-pairing strengths in an oligonucleotide system, the more difficult it becomes to construct (with only two nucleobases) a sequence that does not show self-pairing, even when the sequence is very short (e.g., octamers). The task of constructing non-selfassociating sequences of corresponding lengths with four instead of only two nucleobases is obviously much easier. This difference is assumed to be an important aspect in any conceptual and/or experimental evaluation of potentially primordial informational oligomers.

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³⁰) Neither pa(C₆) nor pa(G₆) show discernable self-pairing under identical conditions ($c \approx 10 \,\mu$ m in 10 mm Na₂HPO₄ buffer, 0.1 mm Na₂(EDTA), pH 7.0 with 1m NaCl). The self-pairing mode conceivable for cytosine would require parallel strand orientation (forbidden in systems with large backbone inclination [28]).

³¹⁾ The duplex strand formed by the A-rich sequence is presumably held together by four A/T base pairs, as well as four A/T and A/A interstrand base stacks.

Experimental Part

General. Solvents for extraction: ACS grade. Solvents for reaction: reagent grade. Reagents: unless otherwise noted, from Acros, Fluka, or Aldrich, highest quality available. Chloro(2-cyanoethoxy)(diisopropylamino)phosphine (97%) was purchased from CheM - Impex Inc., Wood Dale, IL, USA. TLC: silica gel 60 F_{254} aluminum plates (Whatman, Type Al Sil G/UV, 250 µm layer); visualization by UV absorption and/or A) by dipping in a soln. of 4-anisaldehyde /H₂SO₄/AcOH/EtOH 5:5:1:18 or B) cerium(IV) sulfate (3 mm)/ammonium molybdate (250 mm) in aq. H₂SO₄ (10%), followed by heating. Flash column chromatography (CC) was performed on silica gel 60 (0.40 - 0.63 mm, 230 - 440 mesh; *EM Science*) at low pressure (max. 2 bar). In case of acid-sensitive compounds, the silica gel was pretreated with solvents containing $ca. 0.5\%$ Et₃N. Melting points (uncorrected) were measured with MEL-TEMP II (Laboratory Devices Inc., USA). NMR: $H: \delta$ values in ppm (TMS as internal standard); J [Hz], assignments of ¹H resonances were in some cases based on 2D experiments (${}^{1}H$, ${}^{1}H-$ COSY); ¹³C: δ values in ppm (TMS as internal standard); *J* [Hz]; assignments and multiplicities were based on 2D experiments (${}^{1}H,{}^{13}C\text{-COSY}$); ${}^{31}P$: δ values in ppm (85% H_3PO_4 as external standard). FAB⁺-MS (matrix-soln.): m/z (intensity in %); performed in the positive-ion mode, with 3-nitrobenzyl alcohol (3-NBA) as matrix, on a VG ZAB-VSE double-focusing high-resolution mass spectrometer equipped with a cesium ion gun. Matrix-assisted laser-desorption-ionization time-of-flight mass spectrometry (MALDI-TOF-MS) was performed on a Voyager-Elite mass spectrometer (Perseptive Biosystems) with delayed extraction with THAP or DHB as the matrix with ammonium citrate added to the sample. Elemental analysis was performed with Perkin-Elmer PE2400 CHN analyzer. Oligonucleotides were synthesized on an Expedite 8909 Nucleic Acid Synthesis system (Perseptive Biosystems). HPLC: Ion exchange (IE) HPLC was performed on Pharmacia Äkta Purifier (900) controlled by UNICORN system. Columns: Mono Q HR 5/5 (Pharmacia) or SAX 1000-8 (Macherey & Nagel); Buffer A: $10 \text{ mm Na}_2\text{HPO}_4$ in H₂O, pH 11.5; Buffer $B:10 \text{ mm Na}_3\text{HPO}_4$ in H₂O, 1m NaCl, pH 11.5. UV Spectra: on a Cary 1 C spectrophotometer (Varian). Melting-point (T_m) measurements of oligonucleotides were determined with Cary 1 Bio spectrophotometer (Varian). CD Spectrum was measured on an AVIV 61 DS CD spectropolarimeter. All measurements were made with the 'phosphate buffer', 10 mm aq. NaH₂PO₄ buffer containing 0.1 mm Na₂EDTA, 150 mm (or 1m) NaCl at pH 7.0, with a total oligonucleotide concentration of ca. 10 µm, unless indicated otherwise, and the samples were thoroughly degassed, either by heating or by vacuum and ultrasonication. Concentrations of oligonucleotide solns. were calculated from the UV absorbance of the solns. at 260 nm (pH 7) at *ca*. 80 $^{\circ}$ with the following molar extinction coefficients: ε (pa(A)) = 15000, ε (pa(T)) = 10000, ε (pa(C)) = 8400, ε (pa(G)) = 11900. Abbreviations: Ag-Tf = silver trifluoromethanesulfonate, BOP: (benzotriazol-1-yl)oxytris(dimethylamino)phosphonium hexafluorophosphate; $BSA = bis(t$ rimethylsilyl)acetamide, $CPG = 'controlled$ -pore glass', DHB = 2,5dihydroxybenzoic acid, DIBAL-H = diisobutylaluminium hydride, DMAP: 4-(dimethylamino)pyridine, DMT: $4.4'$ -dimethoxytrityl, Fmoc = $[(9H$ -Fluoren-9-yl)methoxylcarbonyl, 1-HOBT = 1-Hydroxy-1H-benzotriazole, LCAA-CPG: long chain aminoalkyl-CPG (500 ä); LCAMA-CPG: long-chain aminomethylalkyl-CPG (500 Å) ; PAC = phenoxyacetyl, TMS-Tf = trimethylsilyl trifluoromethanesulfonate, THAP = 2,4,6-trihydroxyacetophenone, TOTU: O-{[(2-cyanoethoxycarbonyl)methylidene]amino}-1,1,3,3-tetramethyluronium tetrafluoroborate, TSOH = toluene-4-sulfonic acid, UF $H_2O =$ ultra-filtered H_2O .

1. Experiments Referring to Scheme 3. – 1',2',3',4'-Tetra-O-benzoyl- α/β -t-arabinopyranose (1) [11]. To an ice-cooled mixture of 600 ml of dry pyridine and 500 ml of dry CH₂Cl₂, over a period of 15 min, were added 200 ml (1.7 mol) of BzCl. To this mixture, 50 g (0.33 mol) of $(+)$ -L-arabinose was added in small portions over 5 h. The mixture was allowed to warm slowly to r.t., and stirred overnight. The syrupy mixture was poured into 1000 ml of ice H₂O, the org. layer was separated and washed twice with 400 ml of sat. aq. NaHCO₃ soln. The org. layer was dried (Na_2SO_4) and evaporated, and the pale yellow oil was co-evaporated twice with 100 ml of toluene. The residue was crystallized from 200 ml of hot EtOH. After storage at 4° overnight, the crystals were filtered, washed with cold EtOH, Et₂O and dried under high vacuum (0.03 Torr/r.t., 12 h) to yield 161.1 g (88 %) of 1. Colorless crystals. M.p. 174–176°. TLC (hexane/AcOEt 3:1): R_f 0.36. ¹H-NMR (600 MHz, CDCl₃): 4.20 $(dd, J=2.0, 13.4, H-C(5))$; 4.42 $(dd, J=0.8, 13.4, H-C(5))$; 5.92 $(m, H-C(4))$; 6.0 $(m, H-C(3), H-C(2))$; 6.88 $(d, J = 1.6, H - C(1))$; 7.32 $(m, 4 \text{ arom. H})$; 7.49 $(m, 2 \text{ arom. H})$; 7.54 $(m, 4 \text{ arom. H})$; 7.66 $(m, 2 \text{ arom. H})$; 7.90 (m, 4 arom. H); 8.15 (m, 4 arom. H). ¹³C-NMR (150 MHz, CDCl₃): 63.42 (t, C(5')); 68.15, 68.51 (2d, C(3'), C(2)); 69.88 (d, C(4)); 91.47 (d, C(1)); 128.81, 128.83, 129.04, 129.16, 129.30, 129.50, 129.72, 130.15, 130.31, 130.31, 133.83, 133.99, 134.26 (arom. C); 165.11, 165.99, 166.15, 166.17 (4s, CO of Bz). ESI-MS (pos., MeOH): 605 (5, $[M + K]^+$), 589 (100, $[M + Na]^+$), 467 (4, $[M + Na - OBz]^+$), 445 (19, $[M - OBz]^+$), 323 (6, $[M - 2OBz]^+$). Anal. calc. for $C_{33}H_{26}O_9$: C 69.96, H 4.63; found: C 70.17, H 4.39.

 N^6 -Benzoyl-9-(2',3',4'-tri-O-benzoyl- α -L-arabinopyranosyl)adenine (2a). To a suspension of 24.9 g (0.10 mol) of N^6 -benzoyladenine and 63.7 g (0.10 mol) of 1 in 450 ml of dry MeCN, 50 ml $(0.2 \text{ mol}, 2 \text{ equiv.})$

of BSA was added, and the mixture was heated to 65° in an oil bath. After 1.5 h, 42 ml (0.37 mol) of SnCl₄ was cautiously added (vigorous boiling) over 20 min. The clear soln. was maintained at 65° for further 90 min until TLC showed complete consumption of the starting material. The mixture was cooled to r.t. and poured into 500 ml of ice-cold sat. aq. NaHCO₃ soln. and 500 ml of AcOEt. The org. layer was separated, washed several times with sat. aq. NaHCO₃ soln., and washed with sat. aq. NaCl soln. The org. phase was dried (Na₂SO₄), filtered, and concentrated in vacuo. The crude product was purified by CC (silica gel; $CH_2Cl_2/MeOH$ 100 : 1 to 100 : 4). The product fractions were combined and evaporated to yield 59.5 g (84%) of 2a. Colorless foam. TLC (toluene/AcOEt 1:1): R_f 0.24. ¹H-NMR (600 MHz, (D₆)DMSO): 4.35 (dd, J = 1.9, 13.1, H – C(5')); 4.49 (d, J = 13.1, $H-C(5')$; 5.87 (m, $H-C(4')$); 6.06 (dd, $J = 3.4$, 9.9, $H-C(3')$); 6.46 (d, $J = 9.1$, $H-C(1')$); 6.65 (t, $J = 9.5$, H-C(2)); 7.37 (m, 3 arom. H); 7.54 (m, 5 arom. H); 7.62 (m, 1 arom. H); 7.69 (m, 5 arom. H); 7.77 (m, 1 arom. H); 7.93 (dd, 1 arom. H); 8.02 (d, 2 arom. H); 8.31 (m, 2 arom. H); 8.86 (s, H – C(8)); 8.89 (s, H – C(2)); 11.22 (s, $H-N(6)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 67.38 (t, C(5')); 69.61 (d, C(2')); 69.77 (d, C(4')); 72.72 (d, C(3')); 82.38 (d, C(1)); 126.33 (s, C(5)); 129.29, 129.32, 129.36, 129.41, 129.57, 129.63, 129.85, 129.89, 129.91, 130.11, 130.67, 133.31, 133.71, 134.63, 134.75, 134.80 (arom. C); 144.33 (d, C(8)); 151.43 (s, C(4)); 152.87 (d, C(2)); 153.07 (s, C(6)); 165.16, 165.46, 165.92 (s, CO of OBz); 168.18 (s, CO of NHBz). FAB-MS(pos., 3-NBA/CsI): 816.1035 (100, $[M + Cs]^+$). Anal. calc. for $C_{38}H_{29}N_5O_8$: C 66.76, H 4.28, N 10.24; found: C 66.40, H 4.43, N 10.46.

 $1-(2',3',4'-Tri-O-benzoyl-a-L-arabinopyranosyl)thymine$ (2b). To a suspension of 13.4 g (0.11 mol) of thymine and $60.0 g (0.11 mol)$ of 1 in 450 ml of dry MeCN, 50 ml $(0.2 mol)$ of BSA was added, and the mixture was heated to 60°, during which time all solid material dissolved. After 45 min, 57.7 ml (0.33 mol) of TMS-Tf was added, and the soln. was maintained at 60° for 4.5 h. The mixture was cooled to r.t., poured into a mixture of 600 ml of ice-cold sat. aq. NaHCO₃ soln. and 600 ml of AcOEt. The org. layer was separated, washed with 500 ml of sat. aq. NaHCO₃ soln., 500 ml of sat. aq. NaCl soln., and dried (Na₂SO₄). The solvent was removed in vacuo, and the residue was crystallized from 150 ml of AcOEt to yield 50.1 g (83%) of 2b. Colorless needles. M.p. $222 -$ 224° TLC (toluene/AcOEt 1:1): R_f 0.59. ¹H-NMR (600 MHz, (D₆)DMSO): 1.85 (s, Me – C(5)); 4.28 (dd, J = 1.7, 13.2, H – C(5')); 4.42 (d, J = 13.2, H – C(5')); 5.75 (m, H – C(4')); 5.86 (br. s, H – C(2')); 6.02 (dd, J = 3.4, 9.8, $H-C(3')$; 6.23 (br. s, $H-C(1')$); 7.36 (t, 2 arom. H); 7.45 (t, 2 arom. H); 7.55 (t, 1 arom. H); 7.61 (t, 3 arom. H); 7.69 (m, H-C(6), 2 arom. H); 7.74 (t, 1 arom. H); 7.80 (d, 2 arom. H); 8.41 (d, 2 arom. H); 11.43 (s, H-N(3)). ¹³C-NMR (150 MHz, (D₆)DMSO): 13.02 (Me-C(5)); 67.45 (t, C(5')); 69.75, 69.86 (C(4'), C(2')); 72.28 (d, $C(3')$; 80.92 (d, $C(1')$); 111.40 (s, $C(5)$); 128.98, 129.38, 129.56, 129.74, 129.81, 129.89, 130.02, 130.49 (arom. C); 134.61, 134.73, 134.86 (arom. C, C(6)); 151.25 (C(2)); 164.24 (C(4)); 165.33, 165.58, 165.96 (s, CO of Bz). ESI-MS (pos., MeOH): 593 (100, $[M + Na]$ ⁺), 571 (6, $[M + H]$ ⁺), 471 (16, $[M + Na - OBz]$ ⁺), 445 (12, $[M + H -$ Thy]⁺), 353 (10, $[M + H + Na - 2OBz]$ ⁺), 323 (5, $[M + H - Thy - OBz]$ ⁺). ESI-MS (neg., MeOH): 569 (100, $[M-H]$ -), 540 (26, $[M-H-CO]$ -). Anal. calc. for $C_{31}H_{20}N_2O_9$: C 65.26, H 4.59, N 4.91; found: C 65.23, H 4.81, 4.83.

 N^4 -Benzoyl-1-(2',3',4'-tri-O-benzoyl-a-L-arabinopyranosyl)cytosine (2c). A suspension of 16.5 g (74 mmol) of 4-N-benzoylcytosine, 45.52 g (81 mmol) of 1, and 46 ml (185 mmol) of BSA was heated in 300 ml of dry MeCN at 80° in an oil bath. When all of the material dissolved (*ca*. 30 min), 26 ml (222 mmol) of SnCl₄ was cautiously added (vigorous boiling). Stirring at 80° was continued for 30 min, until TLC indicated complete reaction. The yellow soln. was allowed to cool to r.t., and poured into 600 ml of ice-cold sat. aq. NaHCO₃ soln. and 400 ml of AcOEt. The org. phase was separated, washed with sat. aq. NaHCO₃ soln., sat. aq. NaCl soln., and dried (Na₂SO₄). The solvent was removed under vacuum, and the residue was crystallized from 250 ml of MeOH to afford 40.2 g (82 %) of 2c. Colorless needles. TLC (toluene/AcOEt 1:1): R_f 0.3. ¹H-NMR (600 MHz, (D_6) DMSO): 4.34 (d, J = 13.2, H – C(5')); 4.49 (d, J = 13.2, H – C(5')); 5.81 (m, H – C(4')); 5.86 (br. s, H – C(2')); 6.08 $(m, H-C(3'))$; 6.48 (br. s, H $-C(1'))$; 7.36, 7.44, 7.49, 7.55 $(4m, 7 \text{ atom. H}, H-C(5));$ 7.61, 7.69, 7.80, 7.97, 8.16 $(4m, 13 \text{ atom. H})$; 8.40 (br. s, H-C(6)); 11.26 (br. s, H-N(4)). ¹³C-NMR (150 MHz, (D₆)DMSO): 67.81 (t, $C(5')$; 69.92 (d, $C(4')$; 70.66 (d, $C(2')$); 72.34 (d, $C(3')$); 82.07 (d, $C(1')$); 98.53 (d, $C(5)$); 129.24, 129.35, 129.54, 129.65, 129.82, 129.88, 130.10, 130.56, 133.62, 134.70 (arom. C); 146.55 (s, C(4)); 155.20 (s, C(2)); 164.38 (d, $C(6)$); 165.40, 165.69, 165.98 (s, CO of OBz); 168.31 (s, CO of NHBz). ESI-MS (pos., MeOH): 682 (14, [M + $\text{Na}^{\text{+}}$), 660 (3, $[M + H]^+$), 429 (100, $[M + H - H_2NBz]^+$). ESI-MS (neg., MeOH): 658 (100, $[M - H]^-,$), 536 (8, $[M-H-OBz]$ ⁻).

N²-Isobutyryl-9-(2',3',4'-tri-O-benzoyl-a-L-arabinopyranosyl)guanine (2d) and N²-Isobutyryl-7-(2',3',4'-tri-O-benzoyl-a-t-arabinopyranosyl)guanine (2e). To a suspension of 22.1 g (0.1 mol) of N^2 -isobutyrylguanine and 67 g (0.12 mol) of 1 in 450 ml of dry MeCN, 77 ml (0.35 mol) of BSA was added, and the mixture was heated under reflux. After 2 h, 58 ml (0.3 mol) of TMS-Tf was added to the clear soln., and heating was continued for further 6 h. The soln. was allowed to cool to r.t., and poured into a mixture of 500 ml of ice-cold sat. aq. NaHCO₃ soln. and 500 ml of AcOEt. The org. layer was separated, washed with sat. aq. NaHCO₃ soln., sat. aq. NaCl soln.,

dried (Na₂SO₄), and evaporated *in vacuo*. The resulting foam was purified by CC (silica gel, 26 \times 8 cm, toluene/ AcOEt 10:1 to 10:4, 10:5 to 10:8 and 10:9 to 10:20). Fractions containing the N^9 -isomer 2d were combined, stored at 4° to afford crystals, which were filtered, washed with AcOEt, and dried in vacuo (0.03 Torr/r.t., 12 h) to yield 26.0 g (44%) of 2d as colorless needles. The remaining fractions containing the N^7 -isomer 2e were combined, evaporated, and the residue was crystallized from 50 ml of AcOEt to afford 2.1 g $(3%)$ of 2e as colorless needles.

Data of 2d. TLC (toluene/AcOEt/MeOH 5 : 4 : 1): R_f 0.42. ¹H-NMR (600 MHz, (D_6) DMSO): 1.12, 1.13 (2d, $J = 6.8, Me₂CH$; 2.80 (m, Me₂CHCO); 4.39 (dd, $J = 1.5, 13.1, H-C(5')$); 4.45 (d, $J = 13.1, H-C(5')$); 5.86 (m, $H-C(4')$; 6.00 $(m, H-C(3'))$; 6.02 $(d, J = 9.5, H-C(1'))$; 6.10 $(dd, J = 9.5, 9.5, H-C(2'))$; 7.37 $(m, 4 \text{ atom. H})$; 7.55 (m, 2 arom. H); 7.63 (m, 2 arom. H); 7.68 (m, 4 arom. H); 7.76 (m, 1 arom. H); 8.18 (d, 2 arom. H); 8.49 (s, $H-C(8)$); 11.71, 12.04 (2 br. s, $H-N(2)$, $H-N(1)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.75 (*q*, Me₂CH); 19.64 (q, Me, CH) ; 35.55 (d, Me₂CHCO); 68.01 (t, C(5')); 69.78 (d, C(4')); 70.74 (d, C(3')); 72.29 (d, C(2')); 81.31 (d, C(1)); 120.52 (s, C(5)); 128.56, 129.25, 129.59, 129.82, 129.86, 129.94, 130.63 (arom. C); 134.70, 134.76, 134.84 (arom. C, C(8)); 149.32, 149.52 (C(2), C(4)); 155.47 (s, C(6)); 165.21, 165.35, 166.00 (s, CO of Bz); 181.02 (s, $Me₂CHCO$). ESI-MS (pos., MeOH): 688 (45, $[M + Na]$ ⁺), 359 (100, $[M + Na - 2 Obz - isobutyramide]$ ⁺), 323 $(47, [M+Na-3 \text{ OBz}]^+)$. ESI-MS (neg., MeOH): 664 (100, $[M-H]^-$). Anal. calc. for $C_{35}H_{32}N_5O_9$: C 63.06, H 4.84, N 10.50; found: C 63.33, H 4.96, 10.41.

Data of 2e. TLC (toluene/AcOEt/MeOH 5:4:1): R_f 0.54. ¹H-NMR (600 MHz, (D₆)DMSO): 1.12 (2d, Me_2CH); 2.72 (m, Me₂CHCO); 4.34 (d, J = 12.8, H – C(5')); 4.44 (br. d, H – C(5')); 5.85 (m, H – C(4')); 6.02, 6.33, 6.75 (3 br. s, H – C(3'), H – C(2'), H – C(1')); 7.35 (m, 2 arom. H); 7.38 (m, 2 arom. H); 7.54 (m, 2 arom. H); 7.62 (m, 2 arom. H); 7.71 (m, 4 arom. H); 7.73 (m, 1 arom. H); 8.34 (m, 2 arom. H); 8.60 (s, H-C(8)); 11.61, 12.18 (2 br. s, H-N(2), H-N(1)). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.63 (*q, Me*₂CH); 19.74 (*q, Me*₂CH); 35.56 (d, Me2CHCO); 67.69 (t, C(5)); 69.66, 70.94, 72.46 (d, C(4), C(3), C(2)); 84.63 (d, C(1)); 111.93, 128.80, 129.35, 129.52, 129.61, 129.81, 129.87, 129.95, 130.83, 134.59, 134.64, 134.72, 148.56, 152.67 (arom. C, C(5), C(8), $C(2)$, $C(4)$, $C(6)$); 165.13, 165.40, 166.16 (s, CO of Bz); 181.97 (s, Me₂CHCO). ESI-MS (pos., MeOH): 704 (17, $[M + K]^+$), 688 (100, $[M + Na]^+$). ESI-MS (neg., MeOH): 664 (100, $[M - H]^-$). Anal. calc. for $C_{35}H_{32}N_5O_9$: C 63.06, H 4.84, N 10.50; found: C 62.95, H 4.81, N 10.52.

9-(a-L-Arabinopyranosyl)-N⁶-benzoyl-adenine (3a). To a soln. of 50 g (73.1 mmol) of 2a in 600 ml of THF/ MeOH/H₂O 5:4:1 ($v/v/v$), cooled to -20° , 150 ml of 2_N aq. NaOH soln. was slowly added over a period of 25 min. Stirring at -20° was continued for further 75 min after which TLC showed complete reaction. The soln. was carefully neutralized with 35% aq. HCl soln. and evaporated, during which the product precipitated as a white solid. At this stage, the mixture was kept at 4° overnight, the precipitate was filtered, washed with acetone, and dried in high vacuum (0.03 Torr/r.t., 12 h) to yield 28.3 g (64.1 mmol, 88%) of 3a. White solid. TLC (CHCl₃/ MeOH 1:1): R_f 0.68. ¹H-NMR (600 MHz, (D₆)DMSO): 3.62 (*m*, H-C(3')); 3.79 (*m*, H-C(4'), H-C(5')); 3.85 $(d, J = 12.2, H - C(5'))$; 4.28 $(m, H - C(2'))$; 4.79 $(d, J = 5.0, HO - C(4'))$; 5.03 $(d, J = 5.7, HO - C(3'))$; 5.33 $(d, J = 5.7)$ $J = 5.8, HO - C(2^{\prime}))$; 5.48 (d, $J = 9.2, H - C(1^{\prime}))$; 7.55 – 8.06 (t, 5 arom. H); 8.62 (s, H-C(8)); 8.75 (s, H-C(2)); 11.17 (s, H – N(6)). ¹³C-NMR (150 MHz, (D₆)DMSO): 69.39 (d, C(2')); 69.63 (d, C(4')); 70.30 (t, C(5')); 74.29 (d, $C(2')$; 84.33 (d, $C(1')$); 126.18 ($C(5)$); 129.33, 133.30, 134.19 (arom. C); 143.83 (d, $C(8)$); 151.07 (s, $C(4)$); 152.57 $(d, C(2))$; 153.54 $(s, C(6))$; 166.43 $(s, CO$ of Bz). ESI-MS (pos., MeOH): 765 $(48, [2M + Na]^+)$, 394 $(46, [M +$ Na]⁺), 240 (100, [A^{Bz} + H]⁺). ESI-MS (neg., MeOH): 406 (14, [M + Cl]⁻); 370 (100, [M – H]⁻).

 $1-(\alpha$ -L-Arabinopyranosyl)thymine (3b). To a soln. of MeONa prepared from 2.5 g (0.11 mol) of Na and 500 ml of dry MeOH was added 50.1 g (88 mmol) of 2b. The mixture was stirred for 12 h at r.t., after which TLC showed complete consumption of the starting material. The white solid formed during the reaction was filtered and was shown to be the Na salt of the product isolated in almost quant. yield. To obtain the neutral form of the product, the suspension was neutralized with AcOH and 50 g of silica gel was added. The mixture was concentrated to dryness, and the residue was purified by CC (silica gel; CHCl₃/MeOH 1:1). Product fractions were combined, evaporated in vacuo, and the residue was crystallized from MeOH to yield 18.9 g (83 %) of 3b. Colorless needles. M.p. 249–250°. TLC (CHCl₃/MeOH 1:1): R_f 0.68. ¹H-NMR (600 MHz, (D₆)DMSO) 1.80 (s, $Me-C(5)$); 3.48 $(m, H-C(3'))$; 3.62 $(d, J=12.1, H-C(5'))$; 3.67 $(m, H-C(4'))$; 3.70 $(m, H-C(2'))$; 3.75 $(dd,$ $J = 1.7, 12.1, H - C(5'))$; 4.69, 4.93 (2d, $J = 5.9, J = 5.6, HO - C(4'), HO - C(3'))$; 5.22 (d, $J = 2.1, HO - C(2'))$; 5.22 $(d, J = 12.4, H - C(1'))$; 7.52 $(s, H - C(6))$; 11.40 $(s, H - N(3))$. ¹³C-NMR (150 MHz, (D_6) DMSO): 12.90 $(q,$ Me – C(5)); 69.06, 69.36 (d, C(2'), C(4')); 70.13 (d, C(5')); 74.10 (d, C(3')); 83.90 (d, C(1')); 110.21 (s, C(5)); 137.44 (d, C(6)); 151.78 (s, C(2)); 164.56 (s, C(4)). FAB-MS (pos., 3-NBA/NaI): 281 (100, $[M + Na]$ ⁺), 259.0937 $(37, [M + H]^+)$. Anal. calc. for C₁₀H₁₄N₂O₅: C 46.51, H 5.46, N 10.85; found: C 46.54, H 5.27, N 10.94.

 $1-(\alpha$ -L-Arabinopyranosyl)-N⁴-benzoyl-cytosine (3c). To a soln. of 36.0 g (54.6 mmol) of 2c in 500 ml of THF/MeOH/H₂O 5:4:1 ($v/v/v$), cooled in an ice bath, 100 ml of 1N NaOH soln. in MeOH/H₂O 1:1 (v/v) was

added during 5 min, and stirring at 4° was continued. After 45 min, TLC showed complete consumption of the starting material, and the mixture was neutralized ($pH 7$) by the addition of 6M aq. HCl. The pale yellow soln. was concentrated to a volume of 100 ml and kept at -20° overnight. The resulting precipitate was filtered, washed with EtOH, and dried under high vacuum $(0.03$ Torr/50 $^{\circ}$, 12 h) to yield 11.9 g of 3c. Concentration of the mother liquor led to another crop of 3.54 g for an overall yield of 15.46 g (82 %) of **3c**. Colorless powder. TLC $(CHCl₃/MeOH 6:1): R_f 0.2. ¹H-NMR (600 MHz, (D₆)DMSO): 3.53 (m, H-C(3')); 3.66 (d, J=12.1, H-C(5'));$ $3.72 \ (m, H-C(4'))$; $3.76 \ (m, H-C(2'))$; $3.80 \ (d, J=12.1, H-C(5'))$; $4.77 \ (d, J=5.3, HO-C(4'))$; $4.98 \ (d, J=5.7, A)$ $HO-C(3'))$; 5.24 (d, $J = 5.4$, $HO-C(2'))$; 5.47 (d, $J = 9.0$, $H-C(1'))$; 7.39 (d, $J = 6.5$, $H-C(5))$; 7.53 (m, 2 arom. H); 7.62 (*m*, 1 arom. H); 8.02 (*m*, 2 arom. H); 8.11 (*d*, *J* = 6.5, H – C(6)); 11.32 (*s*, H – N(4)). ¹³C-NMR $(150 \text{ MHz}, (\text{D}_6) \text{ DMSO})$: 69.37 $(t, \text{C}(4'))$; 69.85 $(d, \text{C}(2'))$; 70.47 $(d, \text{C}(5'))$; 74.23 $(d, \text{C}(3'))$; 85.05 $(d, \text{C}(1'))$; 97.53 (d, C(5)); 129.30, 133.60, 133.92 (arom. C); 147.05 (d, C(6)); 155.89 (s, C(2)); 163.90 (s, C(4)); 168.17 (s, CO of Bz). ESI-MS (pos., MeOH): 281 (100, $[M + Na]^+$), 238 (27, $[Cyt^{Bz} + Na]^+$), 216 (10, $[Cyt^{Bz} + H]^+$).

9-(α -L-Arabinopyranosyl)-N²-isobutyryl-guanine (3d). To a soln. of 25.0 g (37.5 mmol) of 2d in 400 ml of a mixture of THF/MeOH/H₂O 5:4:1 ($v/v/v$), cooled in an ice bath, 80 ml of 1N NaOH (aq.) was added over 15 min. After complete addition, stirring was continued for further 15 min, followed by neutralization with 37% aq. HCl soln. The clear soln. was evaporated to a volume of 200 ml, adsorbed on 50 g of silica gel, and purified by CC (silica gel; CH₂Cl₂/MeOH 3 :1 to 2 :1). The product fractions were combined and evaporated to yield 10.6 g (80%) of **3d**. White solid. TLC (CHCl₃/MeOH 3 : 1): R_f 0.33. ¹H-NMR (600 MHz, (D₆)DMSO): 1.10 (*d*, $J = 6.8, \text{MeCHCO}$); 1.12 (s, Me_2 CH); 2.77 (m, Me₂CHCO); 3.53 (m, H – C(3')); 3.64 (d, J = 12.1, H – C(5')); 3.77 $(m, H-C(4'))$; 3.86 $(dd, J = 1.7, 12.1, H-C(5'))$; 4.09 $(m, H-C(2'))$; 4.88 $(d, J = 5.1, HO-C(4'))$; 5.09 $(d, J = 1.7, 12.1, H-C(5'))$; 4.09 $(m, H-C(2'))$; 4.88 $(d, J = 5.1, HO-C(4'))$; 5.09 $(d, J = 1.7, 12.1, H-C(5'))$; 4.09 $(m, H-C(2'))$; 4.88 $(d, J = 5.1, HO-C(4'))$; 5.09 5.4, HO – C(3')); 5.14 (d, J = 9.3, H – C(1')); 5.39 (d, J = 5.4, HO – C(2')); 8.12 (s, H – C(8)); 11.76, 12.14 (2 br. s, $H-N(2)$, $H-C(1)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.67 (*q, Me*₂CH); 19.71 (*q, Me*₂CH); 35.55 (*d*, $Me₂CHCO$); 69.24 (d, C(4')); 69.74 (d, C(2')); 70.36 (t, C(5')); 74.35 (d, C(3')); 84.06 (d, C(1')); 120.53 (s, C(5)); 138.56 (d, C(8)); 149.01 (s, C(4)); 149.94 (d, C(2)); 155.70 (s, C(6)); 181.01 (s, Me₂CHCO). ESI-MS (pos., MeOH): 729 (100, $[2 M + Na]^+$), 376 (24, $[M + Na]^+$). ESI-MS (neg., MeOH): 388 (6, $[M + Cl]^-$), 352 (100, $[M-H]$ ⁻). Anal. calc. for C₁₄H₁₈N₅O₈: C 47.59, H 5.42, N 19.82; found: C 47.25, H 5.25, 19.66.

 $N⁶$ -Benzoyl-9-(3',4'-O-isopropylidene-a-L-arabinopyranosyl)adenine (4a). A soln. of 25.0 g (67 mmol) of 3a and 400 mg (2.1 mmol) of TsOH \cdot H₂O, in 200 ml of dry DMF, was cooled to 4° in an ice bath, and 20 ml (209 mmol) of 2-methoxyprop-1-ene was added over 30 min. Stirring at 4° was continued for another 30 min before the mixture was neutralized by the addition of 25% aq. NH₃ soln. The solvent was removed in vacuo and 50 ml of AcOEt was added to the residue. A fine precipitate formed was removed by filtration over silica gel, followed by washing the silica gel with CHCl₃/MeOH 10 : 1 (v/v). The filtrate was combined with the washings and evaporated *in vacuo*. The resulting foam was dissolved in 80 ml of hot AcOEt, from which crystals were formed overnight at r.t. The crystals were filtered and dried in high vacuum (0.03 Torr/r.t., 12 h) to yield 24.0 g (87%) of **4a**. Colorless crystals. TLC (CHCl₃/MeOH 9:1): R_f 0.59. ¹H-NMR (600 MHz, (D₆)DMSO): 1.35, 1.57 (2s, Me₂C); 4.09 (dd, J = 2.5, 13.4, $H-C(5')$; 4.22 (dd, J = 6.3, 6.4, H – C(3')); 4.25 (d, J = 13.4, H – C(5')); 4.33 (m, H – C(4'), H – C(2')); 5.52 (d, J = 9.6, H – C(1')); 5.70 $(d, J = 6.2, HO - C(2'))$; 7.57 – 8.05 $(m, 5 \text{ arom. H})$; 8.80 $(s, H - C(8))$; 8.93 $(s, H - C(2))$; 11.22 $(H-N(6))$. ¹³C-NMR (150 MHz, (D₆)DMSO): 27.07 (q, Me₂C); 28.87 (q, Me₂C); 66.13 (t, C(5')); 70.76 (d, C(2')); 74.11 (d, C(4')); 79.87 (d, C(3')); 83.19 (d, C(1')); 109.72 (s, Me₂C); 126.34 (s, C(5)); 129.32, 133.30, 134.22 (arom C); 144.11 (d, C(8)); 151.12 (s, C(4)); 152.57 (s, C(2)); 153.50 (s, C(6)); 166.48 (s, CO of Bz). ESI-MS (pos., MeOH): 845 (55, $[2 M + Na]^+$), 434 (21, $[M + Na]^+$), 240 (100, $[A^B z + H]^+$). ESI-MS (neg., MeOH): 410 (100, $[M - H]^-$). Anal. calc. for $C_{20}H_{21}N_5O_5$: C 58.39, H 5.14, N 17.02; found: C 58.25, H 5.18, N 17.04.

 $1-(3',4'-O-Isopropylidene-\alpha$ -L-arabinopyranosyl)thymine (4b). To a suspension of 23.8 g (92 mmol) of 3b in 200 ml of dry DMF, cooled in an ice bath, was added 27 ml of 4 HCl in dioxane followed by 15 ml (157 mmol) of 2-methoxyprop-1-ene over a 10 min. Stirring at 4° was continued for 1.5 h before the mixture was carefully neutralized with 25% aq. NH₃ soln. The mixture was filtered through silica gel, and the silica gel was washed with CHCl₃/MeOH 9:1 (v/v). The washings were combined with the filtrate, evaporated and 200 ml of $Et₂O$ was added to the residue. The resulting white precipitate was filtered and dried in high vacuum (0.03 Torr/ r.t., 12 h) to yield 26.2 g (95 %) of 4b. White solid. TLC (toluene/AcOEt/MeOH 5:4:1): R_f 0.46. ¹H-NMR $(600 \text{ MHz}, (D_6) \text{ DMSO})$: 1.30 $(s, Me_2\text{C})$; 1.51 $(s, Me_2\text{C})$; 1.81 $(s, \text{Me}-\text{C}(5))$; 3.69 $(m, \text{H}-\text{C}(2'))$; 3.91 $(dd, J=2.6$, $13.5, H-C(5')$; 4.11 (dd, $J = 6.0, 6.3, H-C(3')$); 4.14 (d, $J = 13.5, H-C(5')$); 4.20 (m, $H-C(4')$); 5.25 (d, $J = 9.7$, $H-C(1')$); 5.53 (d, J = 5.8, HO - C(2')); 7.50 (s, H - C(6)); 11.32 (br. s. H - N(3)). ¹³C-NMR (150 MHz, (D_6) DMSO): 12.75 $(q, Me-(5))$; 26.98 (q, Me_2C) ; 28.85 (q, Me_2C) ; 66.17 $(t, C(5'))$; 70.32 $(d, C(2'))$; 74.09 $(d,$ $C(4')$; 82.61 (d, $C(1')$); (d, $C(1')$); 109.50 (d, Me₂CH); 110.44 (s, $C(5)$); 137.30 (d, $C(6)$); 151.85 (s, C(2)); 164.25 $(s, C(4))$. ESI-MS (pos., MeOH): 321 (100, $[M + Na]^+$). ESI-MS (neg., MeOH): 297 (100, $[M - H]^-)$. Anal. calc. for $C_{13}H_{18}N_2O_6$: C 52.34, H 6.08, N 9.39; found: C 51.93, H 6.18, N 9.17.

 N^4 -Benzoyl-1-{3',4'-O-[(4-methoxyphenyl)methylidene]- α -L-arabinopyranosyl]cytosine (4c). A soln. of 11.41 g (33 mmol) of 3c, 630 mg (3.3 mmol) of TsOH \cdot H₂O, and 11.3 ml (66 mmol) of 4-anisaldehyde dimethyl acetal in 500 ml of dry DMF was stirred at r.t. overnight. The mixture was neutralized with conc. 25% aq. NH₃ soln. and evaporated (8 Torr 45°). The residue was co-evaporated twice with toluene, and 200 ml of MeOH was added, which caused spontaneous precipitation of a white solid. The solid was filtered, washed with MeOH, and dried under high vacuum (0.03 Torr/r.t., 12 h) to yield 13.6 g (89%) of $4c$. Mixture of diastereoisomers. TLC $(CHCl₃/MeOH 9:1): R_f 0.56, 0.64. ¹H-NMR (600 MHz, (D₆)DMSO): 3.79, 3.80 (2s, MeO); 3.83 (m, H-C(2²) of$ isomer a); 3.94 (dd, $J = 2.1, 13.5, H - C(5')$ of isomer a); 4.01 (m, $H - C(2')$ of isomer b); 4.08 (dd, $J = 2.0, 12.5$, $H-C(5')$ of isomer b); 4.32 $(m, H-C(4'))$; 4.36 $(d, J=12.5, H-C(5'))$ of isomer b); 4.54 $(dd, J=5.7, 6.9,$ $H-C(3')$; 5.59 (2d, $J=9.3$, $H-C(1')$); 5.78, 5.81 (2d, $J=6.0$, $HO-C(2')$); 5.85, 6.19 (2s, ArCH); 6.99 (m, 2 arom. H); 7.39, 7.51 (2m, H-C(5), 4 arom. H); 7.63 (m, 1 arom. H); 8.01 (m, 2 arom. H); 8.15, 8.25 (2 br. s, $H-C(6)$); 11.29 (H–N(4)). ¹³C-NMR (150 MHz, (D₆)DMSO): 56.01 (MeO); 66.32 66.75 (C(5')); 68.11, 71.92 $(C(2')$ isomer a and b); 74.32 $(C(3'))$; 76.37, 79.35 $(C(4'))$; 81.16 $(C(3'))$; 83.76, 83.91 $(C(1'))$; 97.60 $(C(5))$; 103.01, 104.29 (ArCH); 114.49, 114.54, 127.57, 128.73, 129.30, 129.54, 130.00, 131.93, 133.64, 133.91 (arom. C); 147.10 (C(6)); 155.81 (C(2)); 160.70, 166.93 (arom. C); 164.02 (C(4)); 168.23 (CO of Bz). ESI-MS (pos., MeOH): 488 $(100, [M + Na]^+), 238 (60, [Cyt^{Bz} + Na]^+), 216 (11, [Cyt^{Bz} + H]^+).$ ESI-MS (neg., MeOH): 464 (100, $[M - H]^-)$. Anal. calc. for C₂₄H₂₃N₃O₇: C 61.93, H 4.98, N 9.03; found: C 61.70, H 5.05, N 8.93.

N²-Isobutyryl-9-(3',4'-O-isopropylidene-a-L-arabinopyranosyl)guanine (4d). To a soln. of 9.3 g (26.2 mmol) of 3d and 262 mg (1.4 mmol) of TsOH \cdot H₂O in 260 ml of dry DMF, cooled in an ice bath, 4.5 ml (47 mmol) of 2-methoxyprop-1-ene was added, during 2 h, through a syringe. After complete addition, the soln. was allowed to cool to r.t. and stirred overnight. The mixture was neutralized with 25% aq. NH₃ soln., 20 g of silica gel was added, and the mixture was evaporated to dryness. The residue was purified by CC (silica gel; $CH_2Cl_2/MeOH$ 10 : 1 (v/v)). The product fractions were collected, combined, and evaporated, and the colorless foam was dried in high vacuum (0.03 Torr/r.t., 12 h) to yield 9.06 g (87%) of 4d. TLC (CHCl₃/MeOH 6:1): R_f 0.37. ¹H-NMR (600 MHz, (D₆)DMSO): 1.12 (d, J = 6.9, 2 Me₂CH); 1.33 (s, Me₂C); 1.55 (s, Me₂C); 3.92 (dd, J = 2.6, 13.3, H-C(5')); 4.08 (m, H-C(2')); 4.15 (t, J=7.0, H-C(3')); 4.23 (d, J=13.3, H-C(5')); 4.30 (m, $H-C(4')$; 5.17 (d, J = 9.5, $H-C(1')$); 5.68 (d, J = 6.0, HO – C(2')); 8.21 (s, H – C(8)); 11.65, 12.11 (2s, H – N(1), $H-N(2)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.69 (Me₂CH); 26.99 (Me₂C); 28.76 (Me₂C); 35.96 (Me₂CH-CO); 66.17 (C(5')); 71.07 (C(2')); 73.93 (C(4')); 79.78 (C(3')); 82.87 (C(1')); 109.77 (Me₂CH); 120.78 (C(5)); 138.81 (C(8)); 148.96 (C(4)); 149.89 (C(2)); 155.68 (C(6)); 180.99 (Me₂CHCO). ESI-MS (pos., MeOH): 416 $(100, [M + Na]^+)$, 394 $(38, [M + H]^+)$, 222 $(68, [G^{ibu} + H]^+)$. ESI-MS (neg., MeOH): 392 $(100, [M - H]^-)$. Anal. calc. for $C_{17}H_{23}N_5O_6$: C 51.90, H 5.89, N 17.80; found: C 51.60, H 6.22, N 17.59.

Nº-Benzoyl-9-(2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl]-3',4'-O-isopropylidene-a-∟-arabinopyranosyl)*adenine* (5a). To a soln. of 9.0 g (21.9 mmol) of 4a in 80 ml of dry pyridine, cooled to 4° in an ice bath, a soln. of 6.2 g (24.1 mmol) of (9H-fluoren-9-yl)methyl chloroformate in 40 ml of dry CH₂Cl₂ was added, over a period of 1.5 h via a dropping funnel. The pale yellow soln. was stirred at r.t. overnight, and the reaction was quenched by the addition of 10 ml of MeOH. The solvent was evaporated in vacuo, followed by co-evaporation with toluene. The residue was dissolved in 250 ml of CH₂Cl₂, and washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln. The org. layer was dried (Na_2SO_4) , evaporated, and the crude product was purified by CC (silica gel; toluene/ AcOEt 5 : 1 to 1 : 1). The product fractions were combined, evaporated, and dried under high vacuum (0.03 Torr/ r.t., 12 h) to afford 10.6 g (76%) of **5a**. Colorless foam. TLC (toluene/AcOEt/MeOH 5:4:1): R_f 0.55. ¹H-NMR $(600 \text{ MHz}, (D_6) \text{DMSO})$: 1.35 (s, Me₂CH); 1.58 (s, Me₂CH); 4.04 (t, J = 6.5, CH of Fmoc); 4.18 (dd, J = 2.3, 13.5, $H-C(5')$; 4.25 (dd, $J = 6.5, 10.6, 1$ H, CH₂ of Fmoc); 4.30 (dd, $J = 6.5, 10.6, 1$ H, CH₂ of Fmoc); 4.35 (d, $J =$ $13.5, H-C(5'))$; $4.40 (m, H-C(4'))$; $4.55 (dd, J = 5.6, 7.4, H-C(3'))$; $5.44 (dd, J = 7.4, 9.7, H-C(2'))$; $5.96 (d, J = 7.4, 9.7)$ 9.7, H $-C(1')$); 7.21 (*m*, 3 arom. H); 7.35 (*m*, 3 arom. H); 7.55 – 8.01 (*m*, 7 arom. H); 8.67 (*s*, H $-C(8)$); 8.74 (*s*, $H-C(2)$); 11.25 (s, $H-N(6)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 27.05, 28.58 (Me₂CH); 46.71 (CH of Fmoc); 66.25 (C(5')); 69.94 (CH₂ of Fmoc); 74.30 (C(4')); 76.66 (C(3')); 76.96 (C(2')); 80.23 (C(1')); 110.55 (Me₂CH); 120.93, 121.00, 125.38 (arom. C); 125.77 (C(5)); 127.89, 127.94, 128.44, 128.56, 129.30, 129.36, 133.34, 134.14, 141.40, 141.53 (arom. C); 143.46 (C(8)); 143.73, 143.76 (arom. C); 151.38 (C(4)); 152.83 (C(2)); 153.03 (C(6)); 154.28 (CO of Fmoc); 166.41 (CO of Bz). ESI-MS (pos., MeOH): 634 (87, $[M + H]$ ⁺), 418 (19, $[M + H - CO_2 (9-(\text{hydroxymethyl})-9H\text{-fluorene})$ $|\cdot|$, 240 (48, $[A^{Bz}+H]\cdot$); 179 (100, [9-methylidene-9H-fluorene + H] $|\cdot|$). ESI-MS (neg., MeOH): 632 (9, $[M - H]^-$); 410 (100, $[M - H -$ Fmoc]⁻); 238 (87, $[A^{Bz} - H]^-$). Anal. calc. for $C_{35}H_{31}N_5O_7$: C 66.34, H 4.93, N 11.05; found: C 66.20, H 5.28, N 10.75.

1-(2'-O-{[(9H-Fluoren-9-yl)methoxy]carbonyl}-3',4'-O-isopropylidene-a-L-arabinopyranosyl)thymine (5b). To an ice-cooled soln. of 24.0 g (80.5 mmol) of 4b in 200 ml of dry pyridine was added, over a period of 2 h, a soln. of 20.8 g (80.5 mmol) of (9H-fluoren-9-yl)methyl chloroformate in 80 ml of dry CH₂Cl₂. The yellow

soln. was stirred at 4° for further 30 min, and the reaction was stopped by the addition of 10 ml of MeOH. The soln. was evaporated to a volume of 50 ml, and 400 ml of CH_2Cl_2 was added. The org. layer was washed twice with sat. aq. NaHCO₃ soln., sat. aq. NaCl soln., dried (Na₂SO₄), and evaporated in vacuo, followed by the coevaporation with toluene. The resulting yellow oil was dissolved in 100 ml of toluene, from which the product crystallized. After storage at 4° overnight, the product was filtered, washed with Et2O, and dried in vacuum (0.03 Torr/r.t., 24 h) to yield 30.6 g (73%) of **5b**. Colorless crystals. M.p. 197–198°. TLC (toluene/AcOEt 1:1): R_f 0.46. ¹H-NMR (600 MHz, (D₆)DMSO): 1.31 (s, Me₂C); 1.52 (s, Me₂C); 1.76 (s, Me – C(5)); 4.04 (dd, J = 2.3, 13.6, $H-C(5')$; 4.23 (m, CH of Fmoc, $H-C(5')$); 4.29 (m, $H-C(4')$); 4.38 (dd, J = 6.7, 10.6, 1 H, CH₂ of Fmoc); 4.46 $(dd, J = 5.7, 7.4, H - C(3'))$; 4.50 $(dd, J = 6.6, 10.6, 1 H, CH_2$ of Fmoc); 4.91 $(dd, J = 7.4, 9.8, H - C(2'))$; 5.65 $(d, J = 7.4, 9.4, H - C(3'))$ $J = 9.8, H - C(1')$; 7.31 (*m*, 2 arom. C); 7.41 (*m*, 2 arom. H); 7.45 (*s*, H – C(6)); 7.50 – 7.88 (*m*, 4 arom. H); 11.42 (*s*, H-N(3)). ¹³C-NMR (150 MHz, (D₆)DMSO):12.77 (q, Me-C(5)); 26.93 (q, Me₂C); 28.56 (q, Me₂C); 46.88 (d, CH of Fmoc); 66.18 (d, C(5')); 70.13 (t, CH₂ of Fmoc); 74.26 (t, C(4')); 75.97 (d, C(2')); 76.76 (d, C(3')); 79.96 (d, C(1')); 110.33, 110.99 (C(5), Me₂CH); 121.04, 121.09, 125.50, 125.57, 128.00, 129.09, 128.66 (arom. C); 136.50 (d, $C(6)$; 141.50, 141.58, 143.78, 143.89 (arom. C); 151.37 (s, C(2)); 154.57 (s, CO of Fmoc); 164.28 (s, C(4)). FAB-MS (pos., 3-NBA/CsI): 653.0920 ($[M + Cs]^+$). Anal. calc. for $C_{28}H_{28}N_2O_8$: C 64.61, H 5.42, N 5.38; found: C 64.71, H 5.29, N 5.12.

N4-Benzoyl-1-(2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl}-3',4'-O-[(4-methoxyphenyl)methylidene]-a-Larabinopyranosyl)cytosine (5c). To a soln. of 12.0 g (25.8 mmol) of 4c in 120 ml of dry pyridine, cooled in an ice bath, was added a soln. of 7.34 g (28.4 mmol) of $(9H$ -fluoren-9-yl)methyl chloroformate in 50 ml of dry CH₂Cl₂, during 1.5 h. Stirring at 4° was continued for another 30 min before the reaction was quenched by the addition of 20 ml of MeOH. The mixture was evaporated under reduced pressure, and the residue was taken up in CH_2Cl_2 . The CH₂Cl₂ soln. was washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and dried (Na₂SO₄). The org. phase was evaporated in vacuo, and the residue was co-evaporated with toluene to remove the pyridine. The resulting foam was purified by CC (silica gel; toluene/AcOEt 5:1 to 1:2 (v/v)). The product fractions were collected, combined, and evaporated to colorless foam, which was dried under high vacuum (0.03 Torr/r.t., 12 h) to yield 15.8 g (82%) of 5c. TLC (toluene/AcOEt 1:2): R_f 0.34, 0.44. ¹H-NMR (600 MHz, (D₆)DMSO): 3.77, 3.78 (2s, MeO); $4.08(d, J = 11.7, H - C(5'))$; $4.18(t, J = 6.8, CH$ of Fmoc); $4.22(m, H - C(5'))$, CH of Fmoc); 4.34 $(m, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc}, \text{H}-\text{C}(4'), \text{H}-\text{C}(5'))$; 4.44 $(m, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc}, \text{H}-\text{C}(4'), \text{H}-\text{C}(5'))$; 4.67 $(dd, J=6.5$, 6.6, H – C(3')); 4.92 $(dd, J = 5.8, 6.7, H - C(3')$; 5.04 (br. t, $J = 8.5, H - C(2')$); 5.20 (br. t, $J = 8.8, H - C(2')$); 5.89 (s, ArCH) ; 5.99 (2d, $J = 8.5, 8.8, \text{H} - \text{C}(1'))$; 6.26 (s, ArCH); 6.98 (m, 2 arom. H); 7.36 (m, 2 arom. H); 7.39 (m, H-C(5), arom. H); 7.51 (m, 5 arom. H); 7.61 (m, 1 arom. H); 7.85 (m, 2 arom. H); 7.96 (m, 2 arom. H); 8.15, 8.33 (2m, H-C(6)); 11.36, 11.39 (2s, H-N(4)). 13C-NMR (150 MHz, (D6)DMSO): 46.87 (CH of Fmoc); 56.02 (MeO); 66.41, 66.72 (t, C(5')); 70.33 (t, CH₂ of Fmoc); 74.30 (d, C(4') of one isomer); 74.50 (d, C(2') of one isomer); 76.12 (d, C(3') of one isomer); 76.47 (d, C(4') of one isomer); 77.98 (d, C(3') of one isomer); 78.17 (d, $C(2')$ of one isomer); 81.04 (d, $C(1')$); 97.84, 98.00 ($C(5)$); 103.49, 104.84 (d, ArCH); 114.59, 114.58, 120.99, 121.05, 125.55, 125.57, 125.79, 127.79, 127.98, 128.13, 128.56, 128.83, 129.29, 129.31, 129.57, 131.35, 133.69, 133.81, 141.45, 141.48, 141.54, 143.82, 143.85, 143.87 (arom. C); 146.69, 146.93 (C(6)); 154.58, 154.64 (CO of Fmoc); 155.29, 155.37 (C(2)); 160.81, 161.07 (arom. C); 164.36 (C(4)); 168.27 (CO of Bz). FAB-MS(pos., NBA/NaI): 710 (100 $[M + Na]^+$), 688 (45, $[M + H]^+$), 216 (17, $[Cy^{Bz} + H]^+$), 179 (97, [9-methylidene-9H-fluorene + H]⁺).

9-(2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl]-3',4'-O-isopropylidene-a-L-arabinopyranosyl)-N²-isobutyrylguanine (5d). To a soln. of 5.4 g (13.7 mmol) of 4d in 50 ml of dry pyridine and cooled to 4° in an ice bath was slowly added, a soln. of 3.6 g (13.9 mmol) of (9H-fluoren-9-yl)methyl chloroformate dissolved in 20 ml of dry CH_2Cl_2 , through a dropping funnel during 2 h. Stirring at 4° was continued for further 30 min, when TLC showed complete consumption of the starting material. The reaction was stopped by the addition of 5 ml of MeOH, and the solvents were removed under vacuum. The residue was dissolved in 300 ml of CHCl₃, washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and dried (Na₂SO₄). The solvent was evaporated to yield pale yellow foam, which was purified by CC (silica gel; 17×5.5 cm, toluene/AcOEt/MeOH 2:1:0 to 4:5:1). The product fractions were collected, combined, and evaporated. During the evaporation, a white solid precipitated, and the mixture was kept at 4° overnight, and the product was filtered and dried in vacuum (0.03 Torr/r.t., 12 h) to yield 7.99 g (88%) of 5d. TLC (toluene/AcOEt/MeOH 5:4:1): R_f 0.37. ¹H-NMR (600 MHz, (D₆)DMSO): 1.13 (d, J = 6.9, Me₂CH); 1.17 (d, J = 6.9, Me₂CH); 1.34 (s, Me₂C); 1.56 (s, Me₂C); 2.81 (m, Me₂CHCO); 4.00 (t, $J = 6.2$, CH of Fmoc); 4.03 (dd, $J = 2.4$, 14.0, H-C(5')); 4.27 (dd, $J = 6.2$, 10.7, 1 H, CH₂ of Fmoc); 4.36 (d, $J = 6.2$, 10.7, 1 14.0, $H-C(5')$; 4.38 $(m, H-C(4'))$; 4.47 $(m, 1 H, CH_2$ of Fmoc, $H-C(3'))$; 5.17 $(dd, J=7.5, 9.7, H-C(2'))$; 9.70 $(d, J=9.7, H-C(1'))$; 7.13 $(m, 1 \text{ atom. H})$; 7.24 $(m, 2 \text{ atom. H})$; 7.30 – 7.84 $(m, 5 \text{ atom. H})$; 8.17 $(s, H-C(8))$; 11.67, 12.07 (2s, H $-N(1)$, H $-N(2)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.69 (*Me*₂CH); 19.74 (*Me*₂CH); 27.01 (Me_2C) ; 28.53 (Me_2C) ; 35.61 (Me₂CHCO); 46.75 (CH of Fmoc); 66.38 (C(5')); 69.88 (CH₂ of Fmoc); 74.15

 $(C(4'))$; 76.34 $(C(3'))$; 77.29 $(C(2'))$; 79.83 $(C(1'))$; 110.61 (Me₂CH); 120.39 $(C(5))$; 120.87, 120.93, 121.03, 125.02, 125.32, 127.52, 127.93, 128.31, 128.58 (arom. C); 138.30 (C(8)); 141.47, 141.56, 143.60, 143.90 (arom. C); 149.18 $(C(4))$; 149.37 $(C(2))$; 154.27 $(CO \text{ of Fmoc})$; 155.47 $(C(6))$; 181.09 (Me_2CHCO) . ESI-MS (pos., MeOH): 654 $(22, [M + K]^+), 638 (100, [M + Na]^+), 616 (100, [M + H]^+), 179 (95, [9-methylidene-9H-fluorene + H]^+).$ ESI-MS (neg., MeOH): 614 (3, $[M - H]$ ⁻), 392 (100, $[M - G^{ibu} - H]$ ⁻). Anal. calc. for $C_{32}H_{33}N_5O_6$ (615.64) C 62.53, H 5.46, N 11.38; found: C 62.53, H 5.28, N 11.29.

N⁶-Benzoyl-(2'-O-{[9H-fluoren-9-yl)methoxy]carbonyl]-a-L-arabinopyranosyl)adenine (6a). A soln. of 10 g (15.8 mmol) of $\textbf{5a}$ in 150 ml of 80% aq. AcOH soln. was stirred at 40°. After 48 h, the solvent was removed in vacuo, followed by co-evaporation with toluene. The residue was purified by CC (silica gel; CH₂Cl₂/ MeOH 100:2 to 100:5). The product fractions were combined, evaporated, and dried under high vacuum (0.03 Torr/r.t., 12 h) to yield 6.77 g (70%) of 6a. TLC (toluene/AcOEt/MeOH 5:4:1): R_f 0.39. ¹H-NMR $(600 \text{ MHz}, (D_6) \text{ DMSO})$: 3.85 $(m, H - C(4'))$; 3.91 (br. s, H-C(5)); 3.97 $(m, H - C(3'))$; 4.00 $(t, J = 6.4, \text{ CH of})$ Fmoc); 4.20 (*m*, CH₂ of Fmoc); 5.15 (*d*, $J = 6.0$, HO $-C(4')$); 5.44 (*m*, HO $-C(3')$, H $-C(2'))$; 5.87 (*d*, $J = 9.3$, $H-C(1')$); 7.19 (*m*, 2 arom. H); 7.28–8.02 (*d*, 11 arom. H); 8.62 (*s*, $H-C(8)$); 8.71 (*s*, $H-C(2)$); 11.24 (*s*, $H-N(6)$). ¹³C-NMR (150 MHz, (D₆)DMSO): 46.72 (CH of Fmoc); 69.44 (C(4')); 69.66 (CH₂ of Fmoc); 70.46 $(C(5'))$; 71.48 $(C(3'))$; 76.55 $(C(2'))$; 81.24 $(C(1'))$; 120.98, 120.89 (arom. C); 125.45, 125.50, 125.65 (arom C., C(5)); 127.90, 128.41, 128.54, 129.31, 129.34, 141.37, 141.53 (arom. C); 143.33 (C(8)); 143.73, 143.90 (arom. C); 151.28 (C(4)); 152.80 (C(2)); 153.04 (C(6)); 154.67 (CO of Fmoc); 166.38 (CO of Bz). ESI-MS (pos., MeOH): 632 (6, $[M + K]^+$), 616 (11, $[M + Na]^+$), 594 (73, $[M + H]^+$), 240 (100, $[A^{Bz} + H]^+$), 179 (89, [9-methylidene-9Hfluorene + H]⁺). ESI-MS (neg., MeOH): 628 (33, [M+Cl]⁻), 592 (28, [M – H]⁻); 370 (100, [M – Fmoc]⁻), 238 $(28, [A^{Bz} - H]⁻)$. Anal. calc. for $C_{32}H_{27}N_5O_7$ (593.59): C 64.75, H 4.58, N 11.80; found: C 64.76, H 4.69, N 11.77.

 $1-(2'-O-f/(9H-Fluoren-9-vl)methoxy/carbonvl-α-t-arabinopyranosyl)thymine (6b)$. To a suspension of 28.0 g (53.8 mmol) of 5b in 450 ml of dry MeOH, 3 ml of 4M HCl in dioxane was added. The mixture was heated to 50° in an oil bath for 2 h. The obtained soln. was allowed to cool to r.t. and was evaporated *in vacuo* to a pale yellow syrup, which was dissolved in 1000 ml of CH₂Cl₂. The org. layer was washed with sat. aq. NaHCO₃ soln., followed by drying (Na_2SO_4) . Evaporation of the solvent to a volume of 100 ml resulted in spontaneous precipitation of the product, which was collected by filtration. After drying in vacuum (0.03 Torr/r.t., 12 h) 19.5 g (75%) of 6b was obtained. White solid. TLC (CHCl₃/MeOH 1:1): R_f 0.45. ¹H-NMR (600 MHz, (D₆)DMSO): 1.77 (s, Me – C(5)); 3.78 (m, H – C(4'), 2 H – C(5')); 3.86 (m, H – C(3')); 4.22 (dd, J = 6.8, 7.0, CH of Fmoc); 4.35 $(dd, J = 6.8, 10.5, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc})$; 4.44 $(dd, J = 7.0, 10.5, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc})$; 5.00 $(dd, J = 9.3, 9.4, \text{H} - \text{C}(2'))$; 5.06 $(d, J = 6.5, HO - C(4'))$; 5.35 $(d, J = 6.0, HO - C(3'))$; 5.60 $(d, J = 9.3, H - C(1'))$; 7.32 $(m, 2 \text{ arrow. H})$; 7.42 $(t, J = 6.5, HO - C(4'))$; 7.42 $(d, J = 6.5, HO - C(4'))$; 7.42 $(d, J = 6.5, HO - C(4'))$; 7.42 $(d, J = 6.5, HO - C(4'))$; 7.42 $(d, J = 6.5, HO - C(4'))$; 7.42 $(d, J$ 2 arom. H); 7.52 (s, H-C(6)); 7.56 (m, 2 arom. H); 7.88 (m, 2 arom. H); 11.42 (s, H-N(3)). ¹³C-NMR $(150 \text{ MHz}, (D_6) \text{ DMSO})$: 12.95 $(q, \text{Me}-\text{C}(5))$; 46.92 $(d, \text{ CH of Fmoc})$; 69.37 $(t, \text{C}(4'))$; 69.91 $(t, \text{C}(5'))$; 70.15 $(t, \text{C}(5'))$ CH2 of Fmoc); 71.47 (d, C(3)); 75.47 (d, C(2)); 81.03 (d, C(1)); 110.58 (s, C(5)); 120.99, 121.05, 125.66, 125.80, 128.02, 128.10, 128.64, 128.78 (arom. C); 137.00 (d, C(6)); 141.48, 141.58, 143.82, 143.82, 144.06 (arom. C); 151.41 (s, C(2)); 154.94 (s, CO of Fmoc); 164.37 (s, C(4)). FAB-MS (pos., 3-NBA/CsI): 613.0606 ($[M + Cs]^+$). Anal. calc. for $C_{25}H_{24}N_2O_8$: C 62.50, H 5.03, N 5.83; found: C 62.20, H 5.41, N 5.67.

 N^4 -Benzoyl-(2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl]- a -L-arabinopyranosyl)cytosine (6c). To a soln. of 5.7 g (8.2 mmol) of 5c in 100 ml of MeOH and 50 ml of THF, cooled in an ice bath, 8 ml (104 mmol) of TFA was added, the ice bath was removed, and stirring was continued at r.t. After 2 h, when TLC indicated complete consumption of the starting material, the mixture was neutralized carefully with a 2M soln. of NH₃ in MeOH under ice cooling. The mixture was evaporated, and the residue was taken up in 400 ml of CH₂Cl₂ and adsorbed on 25 g of silica gel. The resulting white solid was purified by CC (silica gel; CH₂Cl₂/MeOH 50:1 to 50:4 (v/v). The product fractions were collected, combined, and evaporated to dryness (high vacuum, 0.03 Torr/50 $^{\circ}$, 12 h) to afford 4.5 g (95%) **6c**. White solid. TLC (CHCl₃/MeOH 9 : 1): R_f 0.42. ¹H-NMR (600 MHz, (D₆)DMSO): 3.81 $(m, H-C(4'))$; 3.83 $(d, J=12.0, H-C(5'))$; 3.88 $(d, J=12.0, H-C(5'))$; 3.93 $(m, H-C(3'))$; 4.20 $(t, J=7.0, CH$ of Fmoc); 4.29 $(dd, J = 7.0, 10.3)$; 4.38 $(dd, J = 7.0, 10.3, CH_2$ of Fmoc); 5.04 $(t, J = 9.0, H - C(2'))$; 5.16 $(d, J = 6.0, H)$ $HO-C(3'))$; 5.44 $(d, J=6.1, HO-C(4'))$; 5.87 (br. $d, J=9.0, H-C(1'))$; 7.32 (m, 2 arom. H); 7.38 (m, 3 arom. H); 7.51 $(m, H-C(5^{\circ}), 2 \text{ arom. H})$; 7.58 – 7.96 $(m, 6 \text{ arom. H})$; 8.17 $(\text{br. } d, J = 6.8, H-C(6))$; 11.34 $(s, H-N(4))$. $13C-NMR$ (150 MHz, $(D_6)DMSO$): 46.89 (d, CH of Fmoc); 69.35 (d, C(4')); 70.04 (t, CH₂ of Fmoc); 70.48 (t, $C(5')$; 71.46 $(C(3'))$; 76.60 $(C(2'))$; 82.14 $(C(1'))$; 97.83 $(C(5))$; 120.97, 121.03, 125.72, 126.03, 128.01, 128.17, 128.56, 128.60, 129.30, 133.67, 133.83, 141.45, 141.53, 143.90, 150.60 (arom. C); 146.83 (C(6)); 154.98 (CO of Fmoc); 155.37 (C(2)); 164.25 (C(4)); 168.24 (CO of Bz). ESI-MS (pos., MeOH): 592 (100 $[M + Na]^+$), 570 (11, $[M+H]^+$), 216 (5, [Cyt^{Bz} + H]⁺). ESI-MS (neg., MeOH): 604 (26, [M + Cl]⁻), 568 (66, [M – H]⁻).

9-(2'-O-{[(9H-Fluoren-9-yl)methoxy]carbonyl}-a-L-arabinopyranosyl)-N²-isobutyrylguanine (6d). A soln. of 7.3 g (11.9 mmol) of 5d in 800 ml of 80% aq. AcOH was stirred at 55° in an oil bath. After 10 h, TLC indicated complete reaction, and the solvent was removed in vacuo. To remove last traces of AcOH, the residue was co-evaporated several times with toluene and MeOH, resulting in a colorless foam, which was dried in high vacuum $(0.03$ Torr/r.t., 12 h) to yield 6.81 g (quant.) of 6d. TLC (CHCl₃/MeOH 9:1): R_f 0.53. ¹H-NMR (600 MHz, (D₆)DMSO): 1.12 (d, J = 6.8, Me₂CH); 1.16 (d, J = 6.8, Me₂CH); 2.78 (m, Me₂CHCO); 3.73 (d, J = $12.1, H-C(5')$; $3.84 \ (m, H-C(3'))$; $3.90 \ (d, J = 12.1, H-C(5'))$; $3.97 \ (t, J = 6.1, CH \ of \ F \text{moc})$; $4.22 \ (dd, J = 6.6, H₂)$ 10.7, 1 H, CH₂ of Fmoc); 4.43 $(dd, J = 5.9, 10.7, 1$ H, CH₂ of Fmoc); 5.17 $(d, J = 5.9, HO-C(3'))$; 5.30 $(t, J = 9.3,$ $H-C(2')$); 5.39 (d, $J = 9.3$, $H-C(1')$); 5.44 (d, $J = 5.7$, $HO-C(2')$); 7.12 (t, 1 arom. H); 7.25 (m, 3 arom. H); 7.38 – 7.83 (*m*, 4 arom H); 8.12 (*s*, H – C(8)); 11.75, 12.06 (2*s*, H – N(1), H – N(2)). ¹³C-NMR (150 MHz, $(D₆)$ DMSO): 19.66 ($Me₂CH$); 19.76 ($Me₂CH$); 35.63 ($Me₂CHCO$); 46.78 (CH of Fmoc); 69.30 (C(4')); 69.57 $(CH₂ of Fmoc); 70.68 (C(5'))$; 71.52 $(C(3'))$; 76.51 $(C(2'))$; 81.07 $(C(1'))$; 120.18 $(C(5))$; 120.89, 121.03, 125.10, 125.39, 127.49, 127.90, 128.26, 128.56 (arom. C); 138.35 (C(8)); 141.43, 141.56, 143.69, 143.93 (arom. C), 149.22, 149.39 (C(2), C(4)); 154.60 (CO of Fmoc); 155.49 (C(6)); 181.07 (Me₂CHCO). ESI-MS (pos., MeOH): 614 (4, $[M+K]^+$), 598 (24, $[M+Na]^+$), 576 (49, $[M+H]^+$), 240 (33, $[G^{ibu}+H]^+$), 179 (100, $[(9\text{-methylidene-9}H^+]$ fluorene + H]⁺). ESI-MS (neg., MeOH): 574 (28, $[M - H]$ ⁻); 352 $[M - G^{ibu} - H]$ ⁻). Anal. calc. for $C_{29}H_{29}N_6O_6 \cdot H_2O$ (593.58): C 58.68, H 5.26, N 11.80; found: C 58.82, H 5.66, N 11.96.

N⁶ -Benzoyl-9-(3-O-benzoyl-2-O-{[(9H-fluoren-9-yl)methoxy]carbonyl}---arabinopyranosyl)adenine (7a). To a soln. of 6.5 g (11 mmol) of 6a in 100 ml of dry pyridine, cooled in an ice/salt bath, 1.42 ml (12.1 mmol) of BzCl was slowly added through a syringe during 2 h. Stirring at 0° was continued for another 30 min, when TLC showed complete reaction. The mixture was poured into 100 ml of ice-cold sat. aq. NaHCO₃ soln. and 200 ml of CH₂Cl₂. The org. layer was washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and dried (Na2SO4). The org. layer was evaporated in vacuo, followed by co-evaporation with toluene, and the residue was purified by CC (silica gel; CH₂Cl₂/MeOH 100 : 2 to 100 : 5). Product fractions were combined, evaporated, and the colorless foam was dried in high vacuum (0.03 Torr/r.t., 12 h) to yield 5.57 g (73%) of 7a. TLC (CHCl₃/ MeOH 9:1): R_f 0.56. ¹H-NMR (600 MHz, (D₆)DMSO): 3.92 (t, J = 6.5, CH of Fmoc); 4.00 (dd, J = 1.0, 12.1, $H-C(5')$; 4.15 (d, J = 12.1, H - C(5')); 4.20 (m, CH₂ of Fmoc); 4.26 (m, H - C(4')); 5.55 (dd, J = 3.3, 9.6, $H-C(3')$; 5.68 (d, J = 6.4, HO - C(4')); 6.04 (t, J = 9.6, H - C(2')); 6.24 (d, J = 9.3, H - C(1')); 7.00 - 8.04 (m, 18 arom. H); 8.75 (s, H – C(8)); 8.78 (s, H – C(2)); 11.27 (s, H – N(6)). ¹³C-NMR (150 MHz, (D₆)DMSO): 46.58 $(d, \text{CH of Fmoc})$; 66.80 $(t, \text{C}(4'))$; 70.06 $(d, \text{C}(5'))$; 70.19 $(t, \text{CH}_2$ of Fmoc); 73.33 $(d, \text{C}(2'))$; 75.27 $(d, \text{C}(3'))$; 81.02 (d, C(1)); 120.84, 120.91 (arom. C); 125.24, 125.30, 125.81 (arom. C, C(5)); 127.73, 127.78, 128.42, 129.33, 129.36, 129.53, 129.98, 130.23, 133.37, 134.09, 134.49, 141.31, 141.39 (arom. C); 143.43 (d, C(8)); 143.49, 143.63 (arom. C); 151.45 (s, C(4)); 152.98 (s, C(2)); 153.09 (s, C(6)); 154.41 (s, CO of Fmoc); 165.81, 166.42 (s, CO of Bz). ESI-MS (pos., MeOH): 720 (9, $[M + Na]$ ⁺), 698 (100, $[M + H]$ ⁺), 418 (10, $[M + H - CO_2 - (9-(hydroxymethyl)-9H$ fluorene)]⁺). ESI-MS (neg., MeOH): 696 (14, [*M* + Cl]⁻), 474 (41, [*M* – H – Fmoc]⁻), 352 (100, [*M* – Fmoc – Bz]⁻). Anal. cal. for C₃₉H₃₁N₅O₆ (697.70): C 67.14, H 4.48, N 10.04; found: C 66.80, H 4.75, N 9.85.

1-(3'-O-Benzoyl-2'-O-{[(9H-fluorenyl)methoxy]carbonyl]-a-L-arabinopyranosyl)thymine (7b). To a soln. of 17.0 g (35.4 mmol) of 6b in 250 ml of dry pyridine, cooled in an ice bath, 5 ml (42.5 mmol) of BzCl was slowly added, over 30 min, through a dropping funnel. Stirring at 4° was continued for another 1.5 h, and the mixture was poured into a mixture of 300 ml of ice-cold sat. aq. NaHCO₃ soln. and 300 ml of CH₂Cl₂. The org. layer was separated, washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and dried (Na₂SO₄). The solvent was evaporated in vacuo, and the residue was co-evaporated twice with toluene. A colorless foam was obtained, which was purified by CC (silica gel; toluene/AcOEt $1:0$ to $1:1$). The product fractions were collected, evaporated, and dried under high vacuum (0.03 Torr/r.t.) to yield 15.9 g (77%) of **7b**. Colorless foam. TLC (toluene/AcOEt/MeOH 5:4: 1): R_f 0.7. ¹H-NMR (600 MHz, (D₆)DMSO): 1.80 (s, Me-C(5)); 3.89 (dd, J= 1.23, 12.1, $H - C(5')$; 4.02 (d, J = 12.1, $H - C(5')$); 4.09 (t, J = 6.7, CH of Fmoc); 4.14 (d, J = 6.8, $H - C(4')$); 4.32 $(dd, J = 6.7, 10.6, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc})$; 4.39 $(dd, J = 6.9, 10.6, 1 \text{ H}, \text{CH}_2 \text{ of Fmoc})$; 5.44 $(m, H - C(2), H - C(3'))$; 5.57 (d, $J = 7.1$, HO $-C(4')$); 5.94 (d, $J = 8.0$, H $-C(1')$); 7.10 -7.93 (m, 14 H, arom. H, H $-(C(6))$; 11.48 (s, H-N(3)). ¹³C-NMR (150 MHz, (D₆)DMSO): 12.96 (q, Me-C(5)); 46.75 (d, CH of Fmoc); 66.75 (t, C(4')); 69.89 (d, C(5')); 70.21 (t, CH₂ of Fmoc); 72.60 (d, C(3')); 75.14 (d, C(2')); 80.64 (C(1')); 111.1 (d, C(5')); 120.93, 120.97, 125.37, 125.51, 127.85, 127.88, 128.52, 128.55, 129.41, 129.54, 129.98, 130.17, 134.47 (arom. C); 136.56 (d, C(6)); 141.40, 141.45, 143.58, 143.76 (arom. C); 151.33 (s, C(2)); 154.69 (s, CO of Fmoc); 164.33 (s, C(4)); 165.72 $(s, CO$ of Bz). ESI-MS (pos., MeOH): 623 (9, $[M + K]^+$), 607 (100, $[M + Na]^+$), 179 (54, [9-methylidene-9Hfluorene + H]⁺). ESI-MS (neg., MeOH): 619 (73, $[M + Cl]^-$), 361 (100, $[M - \text{Fmoc}]^-$). Anal. calc. for $C_{32}H_{28}N_2O_9$ (584.58): C 65.75, H 4.83, N 4.79; found: C 65.40, H 5.12, N 4.57.

N⁴-Benzoyl-(3'-O-benzoyl-2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl]-a-L-arabinopyranosyl)cytosine $(7c)$. BzCl (1.3 ml, 10.18 mmol) was slowly added to an ice-cold mixture of 5.8 g (11.2 mmol) of 6c in 100 ml of dry pyridine during 1.5 h. Stirring at 4° was continued for another 30 min, when TLC showed complete reaction.

The mixture was poured into 100 ml of ice-cold sat. aq. NaHCO3 soln. and 200 ml of CH₂Cl₂. The org. layer was separated, washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., dried (Na₂SO₄), and evaporated. Last traces of pyridine were removed by co-evaporation with toluene, and a pale yellow foam was obtained, which was purified by CC (silica gel; toluene/AcOEt 5 : 1 to 1 : 1). The product fractions were collected, combined, and evaporated. During the evaporation, a white solid precipitated, which was filtered, washed with Et₂O, and the mixture was kept at 4° overnight. Drying of the product under vacuum (0.03 Torr/r.t., 12 h) yielded 5.69 g (83%) of **7c**. TLC (toluene/AcOEt 1:2): R_f 0.43. ¹H-NMR (600 MHz, (D₆)DMSO): 3.96 (d, J = 11.31, H – C(5')); 4.08 $(m, H-C(4'), H-C(5'))$; 4.19 (br. $d, J = 6.2$, CH of Fmoc); 4.28 (dd, J = 7.0, 10.4, 1 H, CH₂ of Fmoc); 4.33 (dd, $J = 7.0, 10.4, 1$ H, CH₂ of Fmoc); 5.49 (*m*, H-C(3'), H-C(2')); 5.65 (*d*, $J = 6.6$, HO-C(4')); 6.18 (br. *d*, $H-C(1')$); 7.11 (*m*, 2 arom. H); 7.27 (*t*, 1 arom. H); 7.31 (*m*, 2 arom. H); 7.44 (*m*, $H-C(5)$, 1 arom. H); 7.50 (*m*, 4 arom. H); 7.64 $(m, 2 \text{ arom. H})$; 7.79 $(m, 2 \text{ arom. H})$; 7.94 $(m, 2 \text{ arom. H})$; 7.99 $(d, 2 \text{ arom. H})$; 8.27 $(d, J = 7.3, J)$ $H-C(6)$); 11.36 (s, H-N(4)). ¹³C-NMR (150 MHz, (D₆)DMSO): 46.71 (CH of Fmoc); 66.76 (C(4')); 70.24, 70.32 (C(2'), C(3)); 73.63, 75.06 (C(5'), CH₂ of Fmoc); 81.77 (C(1')); 98.17 (C(5)); 120.91, 120.95, 125.42, 125.65, 127.86, 127.95, 128.44, 128.52, 129.32, 129.56, 129.98, 130.20, 133.70, 133.84, 134.49, 141.37, 141.41, 143.63, 143.73 (arom. C); 146.50 (C(6)); 154.69 (CO of Fmoc); 155.69 (CO of Fmoc); 155.26 (C(2)); 164.38 (C(6)); 165.75 (CO of BzO); 168.27 (CO of NHBz). ESI-MS (pos., MeOH): 696 (100, $[M + Na]^+$), 674 (17, $[M + H]^+$). ESI-MS $(\text{neg. MeoH}): 672 (100, [M - H]^{-})$, 450 (25, $[M - H$ -Fmoc]⁻), 432 (21, $[M - 2OBz]^{-})$. Anal.: cal. for $C_{38}H_{31}N_3O_9$ (673.56); C 67.76, H 4.64, N 6.24; found: C 67.73, H 4.73, N 6.25.

9-(3'-O-Benzoyl-2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl}-a-L-arabinopyranosyl)-N²-isobutyrylguanine (**7d**). A soln. of 5.35 g (12.6 mmol) of 6d in 100 ml of dry pyridine and 50 ml of dry CH₂Cl₂ was cooled to -20° (i-PrOH / dry ice), and 1.1 ml (9.4 mmol) of BzCl was slowly added during 2 h. After complete addition, stirring was continued for another h, when TLC showed consumption of the starting material. The soln. was poured into 200 ml of ice-cold sat. aq. $NaHCO₃$ soln. and 200 ml of $CH₂Cl₂$. The org. phase was extracted with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., dried (Na₂SO₄), and evaporated. The crude product was purified by CC (silica gel; CH₂Cl₂/MeOH 95 : 5 (v/v)). The product fractions were collected, combined, and evaporated to yield the product as colorless foam, which was dried in high vacuum (0.03 Torr/r.t.) overnight to yield 5.2 g (82%) of **7d**. TLC (CHCl₃/MeOH 6:1): R_f 0.55. ¹H-NMR (600 MHz, (D₆)DMSO): 1.12, 1.14 (2*d, J* = 6.9, *Me*₂CH); 2.82 (m, Me_2CHCO) ; 3.92 (br. t, J = 6.2, CH of Fmoc); 4.00 (br. s, H – C(5')); 4.22 (m, 1 H, CH₂ of Fmoc), H – C(4')); 4.38 (dd, $J = 6.2, 10.7, 1$ H, CH₂ of Fmoc); 5.43 (dd, $J = 3.4, 9.8, H - C(3')$); 5.69 (br. d, $J = 4.8, HO - C(4')$); 5.74 $(d, J = 9.3, H - C(1))$; 5.80 (br. t, $J = 9.5, H - C(2))$; 7.04 – 7.94 (m, 13 arom. H); 8.25 (s, H – C(8)); 11.76, 12.13 (2s, HN-C(1), HN-C(2)). ¹³C-NMR (150 MHz, (D₆)DMSO): 19.53 (Me_2CH); 19.71 (Me_2CH); 35.56 (Me₂CHCO); 46.66 (CH of Fmoc); 66.71 (C(4')); 69.99 (CH₂ of Fmoc); 70.37 (C(5')); 73.43 (C(2')); 75.14 (C(3)); 80.78 (C(1)); 120.51, 120.84, 120.93 (C(5), arom. C); 125.10, 125.16, 127.56, 127.73, 128.34, 128.45, 129.39, 129.59, 129.92, 130.10, 130.20, 133.66, 134.56 (arom. C); 138.25 (C(8)); 141.35, 141.43, 143.50, 143.63 (arom. C); 149.32, 149.51 (C(2), C(4)); 154.39 (CO of Fmoc); 155.54 (C(6)); 165.76 (CO of Bz); 181.14 $(Me,CHCO)$. ESI-MS (pos., MeOH): 702 (100, $[M+Na]^+$), 680 (58, $[M+H]^+$); 222 (19, $[G^{ibu}+H]^+$), 179 (24, [9-methylidene-9H-fluorene + H]⁺). ESI-MS (neg., MeOH): 678 (9, [M – H]⁻), 456 (100, [M – G^{ibu} – H]⁻). Anal. calc. for $C_{36}H_{33}N_5O_9 \cdot H_2O$ (697.70): C 61.97, H 5.06, N 10.04; found: C 61.82, H 5.45, N 9.90.

2. Experiments Referring to Scheme 4. – N⁶-Benzoyl-9-(3'-O-benzoyl-4'-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2'-O-{[(9H-fluoren-9-yl)methoxy]carbonyl]-a-L-arabinopyranosyl)adenine (8a). A soln. of 1.0 g (1.43 mmol) of **7a** in 15 ml of dry CH₂Cl₂ was stirred over molecular sieves (4 Å) for 1 h at r.t. To this soln., 100 mg (1.43 mmol) of 1H-tetrazole and 682 ml (2.15 mmol, 1.5 equiv.) of (2-cyanoethoxy)bis(diisopropylamino)phosphine were added, and stirring at r.t. was continued for another h. The mixture was washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln. The org. layer was dried (Na_2SO_4) and evaporated in vacuo to afford colorless foam. This residue was purified rapidly (10-15 min.) by CC (silica gel; 7×2 cm, conditioned with toluene containing 1% of $E(N(i-Pr)_{2}$, toluene/AcOEt 1:1). The product fractions were combined, and the solvent was removed in vacuo and dried under high vacuum $(0.03$ Torr/r.t.) overnight to yield 1.02 g (80%) 8a as a mixture of diastereoisomers. TLC (toluene/AcOEt 1:2): R_f 0.52, 0.59. ¹H-NMR (600 MHz, CDCl₃): 1.15, 1.22 $(4d, J = 6.8, Me₂CH);$ 2.37 (m, CH₂CH₂CN); 2.63 (t, $J = 6.2$, CH₂CH₂CN); 3.58 (m, CH₂CH₂CN); 3.71 (m, $Me₂CH, CH₂CH; 3.78 (m, CH₂CH₂CN); 3.87 (m, CH₂CH₂CN); 3.88 (t, J = 7.5, CH of Fmoc); 3.93 (t, J = 7.5, CH)$ 7.5, CH of Fmoc); 4.05 (2d, $J = 12.7$, H $-C(5')$); 4.11 (dd, $J = 3.7$, 7.5, 1 H, CH₂ of Fmoc); 4.15 (*m*, 1 H, CH₂ of Fmoc); 4.25, 4.39 (2dd, $J = 1.8$, 12.7, H $-C(5)$); 4.63, 4.73 (2m, H $-C(4')$); 5.39, 5.46 (2dd, $J = 3.0$, 9.6, H $-C(3')$); 6.03 (m, H-C(2), H-C(1)); 7.01, 7.10, 7.22, 7.25, 7.30, 7.36, 7.41, 7.52, 7.63 (9m, 13 arom. H); 8.31, 8.36 (2s, H-C(8)); 8.83, 8.85 (2s, H-C(2)); 9.03, 9.05 (2 br. s, H-N(6)). 13C-NMR (150 MHz, CDCl3): 20.34, 20.77 (CH₂CH₂CN); 24.94, 25.07, 25.12 (Me₂CH); 43.72, 43.76, 43.80, 43.84 (Me₂CH); 46.50 (CH of Fmoc); 57.91, 58.81 (2d, $J(C,P) = 19.5$, CH₂CH₂CN); 69.24 (C(4')); 69.54, 69.60 (C(5')); 69.66 (C(4')); 70.95, 71.00 (CH₂ of Fmoc); 72.45, 72.66 (C(2)); 73.91, 74.40 (C(3)); 81.34, 81.50 (C(1)); 117.95, 120.24, 120.28, 123.04, 123.11, 125.11 (CN, C(5), arom. C); 127.40, 127.43, 128.14, 128.18, 128.28, 128.76, 129.22, 129.44, 130.12, 130.35, 133.19, 133.92, 134.02, 134.13 (arom. C); 141.27, 141.35, 141.41, 141.57 (C(8), arom. C); 142.96, 143.01, 143.17 (arom. C); 150.02 (C(4)); 152.53 (C(2)); 153.55 (C(6)); 154.43, 154.50 (s, CO of Fmoc); 164.96, 165.84, 165.93 (s, CO of Bz). ³¹P-NMR (242 MHz, CDCl₃): 151.40; 152.74. ESI-MS (pos., MeOH): 920 (100, $[M + Na]^+$), 898 (95, $[M + H]^+$). ESI-MS (neg., MeOH): 874 ($[M-H]$ ⁻).

1-(3-O-Benzoyl-4-O-[(2-cyanoet hoxy)(diisopropylamino)phospino]-2-O-{[(9H-fluoren-9-yl)methoxy] carbonyll-a-L-arabinopyranosyl)thymine (8b). To a soln. of 1.0 g (1.71 mmol) of 7b in 15 ml of dry CH_2Cl_2 , stirred over molecular sieves (4 Å) for 1 h at r.t., 119 mg (1.71 mmol) of 1H-tetrazole and 0.82 ml (2.57 mmol, 1.5 equiv.) of (2-cyanoethoxy)bis(diisopropylamino)phosphine were added. The mixture was stirred for 60 min at r.t., until TLC showed complete consumption of the starting material. The mixture was diluted with 20 ml of CH_2Cl_2 and washed twice with sat. aq. NaHCO₃ soln., followed by sat. aq. NaCl soln. The org. phase was dried (Na₂SO₄) and evaporated *in vacuo* to afford colorless foam, which was purified rapidly (10-15 min.) by CC (silica gel; 7×2 cm, conditioned with toluene containing 1% of EtN(i-Pr)₂, toluene/AcOEt 3:1). The product fractions were combined, evaporated, and dried under high vacuum (0.03 Torr/r.t., 12 h) to afford 1.12 g (84%) of 8b as a mixture of diastereoisomers. TLC (toluene/AcOEt 1:1): R_f 0.63, 0.72. ¹H-NMR (600 MHz, CDCl₃): 1.20 $(4d, J = 6.8, Me₂CH)$; 1.95, 1.96 $(2s, Me-C(5))$; 2.35 $(m, CH₂CH₂CN)$; 2.61 $(t, J = 6.2, CH₂CH₂CN)$; 3.55 $(m, CH, CH, CN); 3.67$ $(m, Me, CH, CH, CH, CN); 3.76$ $(m, CH, CH, CN); 3.82$ $(m, CH, CH, CH, CN); 3.92$ (2d, J = 11.6, 12.1, H – C(5')); 4.09 (2t, J = 7.8, J = 7.9, CH of Fmoc); 4.16 (d, J = 11.6, H – C(5')); 4.29 (m, H – C(5'), CH₂ of Fmoc); 4.55, 4.65 $(2m, H-C(4'))$; 5.31, 5.37 $(2dd, J = 3.1, 10.0, H-C(3'))$; 5.47, 5.54 $(2t, J = 9.6, H-C(2'))$; 5.93, 5.95 (2d, $J = 9.3$, H-C(1')); 7.10–8.00 (8m, 13 arom. H, H-C(6)); 8.45 (br. s, H-N(3)). ¹³C-NMR $(150 \text{ MHz}, \text{CDCl}_3)$: 12.91, 13.00 $(q, \text{Me}-\text{C}(5)$ of both isomers); 20.32, 20.75 $(t, \text{CH}_2\text{CH}_2\text{CN}$ of both isomers); 24.92, 25.13, 25.19 (q, Me₂CH of both isomers); 43.67, 43.70, 43.75, 43.78 (d, Me₂CH of both isomers); 46.65 (d, CH of Fmoc of both isomers); 58.00, 58.93 (t, $J(C,P) = 18.7, 20.0, CH_2CH_2CN$ of both isomers); 68.55 (t, C(4') of isomer a); 68.80, 69.17 (d, $C(5')$ of both isomers); 69.53 (t, $C(4')$ of isomer b); 71.09, 71.12 (t, CH_2 of Fmoc of both isomers); 71.80, 72.07 (d, C(2') of both isomers); 73.70, 74.29 (d, C(3') of both isomers); 81.17 (d, C(1') of both isomers); 112.27, 112.46 (s, CN of both isomers); 117.76 (d, C(5)); 120.33, 120.37, 125.34, 125.44, 127.48, 127.52, 128.22, 128.29, 128.73, 128.97, 129.28, 129.48, 130.07, 130.30, 133.95, 134.07 (arom. C); 135.39 (t, C(6)); 141.49, 141.53, 143.18, 143.27 (arom. C); 150.64, 150.72 (s, C(2)); 154.79, 154.81 (s, CO of Fmoc); 163.51 (s, C(4)); 165.79, 165.88 (s, CO of Bz). ³¹P-NMR (242 MHz, CDCl₃): 151.64, 153.03. ESI-MS (pos., MeOH): 807 $(100, [M+Na]^+)$. Anal. calc. for $C_{41}H_{45}N_4O_{10}P$: C 62.71, H 5.78, N 7.13; found: C 62.79, H 5.81, N 7.13.

N⁴ -Benzoyl-1-(3-O-benzoyl-4-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2-O-{[(9H-fluoren-9 yl)methoxy]ca rbonyl]-a-L-arabinopyranosyl)cytosine (8c). A suspension of 1.0 g (1.5 mmol) of 7c in 15 ml of dry CH₂Cl₂ was stirred over molecular sieves (4 Å) for 1 h at r.t., and 105 mg (1.5 mmol) of 1H-tetrazole and 715 µ (2.25 mmol, 1.5 eq.) of (2-cyanoethoxy)bis(diisopropylamino)phosphine were added, and stirring at r.t. was continued for another h, until TLC showed complete consumption of the starting material. The mixture was washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., dried (Na₂SO₄), and was evaporated to a colorless foam. The crude product was purified by CC (silica gel; 7×2 cm, toluene/AcOEt 2:1 (v/v)). To avoid acidcatalyzed decomposition of the phosphoramidite the silica gel was first washed with toluene containing 1% of EtN(i-Pr)₂, and CC was performed within 10 to 15 min. The product fractions were collected, combined, and evaporated to yield colorless foam, which was dried under high vacuum (0.03 Torr/r.t., 12 h) to afford 990 mg (84%) of **8c** as a mixture of diastereoisomers. TLC (toluene/AcOEt 1:1): R_f 0.41, 0.49. ¹H-NMR (600 MHz, $CDCl₃$): 1.16, 1.19, 1.24 (4d, Me₂CH); 2.37, 2.63 (t, J = 6.2, CH₂CH₂CN); 3.56 (m, CH₂CH₂CN); 3.69 (m, Me₂CH, CH_2CH_2CN); 3.77, 3.84 (*m*, CH_2CH_2CN); 3.97 (2*d*, *J* = 12.5, H – C(5')); 4.09 (2*t*, *J* = 7.7, *J* = 7.9, CH of Fmoc); $4.19, 4.29 \ (m, \text{CH}_2 \text{ of } \text{Fmoc}, \text{H} - \text{C}(5'))$; $4.58, 4.68 \ (2m, \text{H} - \text{C}(4'))$; $5.35, 5.41 \ (2dd, J = 3.2, 10.0, \text{H} - \text{C}(3'))$; 5.49 , 5.57 $(2t, J = 9.8, H - C(2'))$; 5.23 (br. $t, J = 9.8, H - C(1'))$; $7.08, 7.18, 7.27, 7.31, 7.37, 7.40, 7.43, 7.50, 7.62, 7.66, 7.87,$ 7.95, 8.00 (18m, 2 br. s, 13 arom. H, H-C(5), H-C(6)); 8.70 (br. s, H-N(4)). 13C-NMR (150 MHz, CDCl3): 20.38, 20.76 (CH₂CH₂CN); 23.00, 24.94, 25.08 (Me₂CH); 43.69, 43.77 (Me₂CH); 46.68 (CH of Fmoc); 58.00, 58.90 (2d, $J(C, P) = 18.2$, 20.0, CH₂CH₂CN); 68.55 (d, $J(C, P) = 16.3$, C(4')); 69.18, 69.45 (C(5')); 69.65 (d, $J(C,P) = 23.7, C(4')$; 71.13, 71.17 (CH₂ of Fmoc); 72.73, 73.08 (C(2')); 73.76, 74.30 (C(3')); 82.21 (C(1')); 120.18, 120.23, 120.26, 125.44, 125.47, 125.54, 125.57, 127.53, 128.11, 128.13, 128.74, 128.97, 129.32, 129.40, 129.49, 130.06, 130.30, 133.67, 133.93, 134.05, 141.37, 141.19, 143.17, 143.48, 143.51 (arom. C); 154.84, 154.88 (CO of Fmoc); 165.80, 165.89 (CO of BzO). ³¹P-NMR (240 MHz, CDCl₃): 151.73, 152.79. ESI-MS (pos., MeOH): 896 (80, $[M + Na]$ ⁺), 874 (100, $[M + H]$ ⁺). Anal. calc. for C₄₁H₄₅N₄O₁₀P (785.30): C 64.60, H 5.54, N 8.01; found: C 64.25, H 5.79, N 8.21.

9-(3-O-Benzoyl-4-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2-O-{[(9H-fluoren-9-yl)methoxy] carbonyl}- α -L-arabinopyranosyl)-N²-isobutyrylguanine (8d). To a soln. of 679 mg (1 mmol) of 7d in 20 ml of dry CH₂Cl₂ stirred over molecular sieves (4 Å) for 1 h at r.t, were added 77 mg (1.1 mmol) of 1H-tetrazole and 476 µ (1.5 mmol) of (2-cyanoethoxy)bis(diisopropylamino)phosphane, and stirring was continued for another h. The mixture was diluted with CH₂Cl₂, washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and evaporated. The colorless foam obtained was purified by CC (silica gel; 7×2 cm) with toluene/AcOEt 1:1 (v/v). The product fractions were collected, combined, and evaporated to yield 588 mg (67%) of 8d. Colorless foam. TLC (toluene/AcOEt 1:2): R_f 0.47. ¹H-NMR (600 MHz, CDCl₃): 1.13, 1.17, 1.23 (4*d, J* = 6.8, 2 *Me*₂CH); 2.36, 2.60 (t, J = 6.1, CH₂CH₂CN); 2.72 (m, Me₂CHCO); 3.59 (m, CH₂CH₂CN); 3.67 (m, Me₂CH, CH₂CH₂CN); 3.73, 3.80 (m, CH₂CH₂CN); 3.86 (d, J = 12.6, H – C(5')); 3.92 (t, J = 7.3, CH of Fmoc); 3.99 (t, J = 7.3, CH of Fmoc); 4.11 (dd, $J = 7.5$, 10.6, CH₂ of Fmoc); 4.11 (dd, $J = 7.5$, 10.6, CH₂ of Fmoc); 4.25, 4.29 (2dd, $J = 2.1$, 12.6, $H-C(5)$; 4.57, 4.70 (2 br. m, $H-C(4')$); 5.38, 5.41 (2dd, $J = 3.1, 9.9$, $H-C(3')$); 5.64, 5.66 (2d, $J = 9.6$, $H-C(1')$); 5.87, 6.20 (2t, J 9.6, H-C(2)); 7.06, 7.10, 7.22, 7.26, 7.30, 7.36, 7.49, 7.52, 7.64, 7.97 (10m, 13 arom. H); 7.92, 7.93 $(2s, H-C(8))$; 9.34 (br. s, H-N). ¹³C-NMR (150 MHz, CDCl₃): 19.31, 20.52 (*Me*₂CH); 20.78, 20.84 (CH_2CH_2CN) ; 24.90, 25.04, 25.09 (Me₂CH); 36.46, 36.69 (Me₂CHCO); 43.75, 43.82 (M₂CH); 46.60 (CH of Fmoc); 57.80 (d, J(C,P) = 19.5, CH₂CH₂CN); 68.84 (C(4')); 69.23 (C(5')); 69.65 (d, J(C,P) = 14.0, C(4'); 70.87 $(CH, of Fmoe)$; 72.46 $(C(2'))$; 73.81 $(C(3'))$; 81.11 $(C(1'))$; 118.20, 120.28, 120.31, 120.33 $(CN, C(5)$, arom. C); 127.39, 127.47, 128.18, 128.23, 128.77, 128.97, 129.16, 129.39, 130.06, 130.29, 134.07 (arom. C); 137.02 (C(8)); 141.40, 142.98, 142.98, 143.14 (arom. C); 148.57, 149.12 (C(4), C(2)); 154.29 (CO of Fmoc); 155.97 (C(6)); 166.02 (CO of Bz); 179.36 (Me₂CHCO). ³¹P-NMR (242 MHz, CDCl₃): 152.88. ESI-MS (pos., MeOH): 918 ([M+ K]⁺), 902 ($[M + Na]$ ⁺), 880 ($[M + H]$ ⁺). Anal. calc. for C₄₅H₅₀N₆O₁₀P (879.90): C 61.43, H 5.73, N 11.11; found: C 61.22, H 5.91, N 11.08.

General Method for the Preparation of Nucleoside-Derivatized LCAMA-Controlled-Pore-Glass Solid Supports (LCAMA-CPG) (10a – 10d) [15] [16]. A mixture of 0.2 – 0.3 g (0.29 – 0.5 mmol) of 7a – 7d, 57 – 100 mg $(0.57 - 1.0 \text{ mmol})$ of succinic anhydride, $3.5 - 30 \text{ mg } (0.03 - 0.25 \text{ mmol})$ of 4-DMAP, and $0.14 - 0.24 \text{ ml}$ of pyridine in 10 ml of CH₂Cl₂ was stirred at r.t. for $3 - 4$ days (monitored by TLC for consumption of starting material). The mixture was extracted with sat. aq. NaHCO₃ soln., followed by 5% aq. citric acid soln. The org. phase was dried (Na₂SO₄) and concentrated in vacuo to afford 67-95% of **9a-9d**, containing traces of starting material as impurity.

A suspension of 2 g of glyceryl-CPG (500 Å) and 3 g of N,N-carbonyldi $[1H$ -imidazole] in 50 ml CH₂Cl₂ were shaken for 24 h at r.t. The solvent was filtered, the CPG was washed with CH₂Cl₂ and dried. The activated CPG was suspended in CH_2Cl_2 , and 3 ml of 1,6-bis-methylamino hexane was added and the mixture was shaken for 24 h at r.t. The solvent was filtered, the support was washed carefully with DMF, MeOH, acetone, Et₂O, and dried. To 300 mg of the derivatized LCAMA-CPG (500 Å), 30 mg (ca. 0.05 mmol) of nucleosides 9a - 9d, 30 mg TOTU, and 0.01 ml N-methylmorpholine in 5 ml of MeCN were added, and the mixture was shaken for 1.5 h. The solvent was filtered, the support was washed with DMF, MeOH, acetone, Et₂O, and dried. To this dried CPG was added 2 ml of 1-methyl-1H-imidazole in pyridine/THF (capping soln. A; c.f. Sect. 5.3) and Ac₂O in THF (capping soln. B; c.f. Sect. 5.3), and the mixture was shaken for 20 min. The solvent was filtered, the support was washed with DMF, MeOH, acetone, $Et₂O$, and dried to afford $10a - 10d$.

Loading-Capacity Determination. Solid support $(5-10$ mg) was suspended in 5 ml of 0.1M DBU in MeCN, and the absorbance was measured at 305 nm ($\varepsilon = 9100$). The loading capacity was calculated from the following formula,

Loading capacity $[\mu mol/g] = A_{305nm} \times 5/9100 \times mg(\text{support})$ and determined to be 19 $\mu mol/g$ for 10a, 18 μ mol/g for 10b, 15 μ mol/g for 10c, and 12 μ mol/g for 10d.

3. Experiments Referring to Scheme 5. -9 -(α -L-Arabinopyranosyl)adenine (11a). A suspension of 37.0 g (54.1 mmol) 2a in 500 ml of 2.0m NH₃ in MeOH was stirred at r.t. for 2 days and concentrated to 60 ml. The resulting white precipitate was filtered, washed with AcOEt, Et₂O, and dried in high vacuum (0.03 Torr/r.t.) overnight to yield 12.0 g (83%) of **11a**. TLC (CH₂Cl₂/MeOH 3 :1) : R_f 0.40. ¹H-NMR (600 MHz, (D₆)DMSO): $3.58(m, H-C(3'))$; $3.73(d, J = 12.1, H-C(5'))$; $3.77(m, H-C(4'))$; $3.80(dd, J = 1.7, 12.1, H-C(5'))$; $4.19(m, H-C(5'))$ $H-C(2')$; 4.80 (d, $J = 5.0$, $HO-C(4')$); 5.05 (d, $J = 5.7$, $HO-C(3')$); 5.26 (d, $HO-C(2')$); 5.31 (d, $J = 9.3$, $H-C(1')$); 8.15 (s, $H-C(2)$); 8.26 (s, $H-C(8)$). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 69.40 (C(4')); 69.64 $(C(2'))$; 70.10 $(C(5'))$; 74.37 $(C(3'))$; 83.98 $(C(1'))$; 119.30 $(C(5))$; 139.99 $(C(8))$; 150.70 $(C(4))$; 153.47 $(C(2))$; $156.79 \text{ (C}(6))$. ESI-MS (pos.): 290 (10, $[M + Na]$ ⁺), 268 (100, $[M + H]$ ⁺). ESI-MS (neg.): 302 (100, $[M + Cl]$ ⁻).

9-(a -L-Arabinopyranosyl)guanine (11d). A suspension of 12.8 g (19.2 mmol) of 2d in 200 ml of sat. NH₃ in MeOH was stirred at r.t. for 2 days. The resulting white precipitate was filtered, washed with AcOEt, Et₂O, and dried in high vacuum (0.03 Torr/r.t.) to afford 5.17 g (95%) of 11d. TLC (CH₂Cl₂/MeOH 3:1): R_f 0.5. ¹H-NMR

 $(600 \text{ MHz}, (D_6) \text{ DMSO})$: 3.49 $(dd, J = 3.2, 9.2, H - C(3'))$; 3.61 $(d, J = 12.2, H - C(5'))$; 3.73 (br. s, $H - C(4'))$; 3.79 $(dd, J = 1.1, 12.2, H - C(5'))$; 4.02 $(dd, J = 9.3, H - C(2'))$; 4.35 (br. s, OH); 4.77 (br. s, OH); 5.04 $(d, J = 9.3,$ $H-C(1'))$; 5.28 (br. s, OH); 6.51 (br. s, NH₂–C(2)); 7.78 (s, H–C(8)). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 69.33 (C(4)); 69.61 (C(2)); 70.10 (C(5)); 74.42 (C(3)); 83.48 (C(1)); 116.95 (C(5)); 136.04 (C(8)); 152.55 $(C(4))$; 154.94 $(C(2))$; 158.19 $(C(6))$). ESI-MS (pos.): 322 (6, $[M+K]^+$), 306 (22, $[M+Na]^+$), 284 (70, $[M+K]^+$) $(H]^+$), 152 (100). ESI-MS (neg.): 318 (6, $[M+Cl]^-$), 282 (100, $[M-H]^-$), 150 (26).

9-(α-L-Arabinopyranosyl)-N²-(phenoxyacetyl)adenine (12a). To a soln. of 4.5 g (16.8 mmol) of 11a (coevaporated with 2×50 ml pyridine) in 70 ml dry pyridine, under Ar, was added 12.8 ml (0.1 mol) of TMSCl, and the mixture was stirred at r.t. for 30 min. In a separate flask under Ar, 2.73 g (20.2 mmol) of 1-HOBT (coevaporated with 2×40 ml pyridine) was dissolved in 1.63 ml (20.2 mmol) pyridine and 20 ml dry CH₂Cl₂, and 2.79 ml (20.2 mmol) of phenoxyacetyl chloride was added. This soln. was cooled to 0° , and the pyridine soln. containing TMS-11a was added. The mixture was stirred overnight at ambient temp., then cooled to 0° , and 20 ml H2O was added, and the mixture was stirred for 20 min. The solvent was evaporated, the residue coevaporated with 2×20 ml of H_2O under reduced pressure, and 20 ml of H_2O was added to the residue. The resulting crystals were filtered, washed with 2×40 ml of i-PrOH, 3×100 ml of Et₂O, and dried under high vacuum (0.03 Torr/r.t.) overnight to afford 5.8 g (85.1%) of 12c. TLC (CH₂Cl₂/MeOH 3:1): R_f 0.75. ¹H-NMR $(600 \text{ MHz}, (D_6) \text{ DMSO})$: 3.61 $(dd, J = 3.2, 9.2, H - C(3'))$; 3.79 – 3.82 $(m, H - C(4'))$, $H - C(5'))$; 3.84 $(dd, J = 1.8$, 12.5, H – C(5')); 4.27 (dd, J = 9.2, 9.2, H – C(2')); 5.05 (s, ArOCH₂); 5.47 (d, J = 9.1, H – C(1')); 6.97 (*m*, 3 arom. H); 7.31 (m, 2 arom. H); 8.73 (s, H-C(8)); 8.82 (s, H-C(2)); 10.99 (br. s, NHCO). 13C-NMR (150.9 MHz, (D_6) DMSO): 68.00 (CH₂O); 69.36 (C(4')); 69.65 (C(2')); 70.30 (C(5')); 74.24 (C(3')); 84.41 (C(1')); 115.39 (arom. C); 121.94 (arom. C); 123.58 (C(5)); 130.36 (arom. C); 143.85 (C(8)); 149.61 (C(4)); 152.59 (arom. H); 153.07 (C(2)); 158.64 (C(6)); 168.34 (CO of PhO). ESI-MS (pos): 424 (4, $[M + Na]⁺$), 402 (93, $[M + H]⁺$), 270 $(100, [G^{PAC}]^+)$. ESI-MS (neg.): 436 (100, $[M+Cl]^-$), 400 (59, $[M-H]^-$), 174 (20).

 N^4 -Isobutyryl-1-(2',3',4'-tri-O-benzoyl-a-t-arabinopyranosyl)cytosine (11c). A suspension of 20 g (35.3 mmol) of 1 and 5.8 g (31.8 mmol) of 4-N-Isobutyrylcytosine in 200 ml dry MeCN was warmed to 60° (oil bath). Addition of 19.6 ml (79.4 mmol) BSA resulted in a clear soln. After 30 min, 11.1 ml (95.3 mmol) of $SnCl₄$ was added dropwise (\rightarrow exothermic reaction), and stirring was continued for another 60 min. The mixture was cooled to r.t. and poured into a mixture of cold sat. aq. NaHCO₃ soln./AcOEt 1:1, (v/v) with stirring. The aq. phase was extracted with 3×150 ml AcOEt, and washed successively with sat. aq. Na₂CO₃ ($2 \times$), H₂O and sat. aq. NaCl soln. The org. phase was dried (MgSO₄) and evaporated. The resulting oil was purified by CC (silica gel; hexane/AcOEt 1:2). The product fractions were combined, evaporated (rotavap), and dried under high vacuum (ca. 0.5 Torr/r.t.) to furnish 18.3 g (83%) of 11c. Colorless amorphous solid. TLC (hexane/AcOEt 1:2): R_f 0.35. ¹H-NMR (600 MHz, CDCl₃): 1.17 (2d, J = 6.6, Me₂CH); 2.84 (m, H – C(9)); 4.22 (d, J = 13.2, H – C(5')); 4.43 $(dd, J = 13.8, 1.20, H-C(5'))$; 5.83 – 5.86 $(m, H-C(3'), H-C(4'))$; 5.98 $(t, J = 9.6, H-C(2'))$; 6.45 $(d, J = 10, 1.4)$ 9.6, H $-C(1')$); 7.27 – 7.48 (*m*, 6 arom. H); 7.55 – 7.59 (*m*, H $-C(5)$, 2 arom. H); 7.63 – 7.94 (*m*, 6 arom.); 7.97 (*d*, $J = 7.8, H - C(6)$; 8.11 – 8.13 (*m*, 4 arom. H). ¹³C-NMR (150.9 MHz, CDCl₃): 19.29 (*Me*₂CH); 36.92 (Me2CHCO); 67.96 (C(5); 69.43, 69.63, 72.08 (C(4), C(3), C(2)); 82.39 (C(1)); 98.16 (C(5)); 128.58; 128.68; 128.79; 128.90; 129.04; 129.18; 129.69; 130.14; 130.17; 130.38; 130.50; 133.65; 133.68; 134.13; 145.07 $(C(6))$; 154.95 $(C(2))$; 163.78 $(C(4))$; 165.70, 165.94, 166.07, 171.63 (4 CO of Bz), 178.33 $(C(8))$. MALDI-FTMS (DHB): 648.1976 ($[M + Na]$ ⁺, calc. 648.1958).

 $1-(a$ -L-Arabinopyranosyl)-N⁴-isobutyrylcytosine (12c). A soln. of 18 g (28.8 mmol) of 11c in 300 ml 2.0m $NH₃$ in MeOH was stirred for 6 h at r.t. The clear soln. was concentrated in vacuo, and the residual gum was dissolved in H₂O (300 ml) and Et₂O (100 ml). The resulting aq. phase was lyophilized, and the resulting material was dissolved in 200 ml of dry DMF, and 8 ml (48 mmol) isobutyric anhydride was added dropwise under Ar. The mixture was stirred at r.t. for 15 h and concentrated in vacuo. The resulting residue was purified by CC (silica gel; CH₂Cl₂/MeOH 6:1) to give 6.4 g (70%) of **12c**. TLC (CH₂Cl₂/MeOH 6:1): R_f 0.22. ¹H-NMR (600 MHz, (D_6) DMSO): ¹H-NMR (600 MHz, (D_6) DMSO): 1.05 (2d, J = 6.6, Me₂CH); 2.74 (m, Me₂CHCO); 3.50 (dd, $J = 9.0; 2.4$, H $-C(2^{\prime})$); 3.63 (d, $J = 11.4$, H $-C(5^{\prime})$); $3.70 - 3.73$ (m, H $-C(3^{\prime})$, H $-C(4^{\prime})$); 3.97 (dd, $J =$ 12.0, 1.20, H-C(5')); 4.74 $(d, J = 4.2, HO - C(4'))$; 4.97 $(s, H - C(3'))$; 5.21 $(d, J = 4.8, HO - C(2'))$; 5.44 $(d, J = 4.8)$ 9.0, H – C(1')); 7.27 (d, J = 7.2, H – C(5)); 8.03 (d, J = 7.2, H – C(6)); 10.86 (s, H – N(7)). ¹³C-NMR (150.9 MHz, (D_6) DMSO): 19.71, 19.89 (Me₂CH); 35.73 (Me₂CHCO); 69.35, 69.86, 70.41 (C(4'), C(2'), C(5')); 74.19 (C(3')); 84.95 (C(1')); 96.86 (C(5)); 146.98 (C(6)); 155.17 (C(2)); 163.36 (C(4)); 178.70 (C(8), C=O). ESI-MS (pos.): 336 (100, $[M + Na]$ ⁺), 314 (65, $[M + H]$ ⁺).

 $9-(a$ -L-Arabinopyranosyl)-N²-(phenoxyacetyl)guanine (12d). To a soln. of 8.4 g (29.6 mmol) of 11d (coevaporated with 2×100 ml pyridine) in 100 ml dry pyridine, under Ar was added 26.2 ml (0.21 mol) of TMSCl, and the mixture was stirred at r.t. for 30 min. In a separate flask under Ar, 5.2 g (38.4 mmol) of 1-HOBT (coevaporated with 2×40 ml pyridine) was dissolved in 3.1 ml (38.4 mmol) pyridine and 20 ml dry CH₂Cl₂, and 5.3 ml (38.4 mmol) of phenoxyacetyl chloride was added. This soln. was cooled to 0° , and the pyridine soln. containing TMS-11d was added. The mixture was stirred overnight at r.t., cooled to 0° , 20 ml H₂O was added, and the mixture was stirred for 20 min. The solvent was evaporated, the residue was co-evaporated with $2 \times$ 20 ml of H2O under reduced pressure, and 40 ml of H2O was added to the residue. The resulting white crystals were filtered, washed with 2×40 ml of i-PrOH, 3×100 ml of Et₂O, and dried under high vacuum (0.03 Torr/r.t.) overnight to afford 10.6 g (85.9%) of 12d. TLC (CH2Cl2/MeOH 3 :1): R_f 0.8. 1 H-NMR (600 MHz, (D₆)DMSO): 3.53 (dd, J = 3.1, 9.3, H-C(3')); 3.66 (d, J = 12.3, H-C(5')); 3.76 (br. s, H-C(4')); 3.84 (dd, J = 1.7, 12.3, $H-C(5')$; 4.10 (dd, J = 9.3, H-C(2')); 4.86 (s, PhOCH₂); 5.14 (d, J = 9.3, H-C(1')); 6.97 (m, 3 arom. H); 7.31 $(m, 2 \text{ arom. H}); 8.14 (s, H-C(8)); 11.80 (s, NH(1)); 11.89 (s, NHCOCH₂).$ ¹³C-NMR (150.9 MHz, (D_6) DMSO): 67.00 (CH₂O); 69.27 (C(4')); 69.68 (C(2')); 70.38 (C(5')); 74.32 (C(3')); 84.22 (C(1')); 115.38 (arom. C); 120.81 $(C(5))$; 122.17 (arom. C); 130.41 (arom. C); 138.83 (C(8)); 148.15 (arom. C); 149.75 (C(4)); 155.76 (C(2)); 158.44 (C(6)); 171.88 (CO of PhO). ESI-MS (pos.): 440 (100, $[M + Na]^+$), 418 (54, $[M + H]^+$), 308 (51), 286 (80) . ESI-MS (neg.): 452 $(8, [M + Cl]^{-})$, 416 $(100, [M - H]^{-})$.

9-{2'-O-[(4",4"'-Dimethoxytriphenyl)methyl]- a -L-arabinopyranosyl]-N²-(phenoxyacetyl)adenine (**13a**). To a soln. of 6.5 g (16.2 mmol) of $12a$ in 140 ml dry pyridine/DMF 1:1, cooled to 0° , was added 3.2 ml (24.3 mmol) of N,N-dimethylformamide dimethyl acetal slowly, and the soln. was stirred for 1 h. The solvent was evaporated under high vacuum at r.t., and to the resulting residue was added a soln. of 6.6 g (19.4 mmol) of DMT-Cl in 50 ml dry CH₂Cl₂ with stirring, followed by 6.4 ml (48.6 mmol) of collidine. After consumption of the starting material (TLC, ca. 6 h), 10 ml MeOH was added, and the soln. was stirred for another 30 min. The mixture was diluted with 300 ml of CH_2Cl_2 , and the soln. was washed sequentially with sat. aq. NaHCO₃ soln., 10% aq. citric acid soln., sat. aq. NaHCO₃ soln., and sat. aq. NaCl soln. The org. phase was evaporated to yield a yellow foam, which was purified by CC (silica gel; washed with AcOEt until yellow color is removed and then eluted with 20:1 CH₂Cl₂/i-PrOH) to afford 8.1 g (71%) of **13a**. TLC (CH₂Cl₂/MeOH 7:1): R_f 0.7. ¹H-NMR (600 MHz, (D_6) DMSO): 3.62 – 3.70 $(m, H-C(5'))$; 3.63, 3.67 $(2s, MeO)$; 3.88 $(m, H-C(4'))$; 4.00 $(m, H-C(3'))$; 4.09 $(dd,$ $J = 5.9, H - C(2')$; 4.49 $(d, J = 5.3, HO - C(3'))$; 4.80 $(d, J = 6.0, HO - C(4'))$; 5.00 $(s, PhOCH₂)$; 5.64 $(d, J = 5.4,$ $H-C(1')$; 6.61–7.31 (*m*, 18 arom. H); 8.28 (*s*, H–C(8)); 8.62 (*s*, H–C(2)); 10.86 (br. *s*, NHCO). ¹³C-NMR (150.9 MHz, (D6)DMSO): 55.71 (MeO); 55.75 (MeO); 65.81 (C(5)); 67.51 (C(4)); 67.98 (CH2O); 72.34 (C(3)); 73.84 (C(2)); 82.35 (C(1)); 113.65; 113.50; 115.40; 121.92; 123.34; 127.26; 128.23; 128.42; 130.34; 130.91; 131.18; 136.38; 136.55; 143.83 (C(8)); 146.89; 149.52, 152.31 (C(2)); 152.66; 158.66; 158.82; 158.91; 168.11 (CO of PhO). ESI-MS (pos.): 726 (16, $[M + Na]$ ⁺), 704 (100, $[M + H]$ ⁺), 424 (54). ESI-MS (neg.): 738 $(6, [M + Cl]^{-})$, 702 (40, $[M - H]^{-})$, 552 (98), 436 (100).

 $1-(2'-O-[(4'',4'''-Dimensionaryt)](methyl)-a-L-arabinopyranosyllthymine (13b))$. The reaction was performed as described for 13c in 200 ml of dry DMF in the presence of molecular sieves $(4 \AA)$ with 10 g (38.8 mmol) of 3b, 20 ml of collidine, 7.7 ml (58.1 mmol) of N,N-dimethylformamide dimethyl acetal, 100 ml of CH2Cl2 , 20 ml of collidine, and 20 g (58.1 mmol) of DMT-Cl at r.t. Workup as described for 13c and purification by CC (silica gel; hexane/AcOEt 1:4) gave 16.2 g (72%) of **13b**. TLC (hexane/AcOEt 1:4): R_f 0.45. ¹H-NMR $(600 \text{ MHz}, (D_6) \text{ DMSO})$: 1.57 (s, Me – C(5)); 3.52 (t, J = 9.0, H – C(2'); 3.58 (d, J = 6.6, H – C(5')); 3.60 (m, $H-C(4')$; 3.68 (*m*, $H-C(5')$, $HO-C(4')$); 3.72, 3.73 (2s, 2 MeO); 3.86 (*m*, $H-C(3')$); 4.60 (*d*, $J=4.8$, $HO-C(3'))$; 5.72 $(d, J = 9.0, H-C(1'))$; 6.50 $(s, H-C(6))$; 6.77 – 6.83 $(m, 4 \text{ atom. H})$; 7.13 – 7.20 $(m, 7 \text{ atom. H})$; 7.31 – 7.33 $(m, 2 \text{ atom. H})$; 11.38 $(s, H-N(3))$. ¹³C-NMR (150.9 MHz, (D_6) DMSO): 12.97 $(Me-C(5))$; 55.71 (MeO) ; 69.24 $(C(5'))$; 69.32, 72.71, 74.11 $(C(2'), C(3'), C(4'))$; 82.69 $(C(1'))$; 87.92 (Ar_3C) ; 109 $(C(5))$; 113.35; 127.01; 128.04; 128.65; 129.76; 131.28; 131.57; 137.05 (C(6)); 137.05; 137.28; 147.61; 152.09 (C(2)); 158.66; 158.73; 164.18 (C(4)). ESI-MS (pos.): 583 (92, $[M + Na]^+$), 303 (100, $[DMT]^+$).

 $1-(2'-O-f(4'',4'''-Dimethoxytriphenyl/methyl)-\alpha-l-arabinopyranosyl}-N'+isobutyrylcytosine$ (13c). To the soln. of 10.5 g (33.5 mmol) of 12c in 200 ml of dry DMF were added molecular sieves (4 Å) , 20 ml of collidine, and 6.7 ml (50.3 mmol) of N,N-dimethylformamide dimethyl acetal at r.t. The soln. was stirred for 3 h and then concentrated to dryness. The residue was dissolved in 100 ml of $CH₂Cl₂$, and 20 ml of collidine was added, followed by 17 g (50.3 mmol) of DMT-Cl. The mixture was stirred overnight, quenched with MeOH, and filtered through Celite. The filtrate was washed successively with 10% aq. citric acid, sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and dried (Na₂SO₄) and concentrated to dryness. The residue was purified by CC (silica gel; hexane/ EtOAc 1:4) to give 15 g (72%) of **13c**. TLC (hexane/EtOAc 1:4): R_f 0.4. ¹H-NMR (600 MHz, (D₆)DMSO): 1.04 (2d, $J = 6.6$, Me_2CH), 2.70 (m, Me₂CHCO); 3.56 (d, $J = 6.4$, H – C(5')); 3.61 (t, $J = 9.0$, H – C(2'); 3.64 (br. s, $H-C(4')$; 3.70–3.72 (m, 2 MeO, $H-C(5')$, $HO-C(3')$; 3.82 (m, $H-C(3')$); 4.65 (d, $J = 5.4$, $HO-C(4')$); 5.90 $(d, J = 8.4, H - C(1'))$: 6.71 – 6.78 (*m*, 4 arom. H); 6.83 (*d*, *J* = 9.0, 1 arom. H); 6.93 (*d*, *J* = 7.8, H – C(5)); 7.04 – 7.36 (*m*, H-C(6), 8 arom. H); 10.87 (*s*, H-N(7)). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 19.65, 19.88 (*Me*₂CH);

35.71 (Me2CHCO); 55.74, 55.82 (MeO); 69.01, 69.25, 73.41, 73.89 (C(5), C(4), C(2), (C(3)); 83.90 (C(1)); 87.96 (Ar3C); 96.76 (C(5)); 113.49; 113.59; 126.99; 127.27; 128.13; 128.25; 128.49; 128.71; 129.76; 131.25; 131.51; 136.88; 137.12; 141.07; 146.08 (C(6)); 147.38; 149.20; 155.72 (C(2)); 158.64; 158.73; 163.18 (C(4)); 178.60 (C(8), C=O). ESI-MS (pos.): 638 (100, $[M + Na]^+$), 303 (30, [DMT]⁺).

9-{2'-O-[(4",4"'-Dimethoxytriphenyl)methyl]-a-l-arabinopyranosyl]-N²-(phenoxyacetyl)guanine (**13d**). To a soln. of 5.8 g (13.9 mmol) of 12d in 140 ml dry pyridine/DMF 1:1, cooled to 0° , was added slowly 2.8 ml (20.9 mmol) of N,N-dimethylformamide dimethyl acetal, and the soln. was stirred for 1 h. The solvent was evaporated under high vacuum at r.t., and to the resulting residue was added a soln. of 5.7 g (16.7 mmol) of DMT-Cl in 50 ml dry CH₂Cl₂ with stirring, followed by 5.5 ml (41.7 mmol) of collidine. After consumption of the starting material (TLC, ca. 6 h), 10 ml MeOH was added, and the soln. was stirred for another 30 min. The mixture was diluted with 300 ml of CH_2Cl_2 , and the soln. was washed sequentially with sat. aq. NaHCO₃ soln., 10% aq. citric acid soln., sat. aq. NaHCO₃ soln., and sat. aq. NaCl soln. The org. phase was evaporated to yield a yellow foam, which was purified by CC (silica gel; washed with AcOEt containing 1% Et₃N, until yellow color is removed and then eluted with CH₂Cl₂/i-PrOH containing 1% Et₃N 20:1 to afford 18.3 g (61%) of **13d**. TLC $(CH_2Cl_2/i\text{-}Problem 10:1): R_f 0.5.$ 'H-NMR (600 MHz, (D_6) DMSO): 3.57 – 3.69 $(m, 2H-C(5'))$; 3.64, 3.68 (2s, MeO); 3.84 (br. s, H–C(4')); 3.90 (m, H–C(2')); 3.96 (br. s, H–C(3')); 4.52 (br. s, HO–C(3')); 4.80 (d, J = 4.9, $HO-C(4'))$; 4.87 (d, J = 15.8, 1 H, PhOCH₂); 4.93 (d, J = 15.9, 1 H, PhOCH₂); 5.42 (d, J = 5.3, H-C(1')); 6.68-7.33 (*m*, 18 arom. H); 7.76 (*s*, H – C(8)); 11.90 (br. *s*, 2 NH). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 55.71 (MeO); 55.77 (MeO); 65.94 (C(5)); 67.29 (ArOCH2); 67.60 (C(4)); 72.52 (C(3)); 74.17 (C(2)); 82.15 (C(1)); 113.19; 113.49; 113.65; 115.36; 120.46; 122.15; 127.31; 128.23; 130.41; 131.01; 131.29; 136.35; 139.24 (C(8)); 146.90; 148.33; 149.60; 155.87; 158.52; 158.64; 158.87; 158.96; 172.16 (CO of PhO). ESI-MS (pos.): 821 (100, [M $Et₃NH]^+$), 742 (16, $[M+Na]^+$), 720 (40, $[M+H]^+$), 435 (44). ESI-MS (neg.): 754 (10, $[M+Cl]^-,$ 718 (100, $[M-H]$ ⁻), 584 (20).

9-{4'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-L-arabinopyranosyl]-N²-(phenoxyacetyl)adenine (14a). To a soln. of 1.8 g (2.6 mmol) 13a in 10 ml of dry CH₂Cl₂ and 10 ml of dry pyridine, cooled to -40° (acetone/dry ice bath), was added 330 ul (2.8 mmol) of BzCl slowly through a syringe over 60 min. The soln, was allowed to warm to -10° and quenched with 10 ml of sat. aq. NaHCO₃ soln., and 100 ml of CH₂Cl₂ were added. The org. layer was washed with sat. aq. NaHCO₃ soln. and dried (Na₂SO₄). The solvent was evaporated, and the residue was purified by CC (silica gel; eluted with AcOEt/hexane 2:1) to yield 1.9 g (89%) of 14a. White foam. TLC: $(CH_2Cl_2/i\text{-}Problem 10:1): R_f\ 0.85.$ 'H-NMR (600 MHz, $(D_6)\text{DMSO}$): 3.68 (s, MeO); 3.70 (s, MeO); 3.96 (d, $J = 12.1, H - C(5')); 4.02 (d, J = 12.8, H - C(5')); 4.30 (m, H - C(3')); 4.44 (m, H - C(2')); 4.58 (br. s, HO - C(3'));$ 5.01 (s, PhOCH₂); 5.23 (s, H-C(4')); 5.98 (d, J = 7.8, H-C(1'); 6.53 - 8.06 (m, 23 arom. H); 8.32 (s, H-C(8)); 8.45 (s, H-C(2)); 10.84 (br. s, NHCO). 13C-NMR (150.9 MHz, (D6)DMSO): 55.71 (MeO); 55.72 (MeO); 65.62 $(C(5'))$; 68.02 (CH_2O) ; 71.92 $(C(3'))$; 72.72 $(C(2'))$; 72.87 $(C(4'))$; 84.61 $(C(1'))$; 113.12; 113.22; 115.39; 121.92; 123.93; 126.90; 127.96; 128.31; 136.61; 144.27 (C(8)); 147.43; 149.45; 152.00 (C(2)); 152.33; 158.66; 158.74; 165.97 $(CO of Bz)$; 168.13 $(CO of PhO)$. ESI-MS (pos.): 846 (5, $[M + K]^+$), 830 (15, $[M + Na]^+$), 808 (85, $[M + H]^+$), 581 (100). ESI-MS (neg.): 806 (35, $[M-H]$ ⁻), 656 (100).

 $1-(4-\text{O-Benzov1-2'-O-}$ $(4'', 4''-\text{dimethoxytriphenyl})$ methyl]- α -L-arabinopyranosyl}thymine (14b). The reaction was performed as described for 14c with 18 g (31.1 mmol) of 13b in 100 ml of dry pyridine, 100 ml of CH₂Cl₂ and 5.2 ml (45 mmol) of BzCl at -10° for 3 h. Workup as described for **14c** with 200 ml of CH₂Cl₂ and 200 ml of sat. aq. NaHCO₃ soln., followed by CC (silica gel; hexane/AcOEt 1:3), gave 16 g (75%) of 14b and 3.1 g (16%) of 13b.

Data of **14b**. TLC (hexane/AcOEt 1:3): R_f 0.42. ¹H-NMR (600 MHz, (D₆)DMSO): 1.57 (s, Me-C(5)); 3.69 (dd, $J = 9.0, H - C(2'))$; 3.73 (s, 2 MeO); 3.85 (d, $J = 12.6, H - C(5'))$; 3.96 (d, $J = 13.2, H - C(5'))$; 4.23 (m, $H-C(3')$; 4.56 (d, J = 4.2, HO – C(3')); 5.05 (s, H – C(4')); 5.93 (d, J = 9.0, H – C(1')); 6.24 (s, H – C(6)); 6.77 – 6.81 $(m, 4 \text{ arom. H})$; 7.15 – 7.21 $(m, 7 \text{ arom. H})$; 7.34 – 7.35 $(m, 2 \text{ arom. H})$; 7.52 – 7.54 $(m, 2 \text{ arom. H})$; 7.71 – 7.75 $(m, 3 \text{ atom. H})$; 11.40 $(s, H-N(3))$. ¹³C-NMR (150.9 MHz, (D_6) DMSO): 13.27 (Me $-C(5)$); 55.51 (MeO); 66.05 (C(5)); 72.05, 72.38, 73.82 (C(2), C(3), C(4)); 82.40 (C(1)); 88.35 (Ar3C); 110.20 (C(5)); 113.32; 127.04; 128.05; 129.45; 130.00; 130.45; 131.35; 131.65; 134.39; 135.99 (C(6)); 151.73 (C(2)); 158.69; 158.83; 164.05 $(C(4))$; 165.68 (C=O). ESI-MS (pos.): 687 (26, $[M + Na]^+$), 303 (100). ESI-MS (neg.): 663 (100, $[M - H]$ ⁻).

1-{4'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-L-arabinopyranosyl]-N⁴-isobutyrylcytosine (14c). To the soln. of 7 g (11.4 mmol) of 13c in 50 ml of dry pyridine and 50 ml of CH₂Cl₂ was added dropwise 1.96 ml (17 mmol) of BzCl at -10° . The soln. was stirred for 3 h, diluted with 100 ml of CH_2Cl_2 , and washed with 100 ml of sat. aq. NaHCO₃ soln. The org. phase was dried (Na_2SO_4) , filtered, evaporated, and purified by CC (silica gel; hexane/AcOEt 1:3) to give 6.4 g (78%) of 14c and 0.9 g (15%) of 13c.

Data of **14c**. TLC (hexane/AcOEt 1:3): R_f 0.45. ¹H-NMR (600 MHz, (D₆)DMSO): 1.04 (2d, $J = 6.6$, $Me₂CH$); 2.70 (m, Me₂CHCO); 3.71 (s, 2 MeO); 3.79 (m, H-C(2')); 3.90–3.99 (m, 2 H-C(5')); 4.24 (br. s,

 $H-C(3')$; 4.48 (br. s, $HO-C(3')$); 5.12 (br. s, $H-C(4')$); 6.12 (d, $J = 8.4$, $H-C(1')$); 6.73–6.75 (m, $H-C(5)$, H-C(6), 18 arom. H); 10.89 (s, H-N(7)). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 19.62, 19.83 (Me₂CH); 35.75 $(Me₂CHCO)$; 55.76 (MeO) ; 66.74, 72.19, 72.95, 73.66 $(C(2'), C(4'), C(5'), (C(3'))$; 83.65 $(C(1'))$; 88.40 $(Ar₃C)$; 97.24 (C(5)); 113.57; 126.12; 126.99; 128.11; 128.25; 128.49; 128.76; 129.02; 129.44; 129.71; 130.16; 130.40; 130.78; 131.32; 131.63; 134.24; 145.32 (C(6)); 155.82 (C(2)); 158.71; 158.81; 163.31 (C(4)); 165.89 (CO of Bz); 178.67 (C(8)). ESI-MS (pos.): 742 (100, $[M + Na]^+$), 303 (25, [DMT]⁺).

9-{4'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-L-arabinopyranosyl}-N²-(phenoxyacetyl)guanine (14d). To 100 ml of dry CH₂Cl₂, 14.4 g (20.0 mmol) of 13d, 244 mg (2 mmol) 4-DMAP, 9.7 ml (120 mmol) dry pyridine, and 4.98 g (22.0 mmol) BzOBz were added sequentially. The soln. was stirred for 15 h at r.t. and quenched with 5 ml i-PrOH for 30 min. The mixture was washed with sat. aq. NaHCO₃ soln. and sat. aq. NaCl soln., and the org. phase was dried (Na_2SO_4) . The solvent was evaporated to yield a yellow foam, which was purified by CC (silica gel; washed with AcOEt containing 1% Et₃N, until the yellow color is removed and then eluted with CH₂Cl₂/i-PrOH containing 1% Et₃N 30:1) to afford 13.3 g (80.7%) of 14d. TLC (CH₂Cl₂/i-PrOH 20:1): R_f 0.65. ¹H-NMR (600 MHz, (D₆)DMSO): 3.66, 3.70 (2s, MeO); 3.89 (dd, J = 8.0, H – C(2')); 3.92 – 3.99 $(m, 2H-C(5'))$; 4.29 $(m, H-C(3'))$; 4.91 (br. s, HO $-C(3')$; 4.92 $(d, J = 15.7, 1 H, PhOCH₂)$; 4.99 $(d, J = 15.8$, 1 H, PhOC H_2); 5.21 (br. s, H – C(4')); 5.63 (d, J = 7.7, H – C(1')); 6.67 – 7.80 (m, 24 arom. H, H – C(8)); 11.85 (br. s, NH); 11.96 (br. s, NH). 13C-NMR (150.9 MHz, (D6)DMSO): 55.72 (MeO); 55.79 (MeO); 65.78 (C(5)); 68.09 (PhOCH2), 71.83 (C(3)); 73.11 (C(4)); 74.13 (C(2)); 82.27 (C(1)); 113.16; 113.34; 115.34; 120.09; 121.86; 127.09; 127.93; 128.61; 129.45; 130.18; 130.33; 130.42; 131.14; 131.49; 134.29; 137.12; 149.86; 150.39; 151.08; 156.55; 158.79; 158.90; 165.88 (CO of Bz); 173.32 (CO of PhO). ESI-MS (pos.): 926 (100, [M + Et₃NH]⁺), 824 $(18, [M + H]^+), 303 (100, [DMT]^+).$ ESI-MS (neg.): 822 (100, $[M - H]$ ⁻).

9-{3'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-L-arabinopyranosyl}-N²-(phenoxyacetyl)adenine (15a). To a soln. of 3.8 g (4.7 mmol) 14a in 50 ml of dry CH₂Cl₂ was added 4.5 ml (32.6 mmol) of dry Et₃N, and the mixture was stirred for 48 h at r.t. The solvent was evaporated to yield a white foam, which was purified by CC (silica gel; AcOEt/hexane 3:1) to afford 1.5 g (39%) of 15a and 1.6 g (39%) of starting material.

Data of **15a**: TLC (CH₂Cl₂/i-PrOH 10:1): R_f 0.7. ¹H-NMR (600 MHz, (D₆)DMSO): 3.55 (s, MeO); 3.65 (s, MeO); 3.84 (dd, J = 12.0, 2.1, H – C(5')); 4.04 – 4.08 (m, H – C(4'), H – C(5')); 4.51 (dd, J = 8.2, H – C(2')); 5.01 (s, PhOCH_2) ; 5.09 $(d, J = 6.8, \text{HO-C}(4');$; 5.51 $(dd, J = 8.9, 3.7, \text{H-C}(3'))$; 6.11 $(d, J = 8.2, \text{H-C}(1'))$; 6.48 - 8.59 (*m*, 23 arom. H); 8.03 (*s*, H-C(8)); 8.63 (*s*, H-C(2)); 10.87 (br. *s*, NHCO). ¹³C-NMR (150.9 MHz, (D_6) DMSO): 55.51 (MeO); 55.69 (MeO); 66.25 (C(4')); 67.99 (CH₂O); 69.09 C(5'); 70.81 (C(2')); 77.26 (C(3')); 83.86 (C(1)); 113.40; 115.38; 121.92; 123.61; 127.03; 127.96; 128.69; 129.79; 130.23, 130.34; 130.51; 131.01; 133.46; 143.48 (C(8)); 146.72; 149.55; 152.36 (C(2)); 152.41; 158.56; 158.66; 158.80; 165.94 (CO of Bz); 168.12 (CO of Ph). ESI-MS (pos.): 846 (60, $[M + K]^+$), 830 (93, $[M + Na]^+$), 808 (100, $[M + H]^+$). ESI-MS (neg.): 842 (11, $[M+Cl]$ ⁻); 806 (100, $[M-H]$ ⁻).

 $1-\frac{3}{2}$ -O-Benzoyl-2'-O- \int (4",4"'-dimethoxytriphenyl)methyl]-a-t-arabinopyranosyllthymine (15b). To the soln. of 5 g (7.5 mmol) of 14b in 100 ml of dry CH₂Cl₂ was added 10.5 ml (75 mmol) of Et₃N at r.t. The soln. was stirred for 12 h and then concentrated. The residue was purified by CC (silica gel; hexane/AcOEt 1:3) to give 2.25 g (45%) of **15b** and 2.15 g (43%) of **14b**.

Data of **15b**. TLC (hexane/AcOEt 1:3): R_f 0.40. ¹H-NMR (600 MHz, (D₆)DMSO): 1.57 (s, Me-C(5)); $3.57, 3.68$ $(2s, \text{MeO})$; 3.73 $(d, J = 10.8, \text{H} - \text{C}(5'))$; $3.92 - 3.94$ $(m, \text{H} - \text{C}(5'))$, $\text{H} - \text{C}(4'))$; 4.06 $(t, J = 9.0, \text{H} - \text{C}(2'))$; 5.12 (d, J = 7.2, HO-C(4')); 5.38 (dd, J = 9.0, 3.6, H-C(3')); 5.96 (d, J = 9.0, H-C(1')); 6.50 (s, H-C(6)); 6.57 $-$ 6.58 (*m*, 1 arom. H); 6.77 $-$ 6.78 (*m*, 2 arom. H); 6.97 $-$ 7.19 (*m*, 9 arom. H); 7.39 $-$ 7.41 (*m*, 3 arom. H); 7.54 – 7.59 (*m*, 3 arom. H); 11.33 (*s*, H – N(3)). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 13.04 (Me – C(5)); 55.51 (MeO); 55.71 (MeO); 66.29 (C(5')); 69.12, 69.86 (C(2'), C(4')); 77.55 (C(3')); 82.79 (C(1')); 88.12 (Ar₃C); 110.20 (C(5)); 113.40; 113.47; 127.11; 128.07; 128.38; 128.64; 129.82; 130.58; 131.40; 133.42; 136.46 (C(6)); 151.63 $(C(2))$; 158.58; 158.85; 164.17 $(C(4))$; 165.90 $(C=O)$. ESI-MS (pos.): 687(30, $[M+Na]^+$), 303 (100)

1-{3'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-l-arabinopyranosyl]-N⁴-isobutyrylcytosine (15c). To the soln. of 4.0 g (5.5 mmol) of 14c in 100 ml of dry CH₂Cl₂ was added 7.7 ml (55.5 mmol) of Et₃N at r.t. The soln. was stirred for 12 h and then concentrated. The residue was purified by CC (silica gel; hexane/ AcOEt 1:3) to give 1.9 g (48%) of 15c and 1.83 g (46%) of 14c.

Data of **15c**. TLC (hexane/AcOEt 1:3): R_f 0.43. ¹H-NMR (600 MHz, (D₆)DMSO): 1.04 (2*d*, $J = 6.6$, $Me₂CH$); 2.69 (m, Me₂CHCO); 3.55, 3.67 (2s, 2 MeO); 3.77 (dd, J = 12.0, 1.8, H-C(5')); 3.94–3.99 (m, $H-C(4')$, $H-C(5')$); 4.10 (t, $J = 9.0$, $H-C(2')$); 5.15 (d, $J = 6.6$, $HO-C(4')$); 5.38 (dd, $J = 9.0$, 3.6, $H-C(3')$); 6.15 $(d, J = 9.0, H - C(1'))$; 6.53 $(d, J = 8.4, 2 \text{ arom. H})$; 6.67 $(d, J = 7.8, 2 \text{ arom. H})$; 6.90 $(d, J = 7.2, H - C(5))$; 6.95 – 7.59 (m, H – C(6'), 14 arom. H); 10.88 (s, H – N(7)). ¹³C-NMR (150.9 MHz, (D₆)DMSO): 19.64, 19.92 (Me_2CH) ; 35.74 (Me_2CHCO) ; 55.46, 55.66 (MeO) ; 66.22, 69.37, 70.57 $(C(2), C(4), C(5))$; 77.50 $(C(3'))$; 83.91 $(C(1'))$; 88.16 $(A₁₅C)$; 97.26 $(C(5'))$; 113.51; 113.66; 126.15; 127.05; 128.13; 128.39; 128.63; 129.04; 129.74; 129.82; 130.41; 130.55; 131.39; 133.41; 137.02; 145.68 (C(6)); 155.46 (C(2)); 158.52; 158.82; 163.16 (C(4)); 165.87 (CO of Bz); 178.69 (C(8)). ESI-MS (pos.): 742 (32, $[M + Na]^+$), 454 (37), 303 (100).

9-{3'-O-Benzoyl-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-L-arabinopyranosyl]-N²-(phenoxyacetyl)guanine (15d). To a soln. of 6.8 g (8.3 mmol) 14d in 100 ml of dry CH₂Cl₂ was added 8.1 ml (57.7 mmol) of dry Et₃N, and the mixture was stirred for 48 h at r.t. The solvent was evaporated to yield a white foam, which was purified by CC (silica gel; eluted with $40:1 \text{ CH}_2\text{Cl}_2/i\text{-}PrOH$ containing 1% Et₃N) to afford 2.8 g (41%) of **15d** (contains ca. 14.5% of **14d**, by ¹H-NMR). Also, 2.9 g (43%) of starting material **14d** was recovered.

Data of **15d**. TLC (CH₂Cl₂/i-PrOH 20:1): R_f 0.50. ¹H-NMR (600 MHz, (D₆)DMSO): 3.56, 3.65 (2s, MeO); 3.85 (d, J = 10.5, H – C(5')); 3.97 (d, J = 11.5, H – C(5')); 4.00 (m, H – C(4')); 4.29 (m, H – C(2')); 4.95 (d, J = 16.0, 1 H, PhOCH₂); 5.01 $(d, J = 16.0, 1 \text{ H}, \text{PhOCH}_2)$; 5.03 (br. s, HO-C(4')); 5.40 $(dd, J = 9.2, 3.4, H-C(3')$; 5.79 (d, $J = 8.5$, H $-C(1')$); 6.53 – 7.60 (m, 24 arom. H, H $-C(8)$); 11.81 (br. s, 2 NH). ¹³C-NMR (150.9 MHz, (D6)DMSO): 55.54 (MeO); 55.71 (MeO); 66.45 (C(4)); 67.50 (PhOCH2); 69.72 (C(5)); 71.11 (C(2)); 77.53 (C(3)); 83.62 (C(1)); 113.34; 113.61; 115.23; 115.41; 120.71; 122.12; 127.12; 127.91; 128.12; 128.68; 129.85; 130.40; 130.51; 130.74; 131.25, 133.52; 136.89; 138.69; 146.66; 147.00; 149.42; 155.89; 158.60; 158.86; 166.05 (CO of Bz); 172.26 (CO of PhO). ESI-MS (pos.): 926 (100, $[M + Et_3NH]^+$), 846 (18, $[M + Na]^+$), 824 (20, $[M +$ $[H]^+$), 303 (100, [DMT]⁺). ESI-MS (neg.): 822 (100, [*M* – H]⁻).

4. Experiments Referring to Scheme 6. $-9-3$ O-Benzoyl-4'-O- $(2$ -cyanoethoxy)(diisopropylamino)phosphino]-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]- α -L-arabinopyranosyl]-N²-(phenoxyacetyl)adenine (**16a**). To a soln. of 1.1 g (1.4 mmol) 15a in 30 ml of dry CH_2Cl_2 was added 0.95 ml (5.4 mmol) of EtN(i-Pr₂) slowly, followed by 0.61 ml (4.9 mmol) of chloro(2-cyanotheoxy)(diisopropylamino)phosphine by syringe. The mixture was stirred for 1.5 h at r.t., diluted with 50 ml of CH_2Cl_2 , and washed with sat. aq. NaHCO₃ soln. The org. phase was dried (Na₂SO₄), concentrated, and subjected to CC (silica gel; AcOEt/hexane 2:1) to afford 1.1 g (79%) of **16a** as a mixture of diastereoisomers (a/b $0.7:1$). TLC (AcOEt: hexanes $3:1$): R_f 0.75 . ¹H-NMR (600 MHz, $CDCl₃$): 1.11, 1.12 (2d, J = 6.9, Me₂CH); 1.14, 1.15 (2d, J = 7.2, Me₂CH); 2.13 (dt, J = 16.7, 6.5, 1 H, CH₂CH₂CN of isomer b); 2.26 $(dt, J = 6.5, 16.7, 1$ H, CH₂CH₂CN of isomer b); 2.51 $(t, J = 6.1, 2$ H, CH₂CH₂CN of isomer a); 3.49 (m, CH₂CH₂CN of isomer b); 3.58 (m, CH₂CH₂CN of both isomers, Me₂CH of both isomers); 3.60 (s, MeO); 3.61 (s, MeO); 3.69 (s, MeO); 3.70 (s, MeO); 3.71 (m, CH₂CH₂CN of isomer a); 3.94 (d, $J = 12.5$, $H-C(5')$ of isomer b); 3.96 (d, $J = 12.4$, $H-C(5')$ of isomer a); 4.06 (m, $H-C(5')$ of isomer b); 4.19 (dd, $J =$ 12.5, 2.3, H-C(5') of isomer a); 4.34 (br. d, $J(H,P) = 8.6$, H-C(4') of isomer a); 4.47 (br. d, $J(H,P) = 11.0$, $H-C(4')$ of isomer b); 4.59 (*m*, $H-C(2')$ of isomer a); 4.67 (*m*, $H-C(2')$ of isomer b); 4.87 (br. *s*, ArOCH₂ of both isomers); 5.53 $(m, H-C(3))$ of both isomers); 5.98 $(d, J = 8.1, H-C(1))$ of isomer b); 6.00 $(d, J = 8.1, H-C(1))$ $H-C(1')$ of isomer a); 6.48 – 7.66 (*m*, 46 arom. H); 8.83 (*s*, $H-C(2)$ of isomer b); 8.84 (*s*, $H-C(2)$ of isomer a); 9.34 (br. s, NHCO of both isomers). ¹³C-NMR (150.9 MHz, CDCl₃): 20.16, 20.21 (d, J(C,P) = 7.0, CH₂CH₂CN of isomer b); 20.67, 20.71 (d, $J(C, P) = 7.2$, CH₂CH₂CN of isomer a); 24.84, 24.88, 24.92 (Me₂CH); 43.58, 43.67 (Me₂CH of isomer a); 43.62, 43.70 (Me₂CH of isomer b); 55.28, 55.44 (MeO); 55.33, 55.51 (MeO); 57.71, 57.84 $(CH_2CH_2CN$ of isomer a, $J(C,P) = 20.2$; 58.41, 58.53 (CH₂CH₂CN of isomer b, $J(C,P) = 18.4$); 68.24 (C(5') of isomer b); 68.53 (CH₂O of both isomers); 68.66 (C(5') of isomer a); 69.03 (C(4'), $J(C,P) = 15.6$ of isomer b); 69.44 (C(4'), $J(C,P) = 9.8$ of isomer a); 70.42 (C(2') of isomer b); 70.56 (C(2') of isomer a); 75.90 (C(3') of both isomers); 83.64 (C(1) of both isomers); 113.21; 115.38; 117.84; 117.90;122.41; 122.82; 127.05; 127.11; 127.80; 128.03; 128.12; 128.21; 129.77; 129.82; 129.94; 130.19; 130.23; 131.02; 132.09; 133.13; 133.17; 135.31; 136.50; 141.95 (C(8) of both isomers); 145.41; 148.40; 148.43; 152.35; 152.38; 152.84 (C(2) of both isomers); 157.43; 157.46; 158.59; 158.67; 158.86; 158.95; 166.03, 166.06 (CO of Bz of both isomers); 166.94 (CO of PhO of both isomers). ³¹P-NMR (241 MHz, CDCl₃): 150.10, 152.46. ESI-MS (pos.): 1030 (100, $[M + Na]$ ⁺), 1008 (75, $[M + Na]$ ⁺) H]⁺). ESI-MS (neg.): 1042 (9, $[M + Cl]$ ⁻), 1006 (100, $[M - H]$ ⁻).

1-{3'-O-benzoyl-4'-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]-a-t-arabinopyranosyllthymine (16b). To a soln. of 1.67 (2.5 mmol) of 15b in 15 ml of dry CH₂Cl₂ was added successively 1.3 ml (7.5 mmol) of $EIN(i-Pr_2)$ and 1.7 ml (7.5 mmol) of chloro(2-cyanotheoxy)(diisopropylamino)phosphine at r.t. The soln. was stirred for 1 h, diluted with 30 ml AcOEt and washed with sat. aq. NaHCO₃. The org. phase was dried (Na₂SO₄), filtered, and evaporated. Purification by CC (silica gel; hexane/ AcOEt 1:1) gave 1.82 g (84%) of 16b as a mixture of diastereoisomers. TLC (hexane/EtOAc 1:1): R_f 0.41. 1 H-NMR (600 MHz, CDCl₃): 1.09 – 1.14 (4 br. s, 4 Me₂CH); 1.68 (s, Me – C(5)); 2.05 (dt, J = 16.8, 6.5, 1 H, CH₂CH₂CN of isomer a); 2.21 (dt, J = 16.8, 6.6, 1 H, CH₂CH₂CN of isomer a); 2.49 (t, J = 16.8, 6.6, 2 H, CH_2CH_2CN of isomer b); 3.43 (m, 1 H, CH₂CH₂CN of isomer a); 3.54 - 3.68 (m, 2 H, CH₂CH₂CN of both isomers, Me₂CH of both isomers, MeO, CH₂CH₂CN of isomer b); 3.83 $(d, J = 12.6, 1 \text{ H}, \text{H} - \text{C}(5')$ of isomer a); 3.84 $(d, J = 12.6, 1 \text{ H}, \text{H} - \text{C}(5')$ of isomer b); 3.99 $(d, J = 12.6, 1 \text{ H}, \text{H} - \text{C}(5')$ of isomer a); 4.08 $(dd, J = 12.6, 1.8$,

1 H, H – C(5') of isomer b); 4.21 (t, $J = 9.0$, H – C(2') of isomer a); 4.23 – 4.28 (m, H – C(2'), H – C(4') of isomer b); 4.39 (br. $d, J = 11.4$, H $-C(4')$ of isomer a); 5.40 (dd, $J = 9.6, 3.0$, H $-C(3')$ of isomer a); 5.43 (dd, $J = 9.6, 3.0$, $H-C(3')$ of isomer b); 6.00 $(d, J = 9.0, H-C(1')$ of isomer a); 6.01 $(d, J = 9.0, H-C(1')$ of isomer b); 6.29 (s, $H-C(6)$ of isomer a); 6.31 (s, $H-C(6)$ of isomer b); 6.58 – 7.66 (m, 38 H). ¹³C-NMR (150.9 MHz, CDCl₃): 12.80 $(Me-C(5))$ of isomer a); 12.87 $(Me-C(5))$ of isomer b); 20.05, 20.09 $(d, J(C,P) = 6.9, CH_2CH_2CN$ of isomer a); 20.61, 20.65 (d, $J(C,P) = 6.9$, CH_2CH_2CN of isomer b); 24.85, 24.95, 25.34, 25.39 (Me_2CH); 43.49, 43.57, 43.66 (Me₂CH of both isomers); 55.21, 55.34, 55.40 (MeO of both isomers); 57.64, 57.78 (d, $J(C, P) = 21.1$, CH₂CH₂CN of isomer a); 58.33, 58.45 (d, $J(C, P) = 19.4$, CH_2CH_2CN of isomer b); 67.67 (C(5') of isomer a); 67.97 (C(5') of isomer b); 69.22 (C(2') of isomer a); 69.39 (C(2') of isomer b); 69.39, 69.45 (d, $J(C, P) = 10.41$, C(4') of isomer a); 69.64, 69. 70 (d, J(C,P) = 9.8, C(4') of isomer b); 76.18 (C(3') of isomer a); 76.30 (C(3') of isomer b); 83.14 (C(1') of isomer a); 83.26 (C(1) of isomer b); 88.63 (Ar3C); 110.93 (C(5) of isomer a); 111.08 (C(5) of isomer b); 113.19; 117.71; 117.78; 127.01; 127.90; 127.98; 128.17; 128.56; 129.83; 130.06; 130.50; 133.09; 135.79 (C(6) of isomer a); 135.91 (C(6) of isomer b); 150.79 (C(2) of both isomers); 158.56; 158.77, 163.63 (C(4) of both isomers); 166.04 (CO of Bz of both isomers). ³¹P-NMR (242.9 MHz, CDCl₃): 149.95, 152.57 ESI-MS (neg.): 900 $(16, [M - Cl]^{-}, 864 (100, [M - H]^{-}).$

1-{3'-O-Benzoyl-4'-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2'-O-[(4", 4"'-dimethoxytriphenyl)methyl]- a -L-arabinopyranosyl]-N⁴-isobutyrylcytosine (16c). To a soln. of 1.9 g (2.6 mmol) of 15c in 20 ml of dry CH_2Cl_2 were added successively 1.4 ml (7.9 mmol) of EtN(i-Pr₂) and 1.8 ml (7.9 mmol) of chloro(2cyanotheoxy)(diisopropylamino)phosphine at r.t. The soln. was stirred for 1 h, diluted with 50 ml of AcOEt, and washed with sat. aq. NaHCO₃ soln. The org. phase was dried (Na₂SO₄), filtered, evaporated, and the residue was purified by CC (silica gel; hexane/AcOEt 1:2) gave 2.0 g (83%) of 16c as a mixture of diastereoisomers.

Data of Major Isomer of **16c**: TLC (hexane/AcOEt 1:2): R_f 0.41. ¹H-NMR (600 MHz, CDCl₃): 1.07 – 1.25 $(m, 6 \text{ Me}_2\text{CH})$; 2.06, 2.21 (dt, J = 16.8, 6.6, CH₂CH₂CN); 2.74 (m, 1 H, Me₂CH); 3.43 – 3.38 (m, CH₂CH₂CN, 2 Me₂CH); 3.58 (s, MeO); 3.66 (s, MeO); 3.84 (d, J = 12.6, 1 H, H – C(5')); 3.96 (d, J = 12.0, 1 H, H – C(5')); 4.24 $(t, J = 9.0, H - C(2'))$; 4.36 $(m, H - C(4'))$; 5.39 $(dd, J = 9.0, 3.0, H - C(3'))$; 6.26 $(d, J = 9.0, H - C(1'))$; 6.57 (d, J) $J = 7.8, 2$ arom. H); 6.66 $(d, J = 7.8, 2$ arom. H); 6.74 $(d, J = 7.2, H - C(5))$; 6.93 – 7.27 $(m, H - C(6), 9$ arom. H), 7.33 (t, $J = 7.8$, 2 H); 7.49 (t, $J = 7.2$, 1 H); 7.64 (d, $J = 7.2$, 2 arom. H). ¹³C-NMR (150.9 MHz, CDCl₃): 19.30, $19.52 \ (Me₂CH)$; 20.07, 20.11 $(d, J(C, P) = 6.5, CH₂CH₂CH)$; 24.85 (Me₂CH); 37.04 (Me₂CHCO); 43.51, 43.59 (d, $J(C, P) = 12.2$, Me₂CHP); 55.27, 55.41 (MeO of both isomers); 58.37, 58.50 (d, $J(C, P) = 19.2$, CH₂CH₂CN); 68.37 $(C(5'))$; 69.27, 69.28 (d, J(C,P) = 16.7, C(4')); 69.96 (C(2')); 76.36 (C(3')); 84.17 (C(1')); 88.68 (Ar₃C); 97.53 (C(5)); 113.26; 113.35; 118.49; 127.03; 127.99; 128.13; 129.76; 130.02; 130.03; 133.08; 144.76 (C(6)); 155.76 $(C(2))$; 158.57; 158.79; 162.31 $(C(4))$; 166.02 $(CO \text{ of } Bz)$; 177.47 $(C(8))$. ³¹P-NMR (242.9 MHz, CDCl₃): 150.05. ESI-MS (pos.): 943 (100, $[M + Na]$ ⁺), 840 (44), 441 (44), 303 (25).

Diagnostic Data of Minor Isomer of **16c**. TLC (hexane/AcOEt 1:2): R_f 0.46. ¹H-NMR (600 MHz, CDCl₃): 2.48 (*m*, CH₂CH₂CN); 5.42 (*dd, J* = 8.9, 3.4, H-C(3')); 6.26 (*d, J* = 8.7, H-C(1')). ³¹P-NMR (242.9 MHz, $CDCl₃$) 152.7.

9-{3'-O-Benzoyl-4'-O-[(2-cyanoethoxy)(diisopropylamino)phosphino]-2'-O-[(4",4"'-dimethoxytriphenyl)methyl]- α -l-arabinopyranosyl]-N²-(phenoxyacetyl)guanine (**16d**). To a soln. of 2 g (2.4 mmol) **15d** in 50 ml of dry CH₂Cl₂ was added 1.7 ml (9.7 mmol) of EtN(i-Pr₂) slowly, followed by 1.08 ml (4.86 mmol) of chloro(2cyanotheoxy)(diisopropylamino)phosphine by syringe. The mixture was stirred for 1.5 h at r.t., diluted with 50 ml of CH₂Cl₂, and washed with sat. aq. NaHCO₃ soln. The org. phase was dried (Na₂SO₄), concentrated, and subjected to CC (silica gel; eluted with 40:1 CH₂Cl₂/i-PrOH containing 1% Et₃N) to afford 2 g (80.4%) of **16d** as a mixture of diastereoisomers (a/b 1:0.9). TLC (CH₂Cl₂/i-PrOH 20:1): R_f 0.75. ¹H-NMR (600 MHz, CDCl₃): 1.07, 1.08, 1.10, 1.12 (br. s, Me₂CH), 2.10 (dt, J = 16.7, 6.5, 1 H, CH₂CH₂CN of isomer a); 2.22 (dt, J = 16.7, 6.5, 1 H, CH₂CH₂CN of isomer a); 2.50 (t, $J = 6.1$, CH₂CH₂CN of isomer b); 3.48 (m, CH₂CH₂CN, of isomer a); 3.55 (m, CH_2CH_2CN) of both isomers, Me₂CH of both isomers); 3.57 (s, MeO of isomer b); 3.58 (s, MeO); 3.67 (s, MeO); 3.68 (s, MeO); 3.70 (m, CH₂CH₂CN of isomer b); 3.88 (d, J = 12.2, H – C(5') of isomer a); 3.89 (d, J = 12.2, $H-C(5')$ of isomer b); 4.01 (dd, $J = 12.4$, 3.1, $H-C(5')$ of isomer a); 4.14 (dd, $J = 2.7$, 12.4, $H-C(5')$ of isomer b); 4.30 (br. d , $J(H, P) = 8.5$, $H - C(4')$ of isomer b); 4.37 (m, $H - C(2')$ of isomer b); 4.45 (br. d , $J(H, P) =$ 11.0, $H-C(4')$, of isomer a); 4.55 $(m, H-C(2'))$ of isomer a); 4.83 (br. s, PhOCH₂ of both isomers); 5.49 $(m,$ $H-C(3')$ of both isomers); 5.67 $(d, J = 7.7, H-C(1')$ of isomer a); 5.72 $(d, J = 8.1, H-C(1')$ of isomer a); 6.52-7.62 (m, 48 arom. H). ¹³C-NMR (150.9 MHz, CDCl₃): 20.10, 20.15 (d, J(C,P) = 7.0, CH₂CH₂CN of isomer a); 20.64, 20.69 (d, J(C,P) = 7.1, CH₂CH₂CN of isomer b); 24.78, 24.83, 24.88, 24.92 (Me₂CH); 43.53, 43.62 (Me₂CH) of isomer b); 43.58, 43.66 (Me₂CH of isomer a); 55.29, 55.49 (MeO); 55.32, 55.55 (MeO); 57.64, 57.77 $(CH_2CH_2CN, J(C,P) = 20.0$ of isomer b); 58.39, 58.51 $(CH_2CH_2CN, J(C,P) = 18.9$ of isomer a); 67.75 (CH₂O of both isomers); 67.84 (C(5') of isomer a); 68.39 (C(5') of isomer b); 68.81 (C(4') of isomer a); 69.36 (C(4') of isomer b); 70.40 (C(2') of isomer a); 70.78 (C(2') of isomer b); 75.81 (C(3') of both isomers); 83.18 (C(1') of isomer b); 83.74 (C(1) of isomer a); 113.23; 115.36; 115.41; 117.83; 117.87; 121.15; 121.40; 123.24; 126.98; 127.03; 127.67; 127.74; 128.03; 128.23; 129.52; 129.68; 129.75; 129.90; 130.34; 130.40; 130.42; 131.29; 133.18; 133.24; 136.44; 137.71; 137.99; 145.85; 147.46; 148.78; 148.84; 156.75; 157.14; 157.22; 158.63; 158.68; 158.94; 158.99; 166.12 (CO of Bz of both isomers); 170.48, 170.55 (CO of PhO of both isomers). ³¹P-NMR (241 MHz, CDCl₃): 150.30; 152.50. ESI-MS (pos.): 1126 (100, $[M + Et_3NH]^+$), 1047 (45, $[M + Na]^+$), 1025 (50, $[M + H]^+$). ESI-MS $(neg.): 1023 (100, [M - H]^-)$, 970 (50), 822 (90).

Preparation of Nucleoside-Derivatized Controlled-Pore Glass (CPG) 17c. To a soln. of 300 mg (0.416 mmol) of 15c in 5 ml of dry CH₂Cl₂ were added successively 76 mg (0.62 mmol) of DMAP and 83 mg (0.83 mmol) of succinic anhydride at r.t. The mixture was stirred for 5 h, diluted with 40 ml of CH₂Cl₂, and washed with 20 ml of 10% aq. citric acid soln. The org. layer was dried (Na₂SO₄), filtered, and evaporated. The residue was purified by CC (silica gel; CH₂Cl₂/MeOH 9:1). The product fractions were collected, evaporated, and dried for 1 h in vacuo (0.2 Torr). This material was dissolved 10 ml dry MeCN, followed by successive addition of 104 µl (949 mmol) N-methylmorpholine, 155 mg (475 mmol) TOTU, and 1.5 g LCAA-CPG. The suspension was gently shaken for 3 h at r.t. Filtration and washing with DMF, MeOH, acetone, and Et₂O afforded 17c after drying in vacuo (0.2 Torr). A suspension of the nucleoside-derivatized solid support 17c in 25 ml dry pyridine and 3.0 ml 1-methyl-1H-imidazole was treated with 3.0 ml Ac₂O for 45 min. Filtration and washing with DMF, MeOH, acetone, and Et₂O afforded capped 17c after drying in vacuo (0.2 Torr) for 3 h. The loading capacity was determined to be 52.5 μ mol/g (by the method described in [13c]).

5. Automated Solid-Phase Synthesis on a Perseptive Expedite Gene Synthesizer. - Oligonucleotide syntheses were carried out on a 0.5 -, 1-, and 10 -um scales. The DNA/RNA synthesizer column was filled with the CPG solid support loaded with the appropriate nucleobase. The substrates and reagents required were prepared as follows.

5.1. With the Fmoc-Protected Building Blocks. To perform the oligonucleotide synthesis in a 0.5-umol scale, 30 ± 40 mg of the corresponding solid support were filled in a commercially available column (height 1.5 mm, diameter 0.3 mm). Before placing the appropriate column on the DNA synthesizer in the reaction chamber, the desired synthesis parameters (sequence, synthesis cycle) were specified, and the reagents/solvents required for the oligonucleotide synthesis were prepared and installed as follows.

Pre-Automation Procedures. 5.1.1. Fmoc-Phosphoramidites. A 0.075 M soln. of phosphoroamidates in dry MeCN (dried over 3-A molecular sieves $(8-12 \text{ mesh})$, freshly activated by heating at *ca.* 300 $^{\circ}$ under high vacuum overnight) was prepared. The number of couplings for the particular phosphoramidite was increased by two additional couplings to give a factor, which was multiplied by 16.0 mg for the thymine (8b) and 18.0 mg for the adenine $(8a)$, cytosine $(8c)$, and guanine $(8d)$ phosphoramidites. The amount of the corresponding phosphoramidite was dissolved in 289 μ dry MeCN (17 pulses at 17 μ each); this resulted in a 22-fold excess of phosphoramidite over the CPG-solid support. The solns. were kept over molecular sieves $(4 \text{ Å}, \text{mesh } 8-12)$ for 2 h prior to use, and used within $24 h^{18}$).

5.1.2. Washing. MeCN (wash) was purchased from PerSeptive Biosysytem™ and stored over molecular sieves (4 Å, mesh $8-12$) at least 2 h before use ($GEN089865$).

5.1.3. Oxidizing Soln. Oxidizer soln. from PerSeptive BiosysytemTM containing I₂ in pyridine/H₂O/THF (GEN089850).

5.1.4. Deblocking Soln. 0.1 DBU in MeCN was prepared by dissolving 1.5 ml of DBU (Aldrich) in 100 ml of MeCN. The soln. was kept over molecular sieves $(4 \text{ Å}, \text{mesh } 8-12)$ at least 2h before use.

5.1.5. Activator Soln. 0.5 pyridine hydrochloride in MeCN was prepared by dissolving 5.78 g of pyridine hydrochloride (Fluka) in 100 ml of anh. MeCN. The activator soln. was stored 12 h over molecular sieves (4 \AA , mesh 8-12) before use. Subsequently, it was replaced by a 0.5M soln. of $1H$ -tetrazole in MeCN as activator from PerSeptive Biosysytem™ (GEN089880).

5.1.6. Capping Soln. A: capping reagent A from PerSeptive BiosysytemTM containing Ac₂O in THF (GEN089810).

5.1.7. Capping Soln. B: from PerSeptive BiosysytemTM containing 1-methyl-1H-imidazole in a mixture of THF/pyridine (GEN089820).

The synthesis of oligonucleotides with the Perseptive Expedite Gene Synthesizer required the following modifications to the protocol provided by Perseptive for the DNA/RNA synthesis: a) the duration of the coupling time of phosphoramidite was ca. 15 min with either 0.5M pyridine hydrocloride or 0.5M 1H-tetrazole in MeCN, and b) the 'deblocking' was accomplished by 0.1M DBU in MeCN over a 1.5-min period. All oligonucleotides were synthesized in the 'Fmoc-off' mode, where the final Fmoc group was removed. The progress of the synthesis was followed by UV monitoring of the fulvene ion resulting from the cleavage of the Fmoc group after each coupling. In a typical 0.5-umol scale synthesis, each Fmoc-derived fulvene eluate (collected in a glass tube) was diluted to 5 ml with MeCN, and the absorbance at 305 nm (ε = 9100) was measured with a 0.1M soln. of DBU in MeCN as reference. The value obtained for each fulvene eluate was compared to that of the fulvene eluate from the previous step, and the coupling efficiency was determined.

5.2. Post-Automation Procedures. 5.2.1. Removal of 2-Cyanoethyl Protecting Group, Removal of Sugar and Nucleobase Protecting Groups, and Detachment from CPG Solid Support. After the automated synthesis was completed, the CPG solid support containing the oligonucleotide ('Trityl-on') was dried in vacuo for 30 min, transferred to a pear-shaped 10-ml flask, and one of the following procedure was used depending on the sequence of the oligonucleotides (*Table 3* lists the specific deprotection method for the specific sequence):

Method A. To the flask containing the dry CPG solid support was added 2.4 ml of a soln. of 0.15M MeONH₂ \cdot HCl in 25%, aq. NH₃/EtOH 3:1, and and the mixture was shaken at r.t. for 2–10 h. After deprotection, the suspension was diluted with H₂O, filtered (Nalgene, PTFE (0.2 μ m)), and evaporated to dryness to afford the crude oligonucleotides.

Method B. To the flask containing the dry CPG solid support was added 2 ml of 25% aq. NH₃ and shaken at 4° for 7–24 h. After deprotection, the suspension was diluted with H₂O, filtered (*Nalgene*, PTFE (0.2 μ m)), and evaporated to dryness to afford the crude oligonucleotides.

Method C. To the flask containing the dry CPG solid support was added 2 ml of 25% aq. $NH₃$ and shaken at 4° for 21 h, and evaporated to dryness. The residue was treated with 25% aq. NH₂NH₂ · H₂O at r.t. for 10 min. After deprotection, the suspension was diluted with H₂O, filtered (Nalgene, PTFE (0.2 μ m)), and evaporated to dryness to afford the crude oligonucleotides.

Method D. To the flask containing the dry CPG solid support was added 2 ml of a mixture of ethylenediamine/EtOH (1:1) and shaken at r.t. for 2 h. After deprotection, the suspension was diluted with H2O, filtered (Nalgene, PTFE (0.2 m)), and evaporated to dryness to afford the crude oligonucleotides.

All of the above deprotections were monitored by anion-exchange HPLC for optimum deprotection time, and the crude oligonucleotides were taken to the next step, HPLC purification.

5.3. With the Phosphoramidite Building Blocks: Pre-Automation Procedures. 5.3.1 Phosphoroamidites. The phosphoramidite soln. (ca. 0.07M) was dried over 3-Å molecular sieves $(8-12 \text{ mesh}, \text{freshly activated by heating})$ at ca. 300° under high vacuum overnight) overnight at r.t. prior to use. An excess of ca. 5 equiv. of phosphoroamidites was used.

5.3.2. Activator Soln.: 5-(ethylthio)-1H-tetrazole in dry MeCN (0.35), which was dried over freshly activated 4-ä molecular sieves for 24 h at r.t.

5.3.3. Capping Soln. A: 1-methyl-1H-imidazole in pyridine/THF, supplied by Perseptive Biosystems™ was used (GEN089810).

5.3.4. Capping Soln. B: Ac₂O in THF, supplied by Perseptive Biosystems™ was used (GEN089820).

5.3.5. Oxidizing Soln.: I₂ in pyridine/THF, supplied by Perseptive Biosystems™ was used (GEN089850).

5.3.6. Detritylation Reagent.: 6% (v/v) Cl₂CHCO₂H soln. in ClCH₂CH₂Cl.

The synthesis of oligonucleotides with the *Perseptive Expedite Gene Synthesizer* required the following modifications to the protocol provided by Perseptive for the DNA/RNA synthesis: a) the duration of the coupling time of phosphoramidite was about 27 min. b) use of 0.35M soln. of 5-(ethylthio)-1H-tetrazole in MeCN as activator, and c) the detritylation was accomplished by 6% (v/v) Cl₂CHCO₂H in ClCH₂CH₂Cl over a 3-min period. All oligonucleotides were synthesized in the 'Trityl-off' mode.

5.4. Post-Automation Procedures. 5.4.1. Removal of 2-Cyanoethyl Protecting Group, Sugar and Nucleobase Protecting Groups, and Detachment from CPG Solid Support (Table 3 lists the specific deprotection method for the specific sequence). Method E. After the automated synthesis was completed, the CPG solid support containing the oligonucleotide was dried in vacuo for 30 min, transferred to a pear-shaped 10-ml flask, and treated with 2.4 ml of pyridine/Et₃N 5:1 for 5.5 h at r.t. Evaporation of pyridine and Et₃N in vacuo, followed by co-evaporation with DMF – avoiding temp. over 35° – resulted in dry CPG solid support. To the flask containing the dry CPG solid support was added 2.4 ml of a mixture of 40% aq. MeNH₂ conc. aq. NH₃ 1:1 and shaken at r.t. for 1 h. The mixture was filtered, washed with H_2O , the washings and filtrates were combined and concentrated in vacuo to afford the crude oligonucleotides. The residue was dissolved in ca. 2 ml of H₂O, filtered (Nalgene syringe filter, 0.2μ M), and taken to the next step, HPLC purification.

All of the above deprotections were monitored by anion-exchange HPLC for optimum deprotection time.

6. HPLC Purification of Oligonucleotides. - Oligonucleotides were purified by ion-exchange HPLC. Anion-exchange HPLC was performed either on Pharmacia Äkta Purifier (900) controlled by UNICORN system. Columns: Mono Q HR 5/5 (Pharmacia); buffer $A:10 \text{ mm Na}_2\text{HPO}_4$ in H₂O, pH 10.5-11.5; buffer $B:$

 10 mm Na_{2} HPO₄ in H₂O, 0.1M NaCl, pH 10.5 - 11.5. The product fractions were collected in a vial containing either 0.1 aq. AcOH soln. or 0.1 aq. phosphate buffer soln.

For certain A,T-containing sequences (see Table 3), the following buffer afforded better separation: buffer $A:25$ mm Tris \cdot HCl buffer and 1.5M urea in H₂O, pH 8.0; buffer B: 25 mm Tris \cdot HCl buffer and 1.5M urea in H₂O, 1M NH₄Cl, pH 8.0. The product fractions were collected in a vial containing 20 μ l of a 1.0m aq. AcOH soln.

7. Desalting of Oligonucleotides $[13c]$. $-$ The product fractions from HPLC purification were combined and diluted with $0.5 - 1$ M aq. Et₃NH \cdot HCO₃ buffer to twice the volume, and was the soln. applied to a previously conditioned reverse-phase Sep-Pak cartridge (C18 Waters). Successive elution with $0.1 - 0.5$ M aq. Et₃NH · HCO₃ buffer soln., MeCN/H₂O 1:1 and lyophilization of the MeCN/H₂O fractions containing the product (monitored by UV at 260 nm) afforded the salt-free oligonucleotides. The residue was dissolved in 1 ml of H₂O and stored at -20° as stock soln.

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